



SCIENCE & ENERGY

convection

Heat generation in a fluid that is caused by a variation in temperature resulting from the movement of molecules. Here, the heated water expands, rises and releases its heat to the surrounding air.

vapor

Gaseous state of water above its boiling point (water boils and is converted to vapor at 212°F or 100°C).

liquid

Matter having a definite mass and volume but no shape; its atoms are relatively mobile in relation to each other.

radiation

Heat generation in the form of electromagnetic waves emitted by a heated body (solid, liquid or gas).

convection current

Movement of fluid caused by a difference in density, which transfers heat. The heated water rises and is replaced by the cooler water from the surface.

flame

Incandescent gas resulting from the combustion of a mixture of gas and air; it produces heat and light.

solid

Rigid body possessing mass, volume and a definite form; its atoms are linked to each other and are almost completely at rest.

conduction

Heat generation in a body (usually a solid) or between two bodies in contact; the molecules vibrate but no matter moves.

SCIENCE & ENERGY

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INTRODUCTION

EDITORIAL POLICY

The Visual Dictionary takes an inventory of the physical environment of a person who is part of today's technological age and who knows and uses a large number of specialized terms in a wide variety of fields.

Designed for the general public, it responds to the needs of anyone seeking the precise, correct terms for a wide range of personal or professional reasons: finding an unknown term, checking the meaning of a word, translation, advertising, teaching material, etc.

The target user has guided the choice of contents for *The Visual Dictionary*, which aims to bring together in 12 thematic books the technical terms required to express the contemporary world, in the specialized fields that shape our daily experience.

STRUCTURE

Each tome has three sections: the preliminary pages, including the table of contents; the body of the text (i.e. the detailed treatment of the theme); the index.

Information is presented moving from the most abstract to the most concrete: sub-theme, title, subtitle, illustration, terminology.

TERMINOLOGY

Each word in *The Visual Dictionary* has been carefully selected following examination of high-quality documentation, at the required level of specialization.

There may be cases where different terms are used to name the same item. In such instances, the word most frequently used by the most highly regarded authors has been chosen.

Words are usually referred to in the singular, even if the illustration shows a number of individual examples. The word designates the concept, not the actual illustration.

DEFINITIONS

Within the hierarchical format of *The Visual Dictionary*'s presentation, the definitions fit together like a Russian doll. For example, the information within the definition for the term *insect* at the top of the page does not have to be repeated for each of the insects illustrated. Instead, the text concentrates on defining the distinguishing characteristics of each insect (the *louse* is a parasite, the female *yellow jacket* stings, and so forth).

Since the definition leaves out what is obvious from the illustration, the illustrations and definitions complement one another.

The vast majority of the terms in the *Visual Dictionary* are defined. Terms are not defined when the illustration makes the meaning absolutely clear, or when the illustration suggests the usual meaning of the word (for example, the numerous *handles*).

METHODS OF CONSULTATION

Users may gain access to the contents of *The Visual Dictionary* in a variety of ways:

- From the TABLE OF CONTENTS at the end of the preliminary pages, the user can locate by title the section that is of interest.
- With the INDEX, the user can consult *The Visual Dictionary* from a word, so as to see what it corresponds to, or to verify accuracy by examining the illustration that depicts it.
- The most original aspect of *The Visual Dictionary* is the fact that the illustrations enable the user to find a word even if he or she only has a vague idea of what it is. The dictionary is unique in this feature, as consultation of any other dictionary requires the user first to know the word.

TITLE

Its definition is found below. If the title refers to information that continues over several pages, after the first page it is shown in a shaded tone with no definition.

DEFINITION

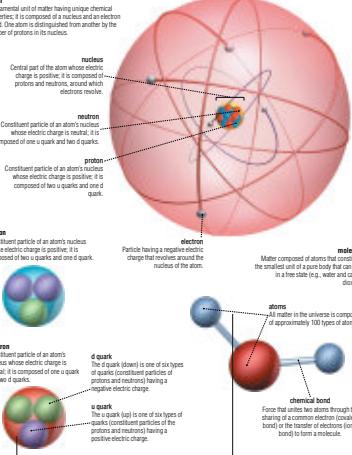
It explains the inherent qualities, function, or characteristics of the element depicted in the illustration.

TERM

Each term appears in the index with a reference to the pages on which it appears.

SUB-THEME

These are shown at the end of the preliminary pages along with their definitions. They are then repeated on each page of a section, but without the definition.

matter	CHEMISTRY	matter
<p>atom Any substance that has mass is composed of atoms and occupies space.</p> <p>nucleus Fundamental unit of matter having unique chemical properties. It is composed of a nucleus and an electron cloud. Its mass is much more than another by the number of protons in its nucleus.</p> <p>neutron Constituent particle of an atom whose electric charge is neutral. It is composed of one u quark and two d quarks.</p> <p>proton Constituent particle of an atom whose electric charge is positive. It is composed of two u quarks and one d quark.</p> <p>electron Particle having a negative electric charge that revolves around the nucleus of the atom.</p> <p>molecule Matter composed of atoms that constitutes the smallest unit of a pure substance that can exist in a free state (e.g., water or carbon dioxide).</p> <p>atoms All matter in the universe is composed of approximately 100 types of atoms.</p> <p>d quark The quarks (down) is one of six types of quarks (constituent particles of protons and neutrons) having a negative electric charge.</p> <p>u quark The up quark (up) is one of six types of quarks (constituent particles of the proton and neutron) having a positive electric charge.</p>		<p>states of matter Matter exists in three fundamental states (solid, liquid and gaseous), which depend on temperature and pressure under which the matter is subjected.</p> <p>condensation Change of a substance from a gaseous state directly to a liquid state; it results from cooling.</p> <p>saturation Change of a substance from a solid state directly to a gaseous state; it results from heating.</p> <p>crystallization Change of a substance from an amorphous state to a crystallized state; it results from cooling, which causes the atoms to become ordered.</p> <p>gas Matter in a gaseous state; its atoms are fully mobile with respect to each other.</p> <p>evaporation Change of a substance from a liquid state to a gaseous state; it results from heating.</p> <p>supersaturation The process of cooling a liquid below the point at which it normally freezes (solidifies); its atoms become unstable.</p> <p>amorphous solid Body that resembles a congealed liquid whose atoms are not ordered.</p> <p>liquid Matter having a definite mass and volume but no shape; its atoms are relatively much closer to each other.</p> <p>solid Rigid body possessing mass, volume and a definite form; its atoms are linked to each other and are almost completely at rest.</p> <p>freezing Change of a substance from a liquid state to a solid state; it results from cooling.</p> <p>melting Change of a substance from a solid state to a liquid state; it results from heating.</p>

ILLUSTRATION

It is an integral part of the visual definition for each of the terms that refer to it.

NARROW LINES

These link the word to the item indicated. Where too many lines would make reading difficult, they have been replaced by color codes with captions or, in rare cases, by numbers.

CONTENTS

8 CHEMISTRY

- 8 Matter
- 12 Chemical elements
- 21 Laboratory equipment
- 24 Chemistry symbols

25 PHYSICS: MECHANICS

- 24 Lever
- 25 Gearing systems
- 26 Double pulley system

27 PHYSICS: ELECTRICITY AND MAGNETISM

- 27 Parallel electrical circuit
- 28 Generators
- 30 Dry cells
- 32 Electronics
- 34 Magnetism

35 PHYSICS: OPTICS

- 35 Wave
- 36 Electromagnetic spectrum
- 37 Color synthesis
- 38 Vision
- 40 Lenses
- 41 Pulsed ruby laser
- 42 Prism binoculars
- 43 Magnifying glass and microscopes
- 48 Telescopic sight
- 50 Reflecting telescope
- 51 Refracting telescope

52 MEASURING DEVICES

- 52 Measure of temperature
- 55 Measure of time
- 60 Measure of weight
- 65 Measure of length
- 65 Measure of distance
- 66 Measure of thickness
- 68 Measure of angles

70 SCIENTIFIC SYMBOLS

- 70 International system of units
- 72 Mathematics
- 74 Biology
- 75 Geometry
- 76 Geometrical shapes

82 GEOTHERMAL AND FOSSIL ENERGY

- 82 Production of electricity from geothermal energy
- 84 Thermal energy
- 86 Coal mine
- 96 Oil

112 HYDROELECTRICITY

- 112 Hydroelectric complex
- 116 Generator unit
- 119 Examples of dams
- 124 Steps in production of electricity
- 126 Electricity transmission
- 130 Tidal power plant

134 NUCLEAR ENERGY

- 134 Production of electricity from nuclear energy
- 138 Fuel handling sequence
- 140 Fuel bundle
- 141 Nuclear reactor
- 142 Nuclear generating station
- 144 Carbon dioxide reactor
- 146 Heavy-water reactor
- 148 Pressurized-water reactor
- 150 Boiling-water reactor

152 SOLAR ENERGY

- 152 Solar cell
- 153 Flat-plate solar collector
- 154 Solar-cell system
- 156 Solar furnace
- 158 Production of electricity from solar energy
- 160 Solar house

163 WIND ENERGY

- 163 Windmill
- 165 Wind turbines and electricity production

169 INDEX

matter

Any substance that has mass, is composed of atoms and occupies space.

atom

Fundamental unit of matter having unique chemical properties; it is composed of a nucleus and an electron cloud. One atom is distinguished from another by the number of protons in its nucleus.

nucleus

Central part of the atom whose electric charge is positive; it is composed of protons and neutrons, around which electrons revolve.

neutron

Constituent particle of an atom's nucleus whose electric charge is neutral; it is composed of one u quark and two d quarks.

proton

Constituent particle of an atom's nucleus whose electric charge is positive; it is composed of two u quarks and one d quark.

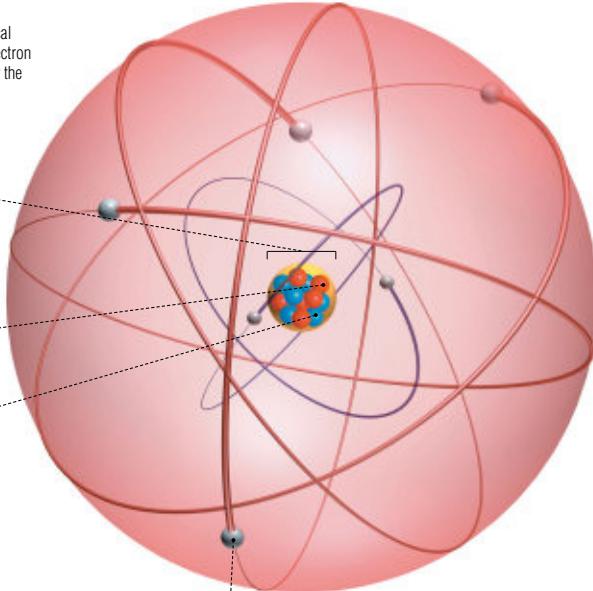
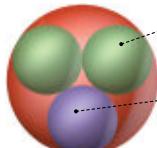
proton

Constituent particle of an atom's nucleus whose electric charge is positive; it is composed of two u quarks and one d quark.



neutron

Constituent particle of an atom's nucleus whose electric charge is neutral; it is composed of one u quark and two d quarks.

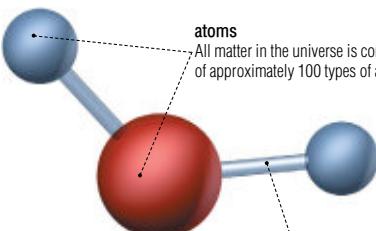


electron

Particle having a negative electric charge that revolves around the nucleus of the atom.

molecule

Matter composed of atoms that constitutes the smallest unit of a pure body that can exist in a free state (e.g., water and carbon dioxide).



atoms

All matter in the universe is composed of approximately 100 types of atoms.

d quark

The d quark (down) is one of six types of quarks (constituent particles of protons and neutrons) having a negative electric charge.

u quark

The u quark (up) is one of six types of quarks (constituent particles of the protons and neutrons) having a positive electric charge.

chemical bond

Force that unites two atoms through the sharing of a common electron (covalent bond) or the transfer of electrons (ionic bond) to form a molecule.

states of matter

Matter exists in three fundamental states (solid, liquid and gaseous), which depend on the temperature and pressure to which the matter is subjected.

condensation

Change of a substance from a gaseous state to a liquid state; it results from cooling.

gas

Malleable and expandable matter whose only definable property is mass; its atoms are fully mobile with respect to each other.

sublimation

Change of a substance from a solid state directly to a gaseous state without passing through the liquid state; it results from heating.

evaporation

Change of a substance from a liquid state to a gaseous state; it results from heating.

crystallization

Change of a substance from an amorphous state to a crystallized state; it results from cooling, which causes the atoms to become ordered.

supercooling

The process of cooling a liquid below the point at which it normally freezes (solidifies); its atoms become unstable.

amorphous solid

Body that resembles a congealed liquid whose atoms are not ordered.

condensation

Change of a substance from a gaseous state to a liquid state; it results from cooling.

liquid

Matter having a definite mass and volume but no shape; its atoms are relatively mobile in relation to each other.

solid

Rigid body possessing mass, volume and a definite form; its atoms are linked to each other and are almost completely at rest.

freezing

Change of a substance from a liquid state to a solid state; it results from cooling.

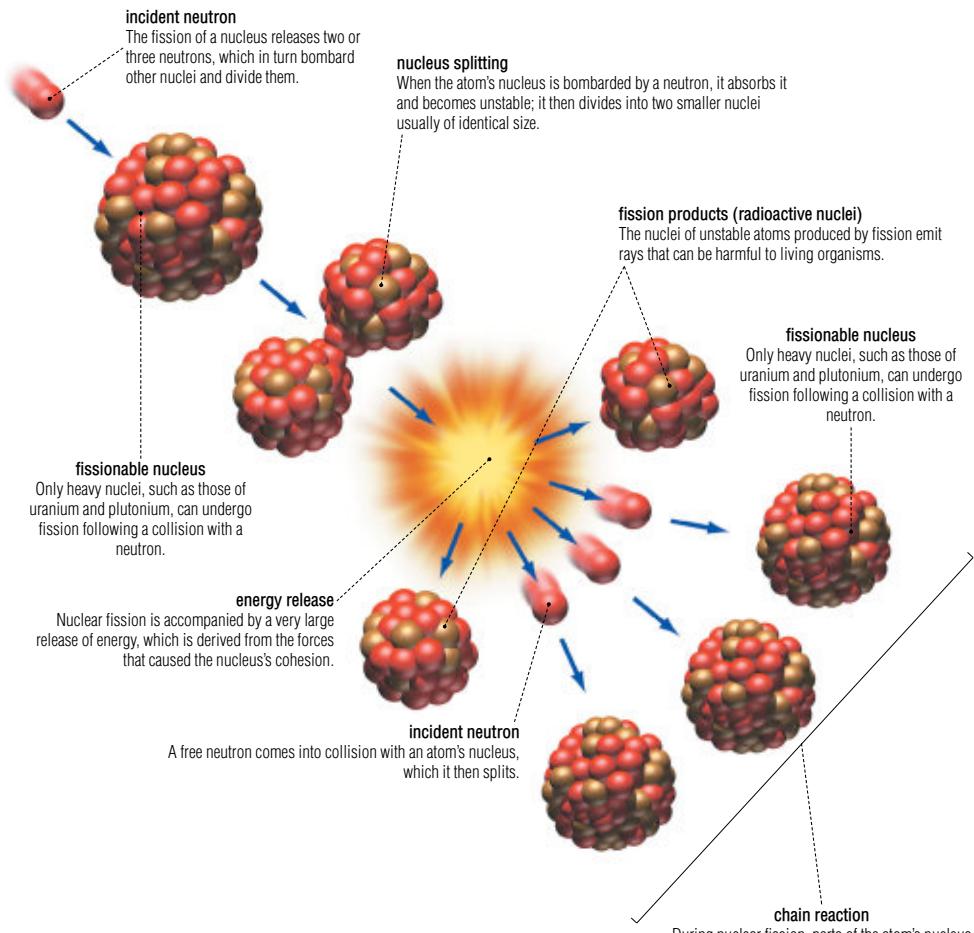
melting

Change of a substance from a solid state to a liquid state; it results from heating.

matter

nuclear fission

Process by which the atoms' nuclei become fragmented (e.g., in a nuclear reactor); neutrons are released and energy is produced in the form of heat.



heat transfer

Heat transfer occurs in three ways that are related to molecular movement: conduction, convection and radiation.

convection

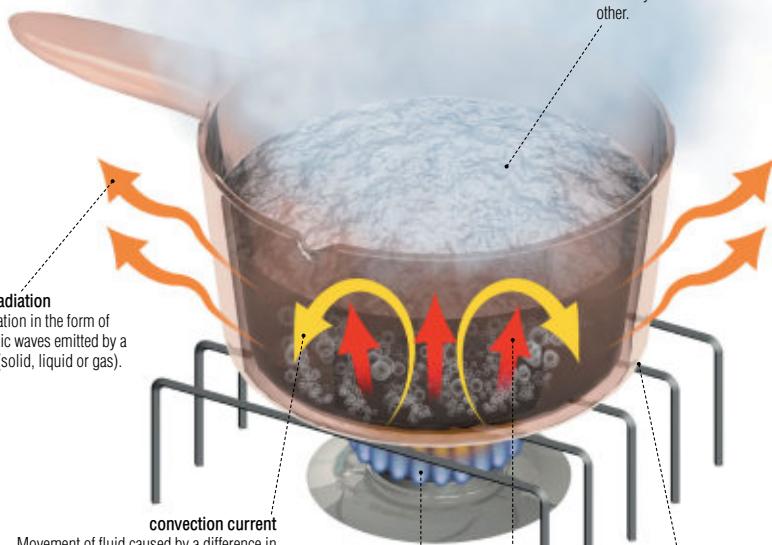
Heat generation in a fluid that is caused by a variation in temperature resulting from the movement of molecules. Here, the heated water expands, rises and releases its heat to the surrounding air.

vapor

Gaseous state of water above its boiling point (water boils and is converted to vapor at 212°F or 100°C).

liquid

Matter having a definite mass and volume but no shape; its atoms are relatively mobile in relation to each other.



convection current
Movement of fluid caused by a difference in density, which transfers heat. The heated water rises and is replaced by the cooler water from the surface.

flame

Incandescent gas resulting from the combustion of a mixture of gas and air; it produces heat and light.

conduction

Heat generation in a body (usually a solid) or between two bodies in contact; the molecules vibrate but no matter moves.

chemical elements

There are more than 110 chemical elements, most of which are naturally present in the universe. The others are created artificially in the laboratory.

table of elements

Table created by Dmitry Mendeleyev in 1869 that classifies the now approximately 110 known chemical elements such as oxygen, hydrogen, iron and lead. The elements are classified in order of their atomic weight and arranged into groups having similar properties.

1 H										
3 Li	4 Be									
11 Na	12 Mg									
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co		
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh		
55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir		
87 Fr	88 Ra	89 Ac	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt		
						58 Ce	59 Pr	60 Nd	61 Pm	62 Sm
						90 Th	91 Pa	92 U	93 Np	94 Pu

¹
H
hydrogen

This gas is the most abundant element in the universe and makes up part of the composition of water. It is used especially in petrochemistry and rocket engines.

								² He
			⁵ B	⁶ C	⁷ N	⁸ O	⁹ F	¹⁰ Ne
			¹³ Al	¹⁴ Si	¹⁵ P	¹⁶ S	¹⁷ Cl	¹⁸ Ar
²⁸ Ni	²⁹ Cu	³⁰ Zn	³¹ Ga	³² Ge	³³ As	³⁴ Se	³⁵ Br	³⁶ Kr
⁴⁶ Pd	⁴⁷ Ag	⁴⁸ Cd	⁴⁹ In	⁵⁰ Sn	⁵¹ Sb	⁵² Te	⁵³ I	⁵⁴ Xe
⁷⁸ Pt	⁷⁹ Au	⁸⁰ Hg	⁸¹ Tl	⁸² Pb	⁸³ Bi	⁸⁴ Po	⁸⁵ At	⁸⁶ Rn
¹¹⁰ Ds	¹¹¹ Rg	¹¹² Uub						

⁶³ Eu	⁶⁴ Gd	⁶⁵ Tb	⁶⁶ Dy	⁶⁷ Ho	⁶⁸ Er	⁶⁹ Tm	⁷⁰ Yb	⁷¹ Lu
⁹⁵ Am	⁹⁶ Cm	⁹⁷ Bk	⁹⁸ Cf	⁹⁹ Es	¹⁰⁰ Fm	¹⁰¹ Md	¹⁰² No	¹⁰³ Lr

alkali metals

Generally soft and silvery and very good conductors of heat and electricity; they are very reactant with nonmetals and break down in cold water.

lithium

The lightest of all the metals is used especially in alloys for the aerospace industry, in household batteries and in medicine.

3
Li**sodium**

Metal that is used especially in streetlights, kitchen salt (sodium chloride) and the manufacture of glass and cosmetic products.

11
Na**potassium**

Very reactant light metal that is used especially in fertilizer and matches; its salts are used in medicine.

19
K**rubidium**

Metal similar to potassium but much rarer that is used in the manufacture of photoelectric cells and special kinds of glass and lasers.

37
Rb**cesium**

Rare metal that is used especially in photoelectric cells, atomic clocks, infrared lamps and treating certain cancers.

55
Cs**francium**

The heaviest of the alkali metals is very rare and radioactive and has a very short life span (about 22 minutes).

87
Fr**alkaline earth metals**

Generally silvery and malleable and good conductors of heat and electricity; they react easily with nonmetals and water.

beryllium

Uncommon metal that is used especially in alloys for the aerospace industry and as a moderator in nuclear reactors.

4
Be**magnesium**

Metal necessary for the growth and metabolism of most living organisms; it is also a component of aluminum alloys.

12
Mg**calcium**

Metal that is one of the most essential elements in bones and teeth; it is also a component of cement, plaster and some alloys.

20
Ca**strontium**

Relatively rare metal that is used especially in pyrotechnics (fireworks), the manufacture of magnets and medicine.

38
Sr**barium**

Relatively abundant metal that is used especially in lubricants, pyrotechnics (fireworks), paint and radiology.

56
Ba**radium**

Extremely radioactive metal present in very low quantities in uranium ore; it is used mainly in medicine as a cancer treatment.

88
Ra

boron

 Boron is a nonmetallic element with atomic number 5. It is a light-colored, brittle, metallicloid used as a neutron absorber in nuclear reactors.

silicon

 Silicon is a nonmetallic element with atomic number 14. It is a light-colored, brittle, metallicloid used as a semiconductor in electronic devices.

germanium

 Germanium is a rare semimetal with atomic number 32. It is used in electronic devices and optical equipment.

semi-metals (metalloids)
Nonmetallic elements that are lusterless and solid; they possess a certain amount of electric and thermal conductivity.

arsenic

 Arsenic is a toxic semimetal with atomic number 33. It is used in very low doses for therapeutic purposes and in the manufacture of semiconductors.

antimony

 Antimony is a semimetal with atomic number 52. It is used in several alloys (mostly with lead) and especially in making metal for printing type and semiconductors.

selenium

 Selenium is a semimetal with atomic number 34. It is used in photoelectric cells and semiconductors; it is an indispensable trace element for organisms.

tellurium

 Tellurium is a rare semimetal with atomic number 52. It is used in the manufacture of detonators, electric resistors, rubber, ceramics and glass.

aluminum

 Aluminum is a light metal with atomic number 13. It is used in aeronautics, cars, buildings, electric cables, kitchen utensils and packaging.

tin

 Tin is a metal with atomic number 50. It is used especially as an anticorrosive for copper and steel and as a component in the preparation of bronze, welding and toothpaste.

bismuth

 Bismuth is a relatively rare metal with atomic number 83. It is used especially in alloys and cosmetics and in medicine (treatments for gastric ulcers and diarrhea).

gallium

 Gallium is a rare metal with atomic number 31. It is used especially in high-temperature thermometers, electroluminescent diodes and television screens (the color green).

thallium

 Thallium is a metal with atomic number 81. It is used especially in infrared detectors and some kinds of glass.

polonium

 Polonium is a very rare radioactive metal with atomic number 84. It is used as fuel in nuclear reactors; it emits radiation that is much more powerful than that of uranium.

indium

 Indium is a very rare metal with atomic number 49. It is used especially in race car engines and electronic devices, and as a coating for glass.

lead

 Lead is a heavy toxic metal with atomic number 82. It is used to prevent corrosion, as a protection against radiation and in accumulator batteries, paint and glass.

chemical elements

non-metals

Nonmetallic elements that are lusterless and nonmalleable; they are mostly gases and solids and are usually poor conductors of heat and electricity.

chlorine

Abundant toxic gas that is used to whiten fabric and paper, disinfect water and manufacture various other products (solvents).

bromine

Very toxic liquid that is used mainly to manufacture teargas, dyes and disinfectants and in photography and medications.

iodine

Solid that is used especially in pharmaceuticals (revulsives, antiseptics), in photography and dyes; it is also essential for the human body.

astatine

Radioactive element that is extremely rare in nature; it is used in medicine to study the thyroid gland and to detect cancerous tumors.

carbon

Element common in its pure state (diamond, graphite) or found in combination (air, coal, petroleum); it is present in animal and plant tissue.

fluorine

fluorine

Gas that is used especially for enriching uranium and manufacturing antistick coatings; it is present in bones and teeth.

phosphorus



phosphorus

Solid used especially in fertilizer (phosphates), matches and pyrotechnics (fireworks); it is also necessary for human beings.

sulfur



sulfur

Solid that is quite common in nature; it is used in car batteries, fertilizer, paint, explosives, pharmaceuticals and rubber.

oxygen



Gas that is the most abundant element on Earth and that comprises about 20% of the atmospheric air; it is used to breathe and in the manufacture of steel.

nitrogen



Gas that constitutes about 78% of the Earth's atmosphere, present in all animal and vegetable tissue (proteins), and in fertilizer, ammonia and explosives.

nitrogen



Gas that constitutes about 78% of the Earth's atmosphere, present in all animal and vegetable tissue (proteins), and in fertilizer, ammonia and explosives.

oxygen



Gas that is the most abundant element on Earth and that comprises about 20% of the atmospheric air; it is used to breathe and in the manufacture of steel.

oxygen



Gas that is the most abundant element on Earth and that comprises about 20% of the atmospheric air; it is used to breathe and in the manufacture of steel.

noble gases

Family of chemical elements also called inert, as they are weakly reactant.

helium



The lightest of the noble gases is noncombustible and abundant in the stars; it is used especially in inflating aerostats (such as balloons and dirigibles).

argon



Most abundant of the noble gases; it is used especially in incandescent lamps and in welding (protective gas).

xenon



Rarest gas in the atmosphere; it is used mainly in discharge lamps, photoflash bulbs and lasers.

neon



Noble gas that is used mainly in lighting (billboards, television tubes and fog lamps), but also as a liquid coolant.

krypton



Noble gas that is used in some incandescent lamps and in photography.

radon



Highly radioactive noble gas that is used mainly in medicine (destroying cancerous tumors) and in predicting earthquakes.

lanthanides (rare earth)

Very reactant elements found in the lanthanide series (monazite, xenotime); some are relatively abundant in the Earth's crust.

lanthanum

Metal that reacts with water to yield hydrogen; it is used especially in flint alloys and optical glass.

57
La**samarium**

Rare radioactive metal that is used especially in optical glass, lasers, nuclear reactors (absorbing neutrons) and permanent magnets.

62
Sm**cerium**

The most common metal of the lanthanide group and the main constituent of flint alloys (misch metal).

58
Ce**europerium**

The most reactant metal of the lanthanide group; it is used especially in television screens (the color orange) and nuclear reactors (absorbing neutrons).

63
Eu**praseodymium**

Metal that is used especially in protective lenses, colorants for glass, flint alloys (misch metal) and permanent magnets.

59
Pr**gadolinium**

Metal that is often alloyed with chromed steel; it is used especially in the manufacture of permanent magnets, magnetic heads and electronic components.

64
Gd**neodymium**

One of the most reactant of rare metals; it is used mainly to manufacture lasers, eyeglasses and permanent-magnet alloys.

60
Nd**terbium**

Rare metal that is used especially in lasers and semiconductors.

65
Tb**promethium**

Radioactive metal that is used mainly in specialized batteries and luminescent coatings for watches, and as a source of X-rays in medicine.

61
Pm**dysprosium**

Very rare metal that is used especially in permanent magnets, lasers and nuclear reactors (absorbing neutrons).

66
Dy**holmium**67
Ho

Very rare metal with limited applications; it is used in lasers and for coloring glass.

erbium

Metal that is used mainly in some alloys (especially with vanadium), lasers and infrared-absorbing glass, and as a colorant for glass and enamel.

thulium

The rarest of the lanthanide group; it is used as a source of X-rays in portable radiology equipment and in the manufacture of ferrites (magnetic ceramics).

69
Tm**ytterbium**

Metal that is used in the manufacture of stainless steel, in lasers and as a source of X-rays in portable radiology equipment.

70
Yb**lutetium**

Very rare metal that is difficult to separate; it has no real industrial applications but can be used as a catalyst (cracking, hydrogenation).

71
Lu

chemical elements

transition metals

Usually less reactant than alkali metals and alkaline earth metals but very good electric and thermal conductors. Many of these metals form vital alloys.

scandium

Rare and very light metal that is employed in aerospace construction because of its high fusion point (about 2,700°F or 1,500°C).

**titanium**

Metal that is used in several alloys employed in the manufacture of precision items and as a coating for light aerospace parts.

**vanadium**

Metal that is used mainly in alloys, to which it provides highly anticorrosive properties.

**chromium**

Bright metal that is used as an anticorrosive coating and in the manufacture of hard and resistant alloys; it gives emeralds and rubies their color.

**manganese**

Hard metal that is used mainly in the manufacture of specialty steels and household batteries; it is also an indispensable trace element for humans.

**iron**

The most used metal in the world due to its variety of alloys (steel, cast iron); it helps move oxygen through the body.

**cobalt**

Strong metal that is used in alloys (cutting tools, magnets) and in radiotherapy; it also yields a blue pigment.

**nickel**

Hard metal that resists corrosion; it is used in the manufacture of coins and cutlery, and as a protective coating for other metals (iron, copper).

**copper**

Reddish-brown metal that is a very good conductor of heat and electricity; it is used mainly in the manufacture of electric wire and alloys (brass, bronze).

**zinc**

Relatively abundant metal that is resistant to corrosion; it is used especially in the manufacture of alloys, tires, paint, ointments and perfume.

**yttrium**

Rare metal used in the manufacture of alloys, electronic components, lasers, television screens and in nuclear reactors.

**zirconium**

Metal that is used in alloys for the nuclear industry (protective sheathing, fuel rods) and in jewelry (imitation diamonds).

**niobium**

Rare metal that is used especially in alloys for jet aircraft, missiles, nuclear reactors, ointments and cutting tools.

**molybdenum**

Rare metal that is used in alloys (aircraft, missiles, nuclear reactors), electric lights and electronic tubes.

**technetium**

Radioactive metal (first element to have been produced artificially) that makes steel corrosion-free and is used in medical imaging.

**ruthenium**

Rare metal that hardens platinum and palladium; it is used in the manufacture of electric contacts, spark plugs and jewelry.

**rhodium**

Rare metal that resists corrosion and hardens platinum and palladium; it is used especially in catalytic converters and jewelry.

**palladium**

Rare and precious metal that is used especially in dentistry (dental prostheses), jewelry (white gold) and in catalytic converters.



silver

Precious metal that is the best conductor of heat and electricity; it is used especially in the manufacture of mirrors, jewelry and coins.

**cadmium**

Metal that is used especially as a protective covering for steel, in rechargeable batteries and in nuclear reactors (control rods).

**hafnium**

Rare metal that is used in the control rods of nuclear reactors, filaments for incandescent lamps and jet engines.

**tantalum**

Somewhat rare metal that is highly resistant to heat; it is used especially in nuclear reactors, missiles and capacitors.

**tungsten**

Metal that is resistant to very high heat; it is used in filaments for incandescent lamps and cutting tools.

**rhenium**

Rare metal that is resistant to wear and corrosion; it is used especially in pen tips and incandescent filaments for ovens.

**osmium**

Rare metal often alloyed with iridium and platinum; it is used in pen tips, bearings, compass needles and jewelry.

**iridium**

Rare metal that is often alloyed with platinum; it is used especially in electric contacts and jewelry.

**platinum**

Very rare metal used especially as a catalyst in chemistry (petrochemicals, vitamins), in jewelry and in precision equipment.

**gold**

Precious metal (nuggets, flakes) that is used as currency (ingots) and in jewelry, dentistry and electronics.

**mercury**

Rare metal that is used in measuring instruments (thermometers, barometers) and in the electricity industry.

**rutherfordium**

Artificial radioactive element that was first produced in laboratories in the 1960s; it has applications only in scientific research.

**dubnium**

Artificial radioactive element that was first produced in laboratories in the 1960s.

**seaborgium**

Artificial radioactive element that was first produced in laboratories in 1974; it is based on californium and oxygen.

**bohrium**

Artificial radioactive element that was first produced in laboratories in 1976; it is based on bismuth and chromium.

**hassium**

Artificial radioactive element that was first produced in laboratories in 1984; it is based on lead and iron.

**meitnerium**

Artificial radioactive element that was first produced in laboratories in 1982; it is based on bismuth and iron.

**darmstadtium**

Artificial radioactive element that was first produced in laboratories in 1994; it is based on nickel and lead.

**roentgenium**

Artificial radioactive element that was first produced in laboratories in 1994; it is based on bismuth and nickel.

**ununbium**

Artificial radioactive element that was first produced in laboratories in 1996; it is based on lead and zinc.



actinides

Radioactive elements that are abundant in nature (elements 89 to 92) or made artificially (elements 93 to 103). Most of them have no industrial applications.

actinium

Metal that is present in small quantities in uranium ore; it is used mainly as a source of neutrons in nuclear reactors.

**thorium**

Natural metal that is used especially in alloys, photoelectric cells and uranium production.

**protactinium**

Very rare metal that is present in uranium ore; it has few applications outside of scientific research.

**uranium**

Naturally abundant metal that is used mainly as fuel in nuclear reactors as well as in nuclear weapons.

**neptunium**

Rare metal that is produced from uranium; it is used in neutron-detection instruments.

**plutonium**

Metal that is produced from uranium; it is used especially as fuel in nuclear reactors as well as in nuclear weapons.

**americium**

Metal that is produced from plutonium; it is used mainly in smoke detectors and in radiology.

**curium**

Metal that is produced in small amounts from plutonium; it is used especially in thermoelectric generators for spacecraft propulsion.

**berkelium**

Metal that is produced in small amounts from americium; it is used for scientific research only.

**californium**

Metal produced from curium that is used especially in the treatment of cancer and in some measuring instruments such as humidistats.

**einsteinium**

Metal that was discovered in 1952 among the debris of the first thermonuclear explosion in the Pacific; it is used for scientific research only.

**fermium**

Metal that was discovered at the same time as einsteinium; it is used for scientific research only.

**mendelevium**

Metal that is produced from einsteinium; it is named in honor of the chemist Mendeleev (who classified the elements).

**nobelium**

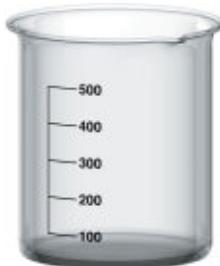
Metal that is produced from curium; it is named in honor of Alfred Nobel (inventor of dynamite and founder of the Nobel Prize).

**lawrencium**

Metal that is produced from californium; it is used for scientific research only.



These materials are highly varied: measurement instruments, various containers, heat sources, experimentation materials and mounting hardware.

**beaker**

Graduated container with a spout; it is used to create reactions (precipitation, electrolysis) and to measure approximate amounts of liquid.

Erlenmeyer flask

Graduated cone-shaped container that is used very frequently in laboratories; it can have a stopper and is used especially for mixing and measuring liquids.

**bottle**

Container of various sizes and shapes and usually with a straight neck for holding liquids.

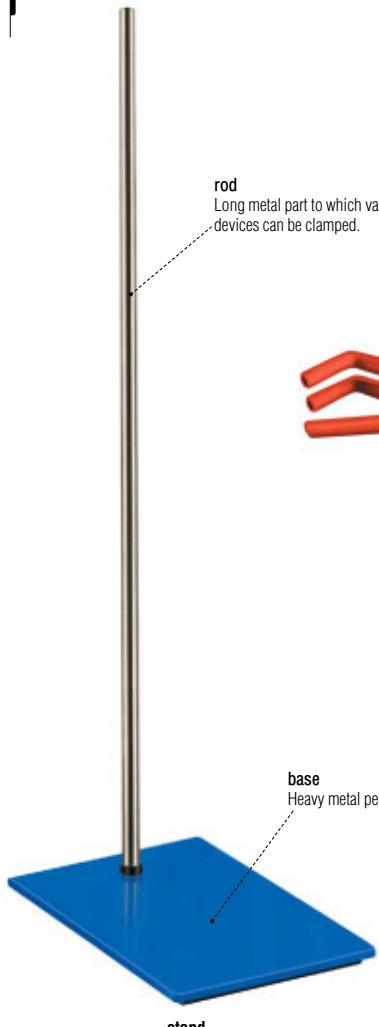
**wash bottle**

Flexible container that is squeezed lightly to squirt a liquid; it is used especially for cleaning equipment (test tubes, pipettes).

**round-bottom flask**

Spherical container used mainly for boiling liquids.

laboratory equipment

**rod**

Long metal part to which various laboratory devices can be clamped.

**clamp/holder**

Part attached to the stand's rod by a holder and having tongs that clamp onto the laboratory equipment to hold it in place.

**holder**

Part with a screw for attaching a clamp onto the stand's rod.

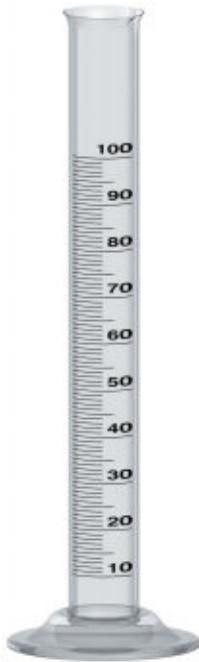
**gas burner**

Device that is fueled by gas to produce a flame for heating chemical products.

stand
Unit consisting of a base and a rod; it supports various laboratory apparatuses such as burettes and flasks.

graduated cylinder

Graduated tube with a spout that is used especially for measuring small amounts of liquid with precision.

**straight stopcock burette**

Long graduated tube for measuring liquids with high precision; it is fitted with a valve for manually regulating the flow.

**serological pipette**

Fine tube that is open at both ends; it is used to transfer very precise quantities of liquids from one container to another.

**Petri dish**

Flat transparent box for culturing microorganisms; it has a cover to protect them from contamination.

**test tube**

Cylindrical tube used to conduct various chemical experiments on small quantities (normally, it is not filled above one-third).

chemistry symbols

Symbols that simplify the writing of the elements, formulas and chemical reactions.



negative charge

Symbol that indicates a surplus of electrons in an atom, which means the atom has a negative electric charge. The chlorine atom, for example, forms a negative ion that is denoted as Cl⁻.



positive charge

Symbol that indicates a loss of electrons in an atom, which means the atom has a positive electric charge. The sodium atom, for example, forms a positive ion that is denoted as Na⁺.



reversible reaction

Chemical reaction that can occur in both directions; the products obtained (direct reaction) react between them to change back into the original reactants (inverse reaction).

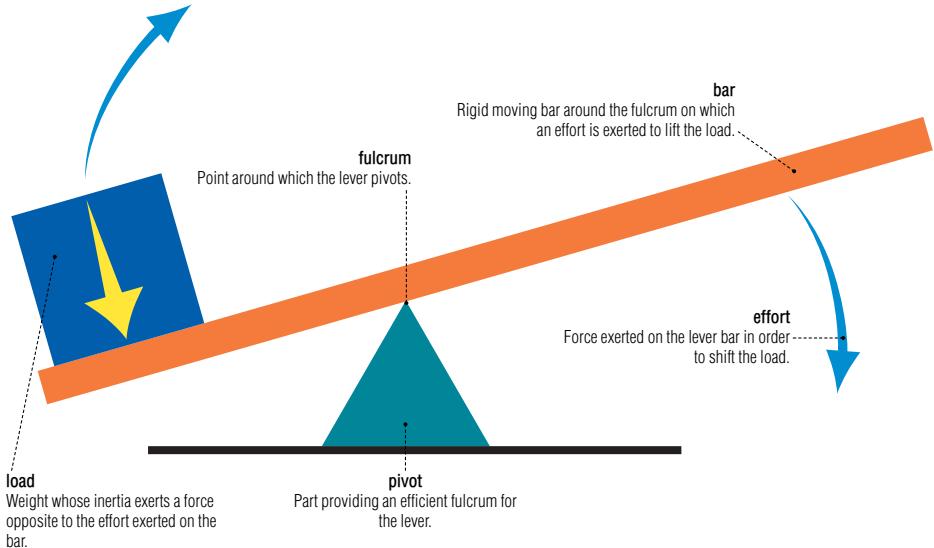


reaction direction

A chemical reaction corresponds to the conversion of reactants in products and is obtained by the loss of one of the reactants. The arrow indicates the direction in which this irreversible reaction occurs.

lever

System consisting of a bar pivoting on a fulcrum to lift a load. The amount of effort required is related to the position of the pivot and the length of the bar.

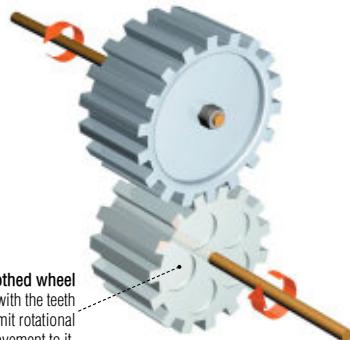


Mechanisms consisting of toothed parts that mesh to transmit the rotational motion of the shafts they are a part of.



rack and pinion gear

Gearing system converting a rotational movement into a horizontal movement (and vice versa); it is often used in the steering systems of automobiles.

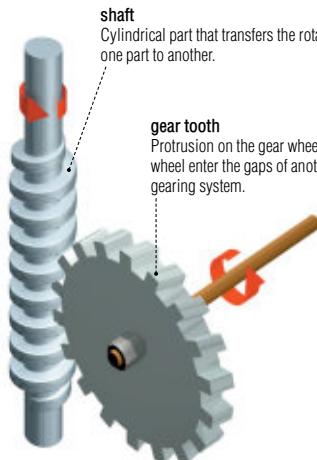


toothed wheel

Wheel with teeth that mesh with the teeth of another wheel to transmit rotational movement to it.

spur gear

Most common gearing system linking two parallel shafts that changes the speed and force of a rotation; it is used especially in automobile transmissions.



shaft

Cylindrical part that transfers the rotational movement of one part to another.

gear tooth

Protrusion on the gear wheel; the teeth of one wheel enter the gaps of another wheel to form a gearing system.

worm gear

One-way gearing system (only the screw can drive the wheel) for slowing down the speed of rotation between two perpendicular axles; it is used especially in the automobile industry (Torsen differential).

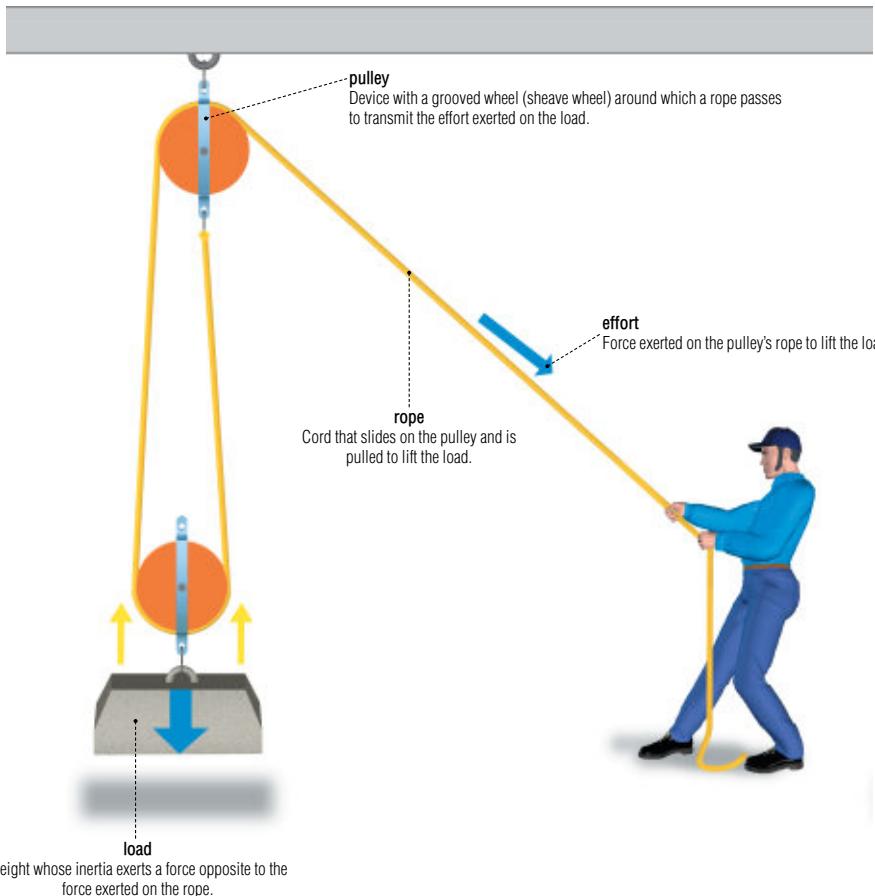


bevel gear

Gearing system linking two shafts at right angles that changes the direction of rotation; it is used especially in car jacks.

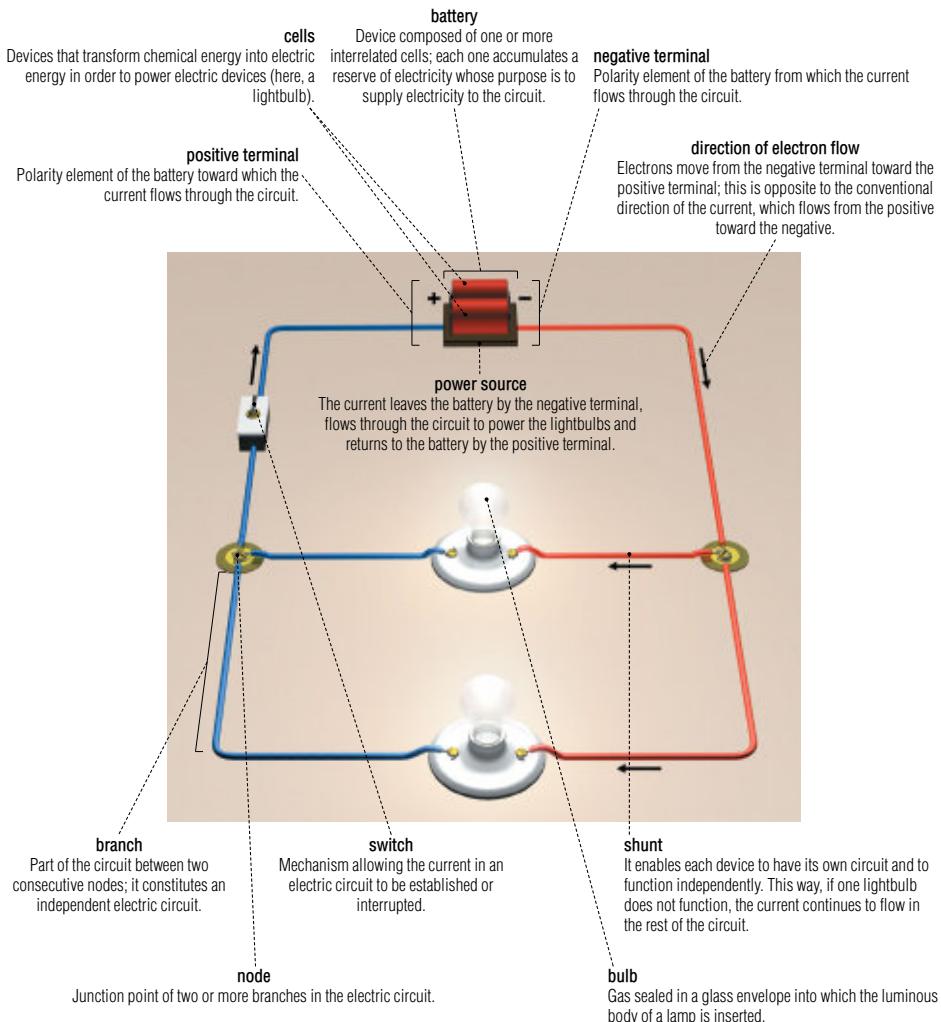
double pulley system

System consisting of two pulleys with a rope running around them to lift a load. Using two or more pulleys reduces the amount of effort needed.



parallel electrical circuit

It is divided into independent branches, through which the current flows with partial intensity (in a series circuit, all the elements receive the same intensity).



generators

Devices that convert mechanical energy (here, a shaft's rotational motion) into electric energy by moving a coil inside a magnet (electromagnetic induction).

dynamo

Continuous generator of electric current; it is used especially on bicycles for lighting.

armature

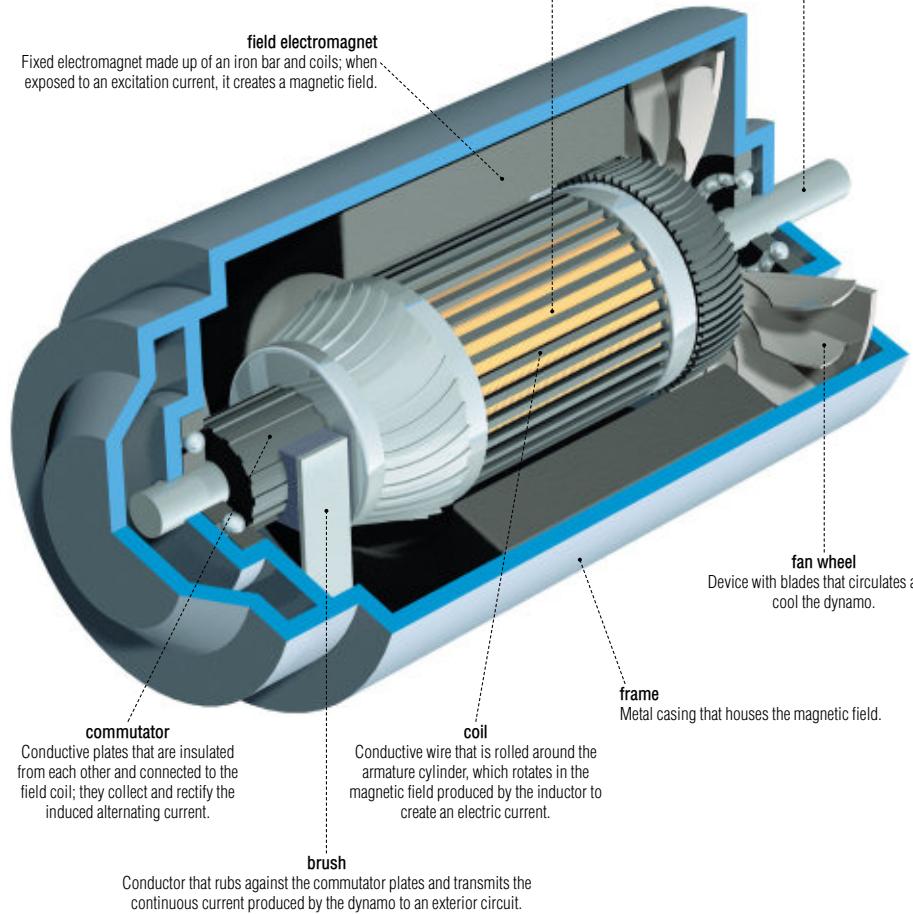
Moving part of the dynamo that is made up of a coil, which produces an electric current as it rotates inside the field.

shaft

Cylindrical part that transmits a rotational motion to the dynamo's armature.

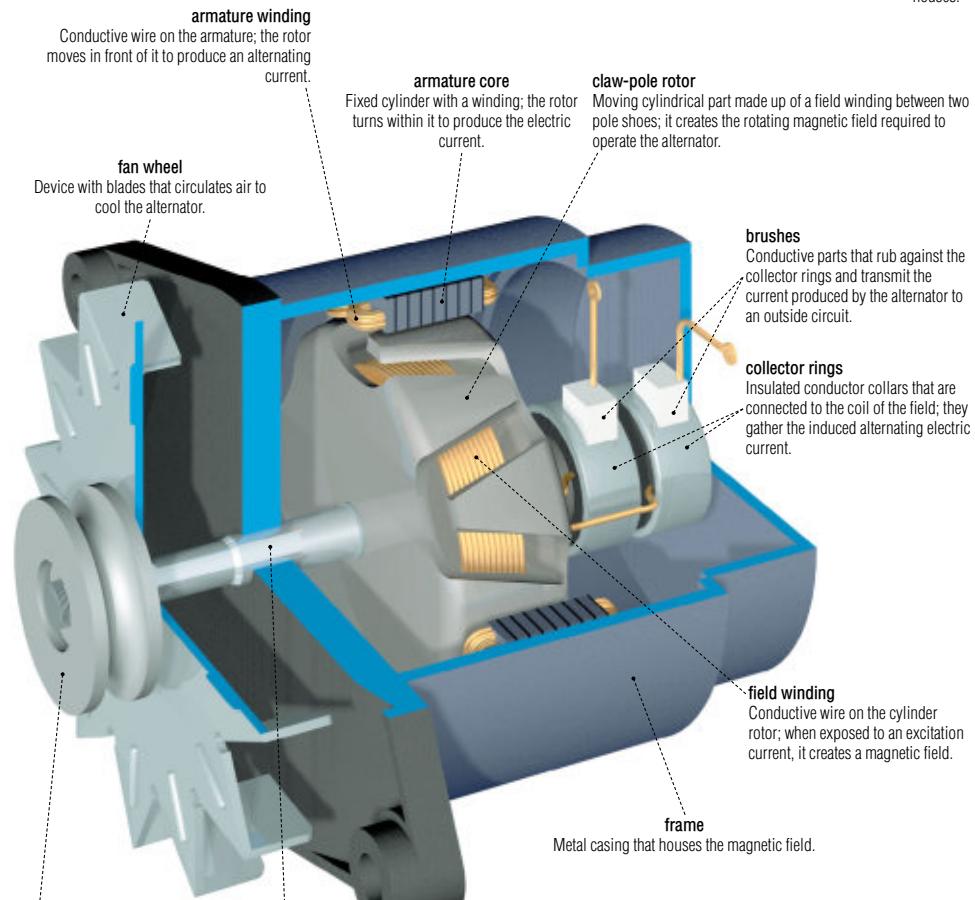
field electromagnet

Fixed electromagnet made up of an iron bar and coils; when exposed to an excitation current, it creates a magnetic field.



alternator

Generator of alternating current that is used especially in the automobile industry (powering electrical devices) and in power houses.

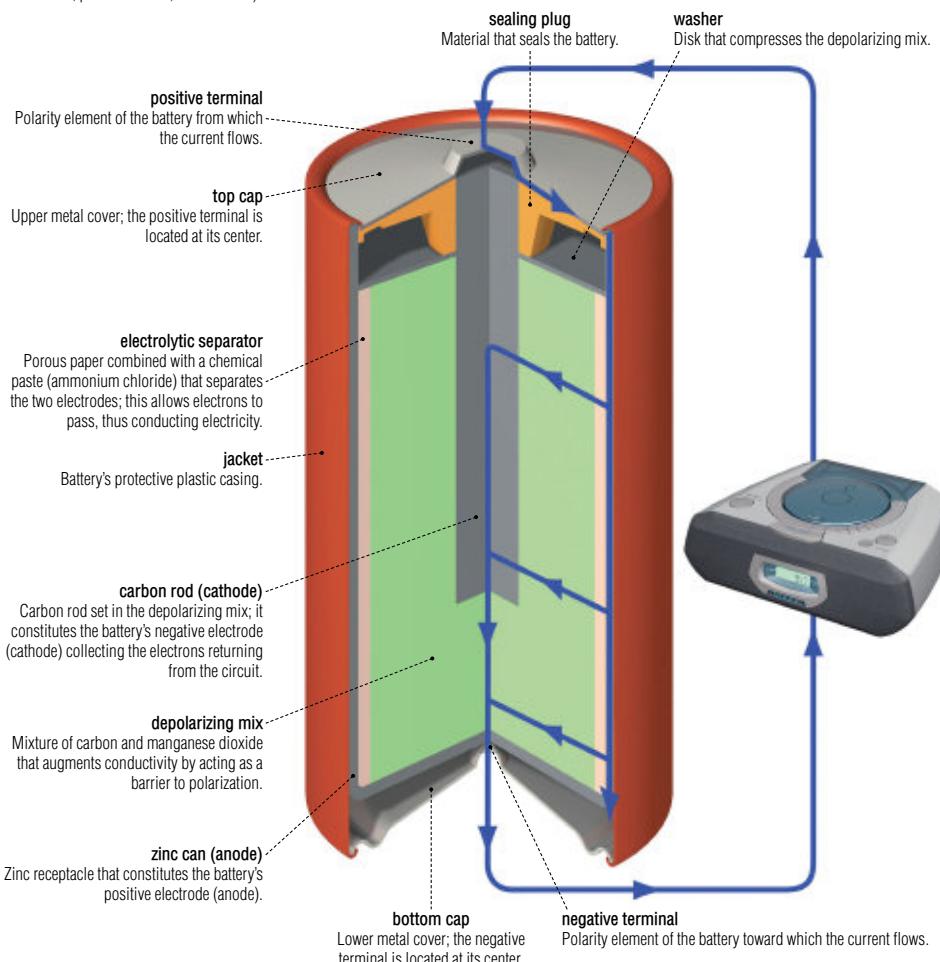


dry cells

Devices that transform chemical energy into electric energy (direct current); they usually cannot be recharged and the electrolyte is fixed in place.

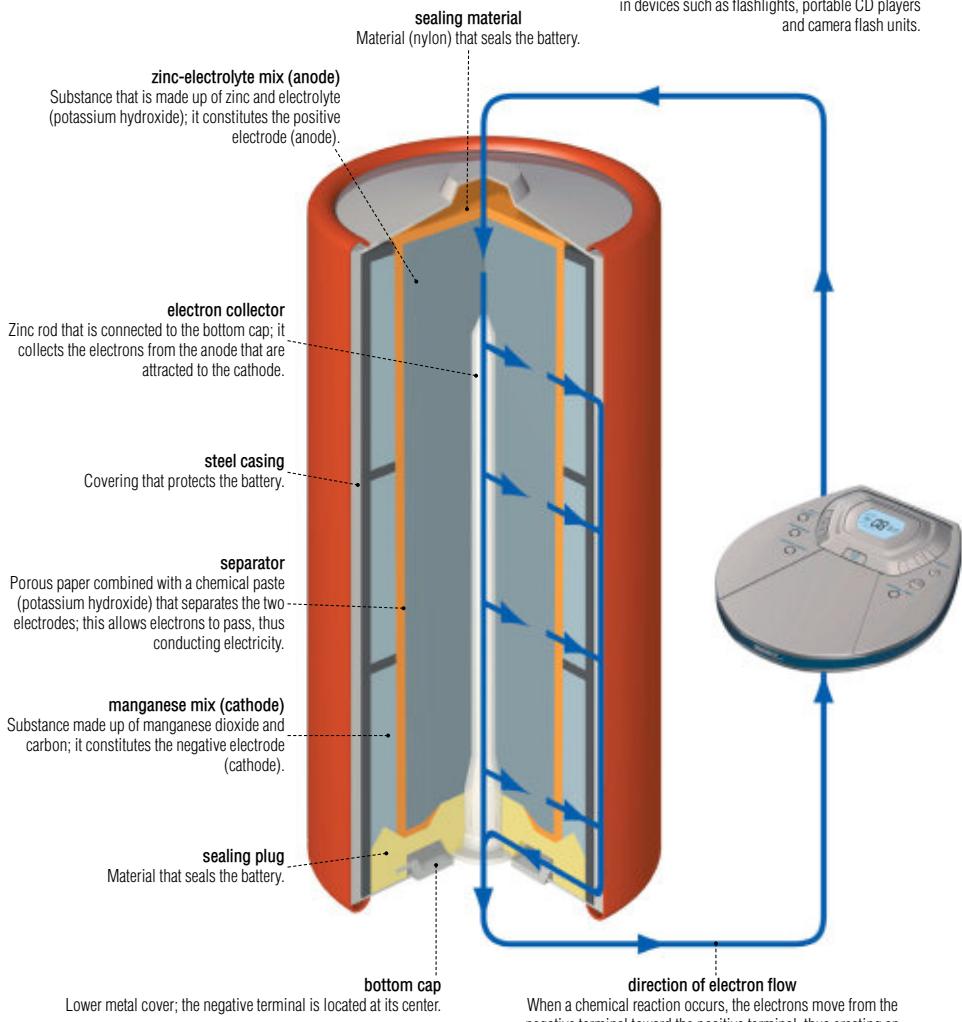
carbon-zinc cell

Battery that produces 1.5 V (also called Leclanché); its use is very widespread (pocket calculators, portable radios, alarm clocks).



alkaline manganese-zinc cell

High-performance battery that produces 1.5 V and has a longer life span than the carbon-zinc cell; it is used in devices such as flashlights, portable CD players and camera flash units.

**direction of electron flow**

When a chemical reaction occurs, the electrons move from the negative terminal toward the positive terminal, thus creating an electric current.

electronics

The scientific study of the behavior of the electron and its applications, such as computers, medicine and automation.

printed circuit board

Usually plastic insulated card with holes containing electronic components; the circuit is printed on its surface.

ceramic capacitor

Component with two conductive plates (silver, copper) separated by an insulator (ceramic); it stores weak electric charge.

plastic film capacitor

Commonly used component with two conductive plates (aluminum, tin) separated by an insulator (plastic); it stores electric charge.

electrolytic capacitors

Polarized components with two conductive components (aluminum, tantalum) separated by an insulator (electrolyte); they store strong electric charge.

packaged integrated circuit

Electric circuit under a plastic or ceramic casing; it has pins for connecting it to the circuit board.

printed circuit

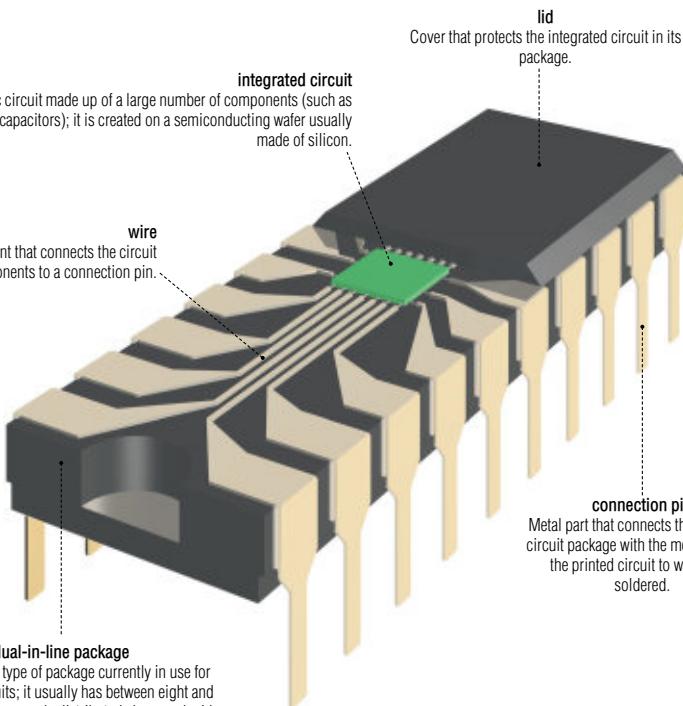
All of the conductive metal bands on an insulated base (card), which connect a circuit's components and allow a current to flow through it.

resistors

Electronic component that regulates the amount of current flowing in a circuit.

packaged integrated circuit

Integrated circuits are used especially in microprocessors, stereo equipment, calculators, watches and electronic games.



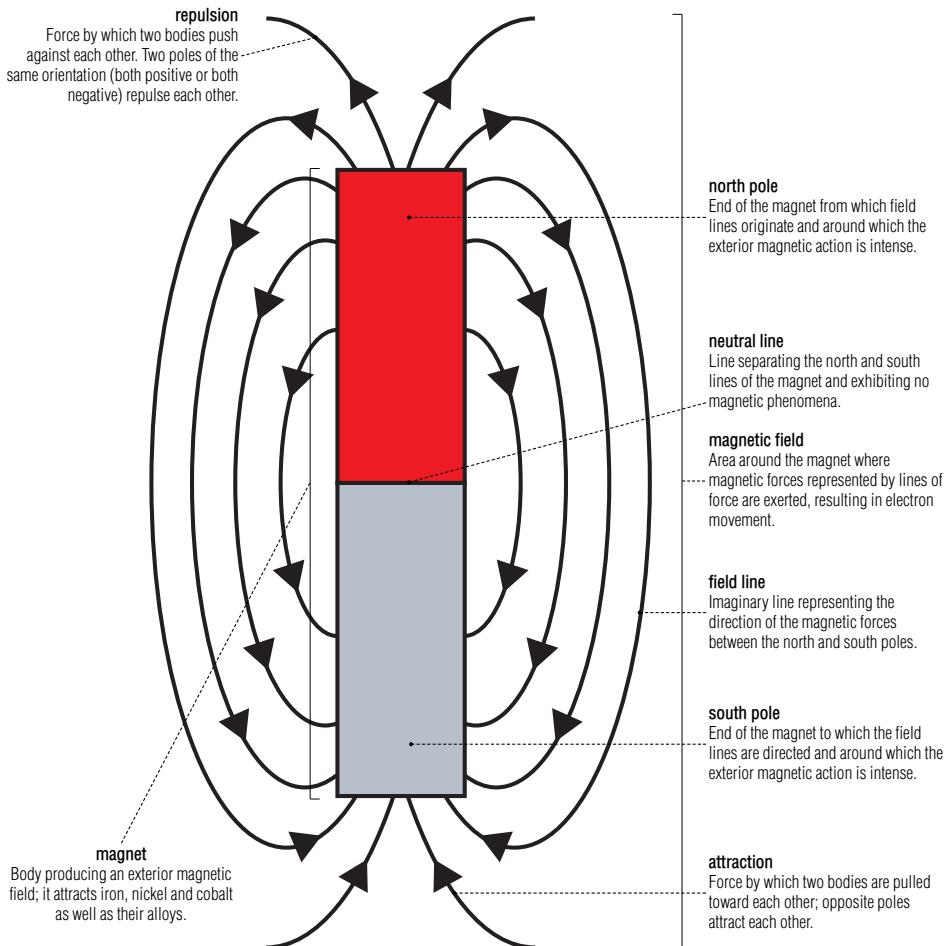
integrated circuit
Miniature electronic circuit made up of a large number of components (such as transistors and capacitors); it is created on a semiconducting wafer usually made of silicon.

wire
Conductive element that connects the circuit components to a connection pin.

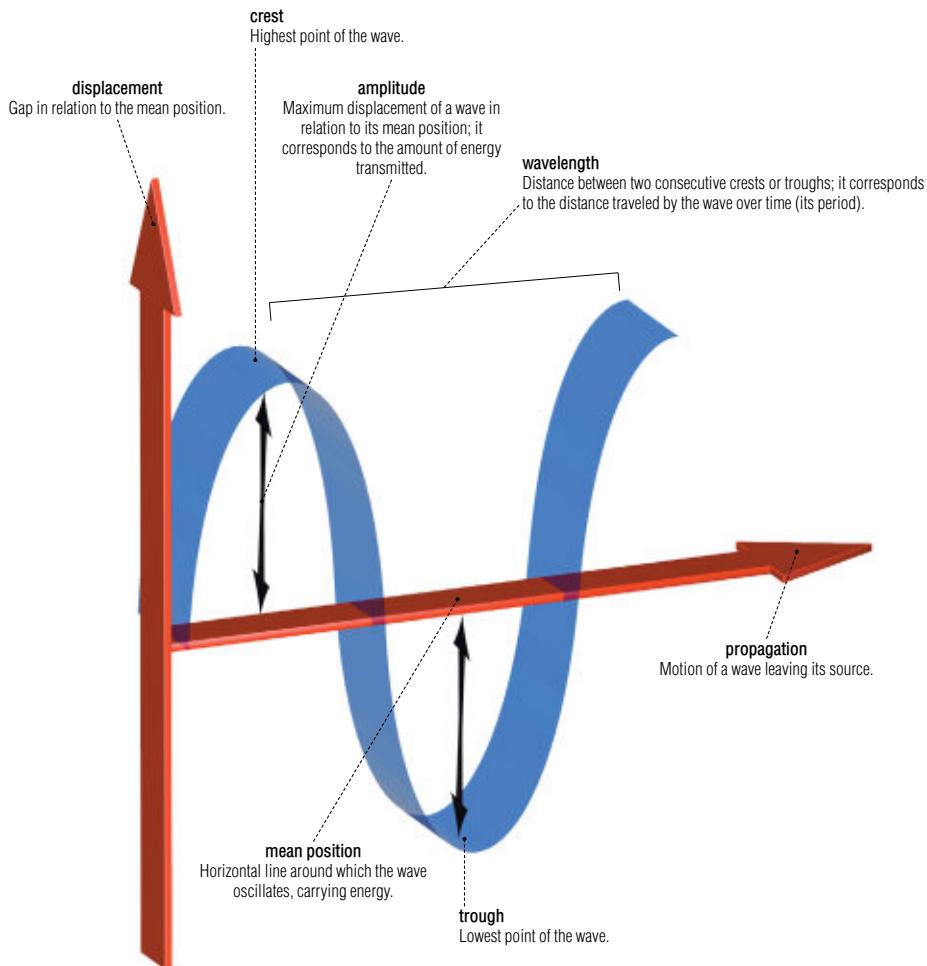
dual-in-line package
Most common type of package currently in use for integrated circuits; it usually has between eight and 48 pins, which are evenly distributed along each side of the package.

magnetism

Action exerted by magnets and magnetic fields and phenomena. Magnetism can be characterized by the forces of attraction and repulsion between two masses.



Oscillation caused by a disturbance; as it propagates through a medium (mechanical waves) or a vacuum (electromagnetic waves), it carries energy.



electromagnetic spectrum

Electromagnetic waves that are classified in ascending order of energy (frequency); they propagate at the speed of light (300,000 km/s).

radio waves

Very long electromagnetic waves (about 1 meter) having low frequency; they are used to transmit information (television, radio).

ultraviolet radiation

Electromagnetic waves used especially to tan skin and in microscopy, medicine and lighting (fluorescent tubes).

infrared radiation

Electromagnetic waves emitted by warm objects; their many uses include heating, medicine, aerial photography and weaponry.

gamma rays

Electromagnetic waves of very high frequency that are emitted by radioactive bodies; they are the most radiant and harmful rays and are used especially in treating cancer.

microwaves

Very short electromagnetic waves; their many applications include radar detection and microwave ovens.

visible light

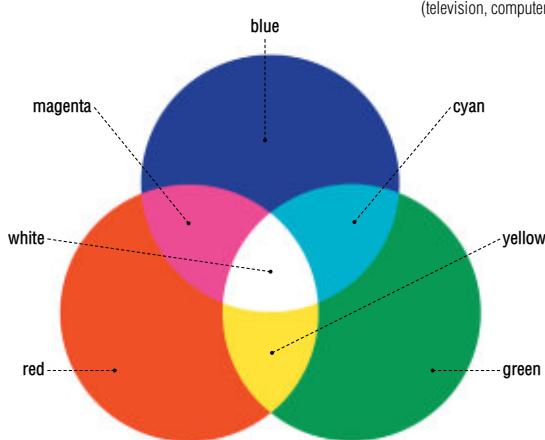
Electromagnetic radiation that is perceived by the human eye and ranges from red to violet.

X-rays

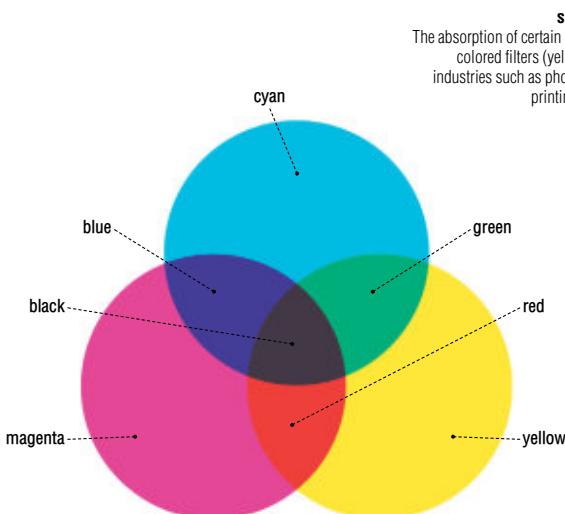
Electromagnetic waves used especially in radiology; frequent exposure can be harmful.

color synthesis

Technique of generating color by combining light rays or subtracting them to obtain a colored image.

**additive color synthesis**

The superimposition of primary colors (blue, green and red) is used especially in electronic screens (television, computer, video) to obtain intermediate tints.

**subtractive color synthesis**

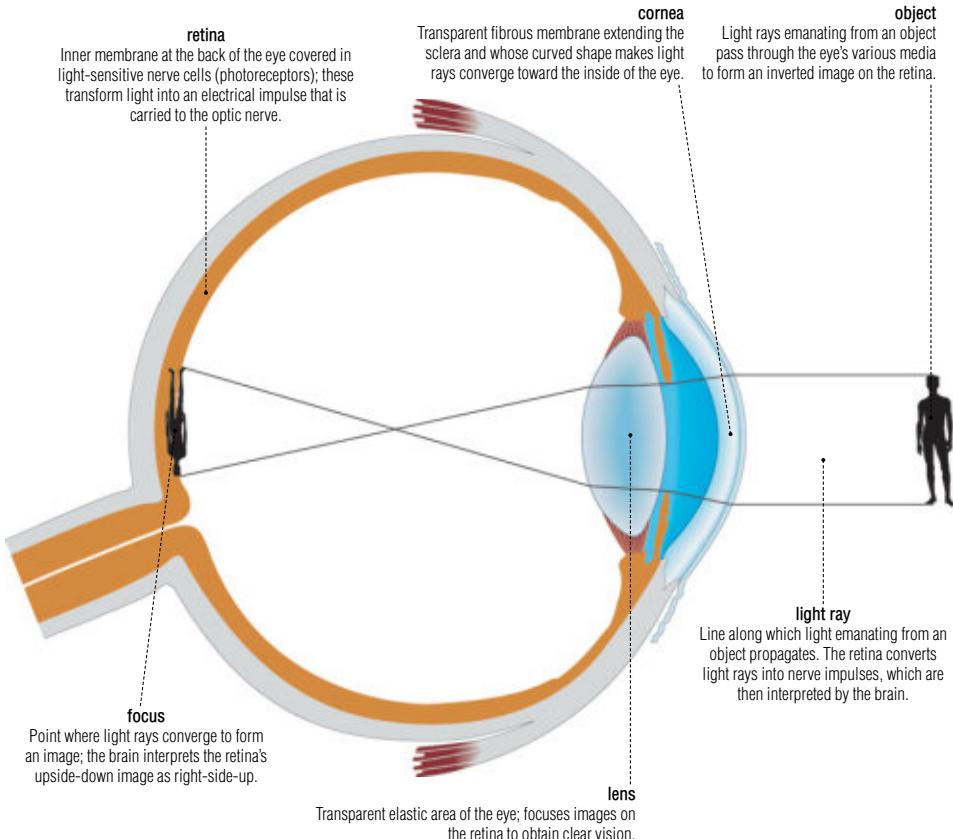
The absorption of certain light rays (blue, green, red) by colored filters (yellow, magenta, cyan) is used in industries such as photography, film production and printing to obtain intermediate tints.

vision

Ability to perceive shapes, distances, motion and colors; it is related to light rays and varies depending on the degree of sensitivity of the eye.

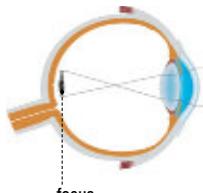
normal vision

The image of an object is formed on the retina after passing through the lens, which, depending on the distance of the object, expands or contracts to give a sharp image.



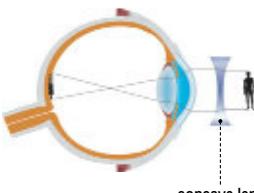
vision defects

Images do not form on the retina, thus resulting in blurry vision; such defects are corrected by eyeglasses, contact lenses or even surgery.



focus

Point where light rays converge to form an image; the brain interprets the retina's upside-down image as right-side-up.

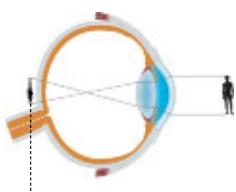


concave lens

Corrects myopia by causing light rays emanating from an object to diverge and project an image onto the focus of the retina.

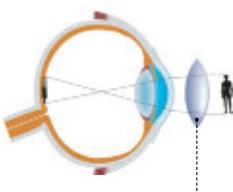
myopia

The image of a distant object is formed in front of the retina due to a defect in the light rays' convergence. This makes distant objects hard to see.



focus

Point where light rays converge to form an image; the brain interprets the retina's upside-down image as right-side-up.

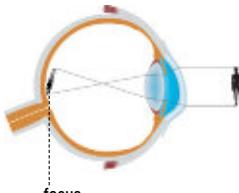


convex lens

Corrects hyperopia by causing light rays emanating from an object to converge and project an image onto the focus of the retina.

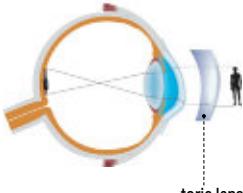
hyperopia

The image of an object is formed behind the retina due to a defect in the light rays' convergence as they pass through the lens. This makes near objects hard to see.



focus

Point where light rays converge to form an image; the brain interprets the retina's upside-down image as right-side-up.



toric lens

Has various powers depending on the rays' axes of convergence; it is used to offset the visual distortion caused by the cornea.

astigmatism

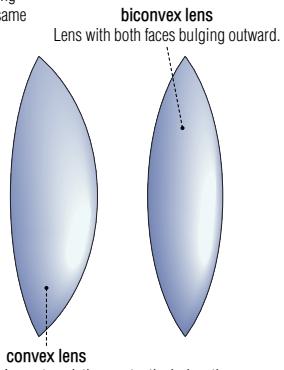
Usually caused by a curvature of the cornea, it is manifested by blurred vision when viewing both near and far objects, depending on various axes.

lenses

Transparent pieces of material (usually glass) that cause light rays to converge or diverge to form a sharp image (eyeglasses, microscopes, telescopes, cameras).

converging lenses

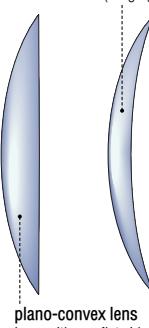
Thicker in the center than on the edges; they cause parallel light rays emanating from an object to converge onto the same point.



convex lens
Lens with one side bulging outward; the greater the bulge, the more the light rays converge.

positive meniscus

Lens where the concave side (curving inward) is less pronounced than the convex side (bulging outward).

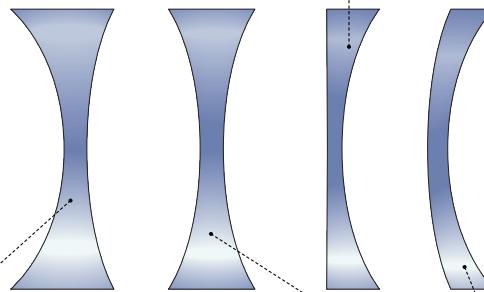


piano-convex lens

Lens with one flat side and one convex side (bulging outward).

diverging lenses

Thicker on the edges than in the center; they cause parallel light rays emanating from an object to diverge.



concave lens
Lens with one side curving inward; the greater the curvature, the more the light rays diverge.

piano-concave lens

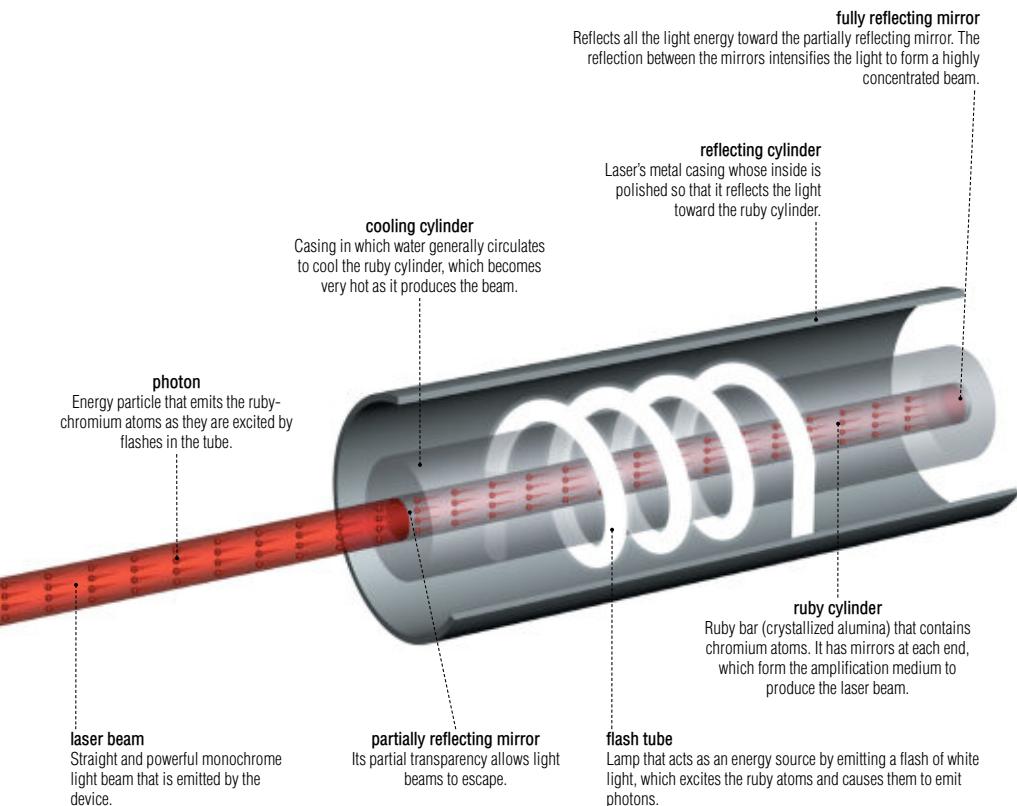
Lens with one flat side and one concave side (curving inward).

biconcave lens
Lens with both sides curving inward.

negative meniscus
Lens where the concave side (curving inward) is more pronounced than the convex side (bulging outward).

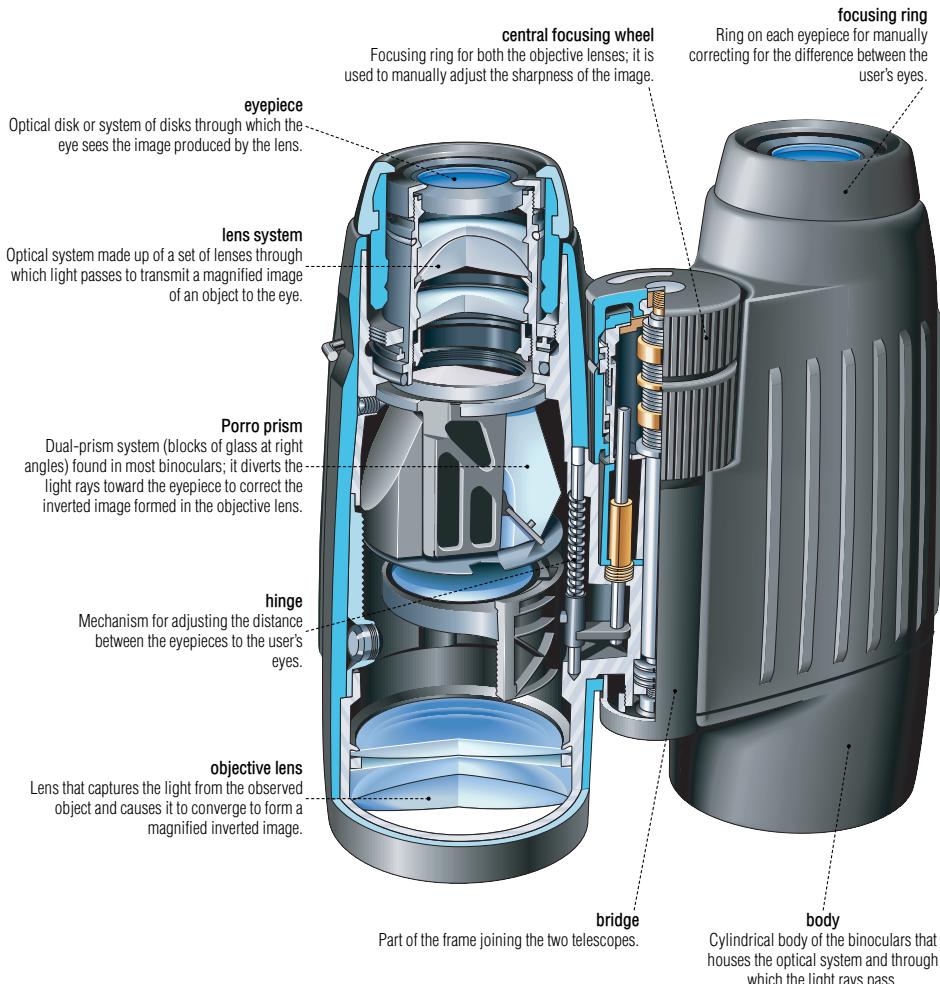
pulsed ruby laser

Device that produces a thin and very intense colored light beam; its various applications include fiber optics, manufacturing and surgery.



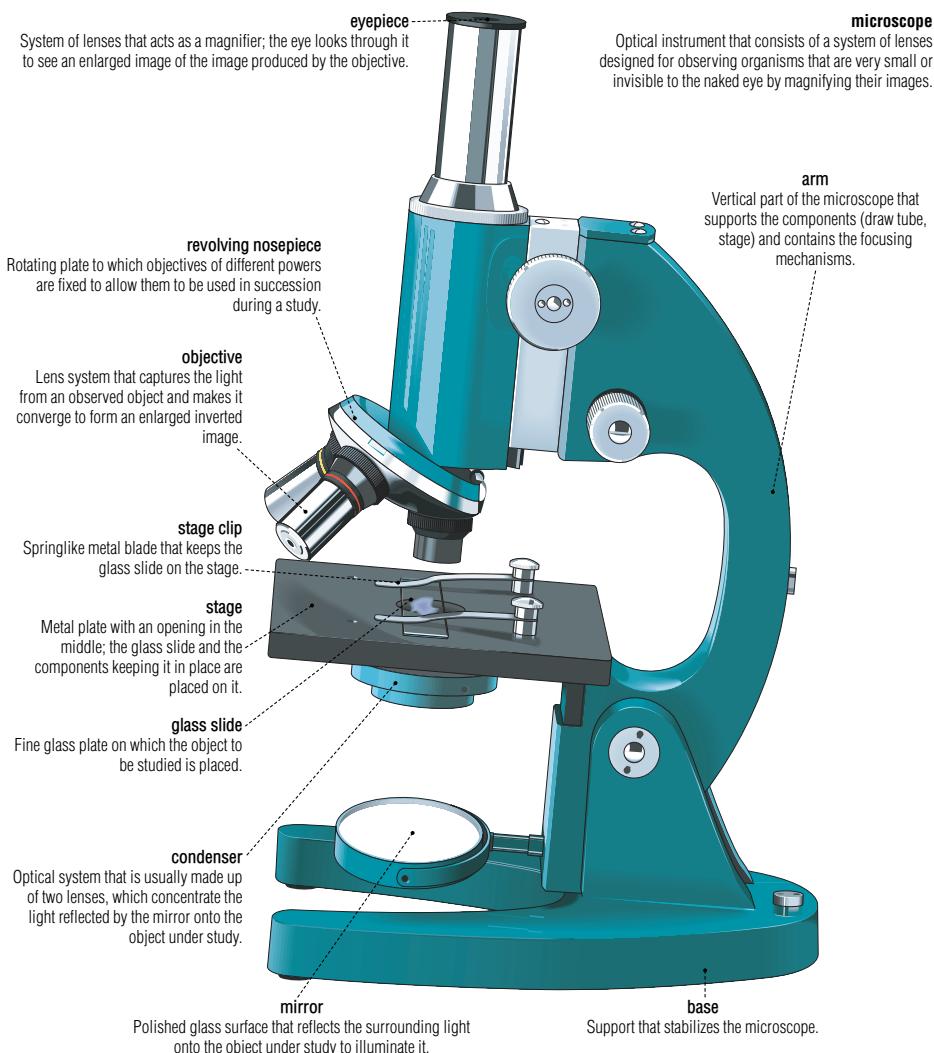
prism binoculars

Optical instrument made up of two identical telescopes, one for each eye; it magnifies both near and distant objects.



magnifying glass and microscopes

Optical instruments used to magnify the image of a near object; they range in strength from low (magnifying glass) to strong (microscope).



magnifying glass and microscopes

binocular microscope

Its two eyepieces allow both eyes to be fully applied; this provides a degree of depth to the image and prevents eyestrain.

limb top

Upper part of the arm that supports the revolving nosepiece.

objective

Lens system that captures the light from the observed object and makes it converge to form an enlarged inverted image.

stage clip

Springlike metal blade that keeps the glass slide on the stage.

glass slide

Fine glass plate on which the object to be studied is placed.

condenser adjustment knob

Screw that centers the condenser's light beam in the field of vision by moving it along a horizontal plane.

base

Support that stabilizes the microscope.

eyepiece

System of lenses that acts as a magnifier; the eye looks through it to see an enlarged image of the image produced by the objective.

draw tube

One of two cylindrical tubes that house the eyepieces; it is often made up of two converging lenses.

revolving nosepiece

Rotating plate to which objectives of different powers are fixed to allow them to be used in succession during a study.

field lens adjustment

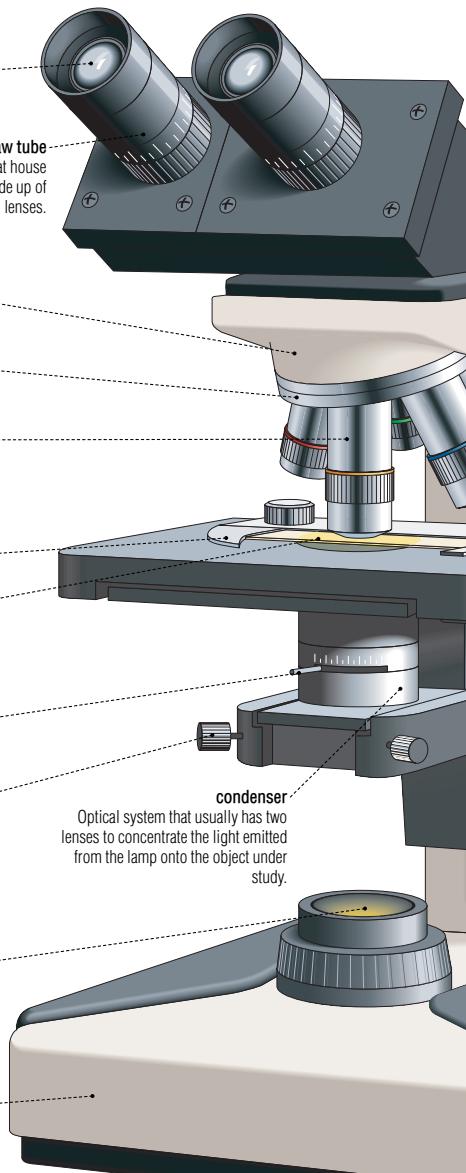
Device with a variable-diameter opening that adjusts the amount of light illuminating the object.

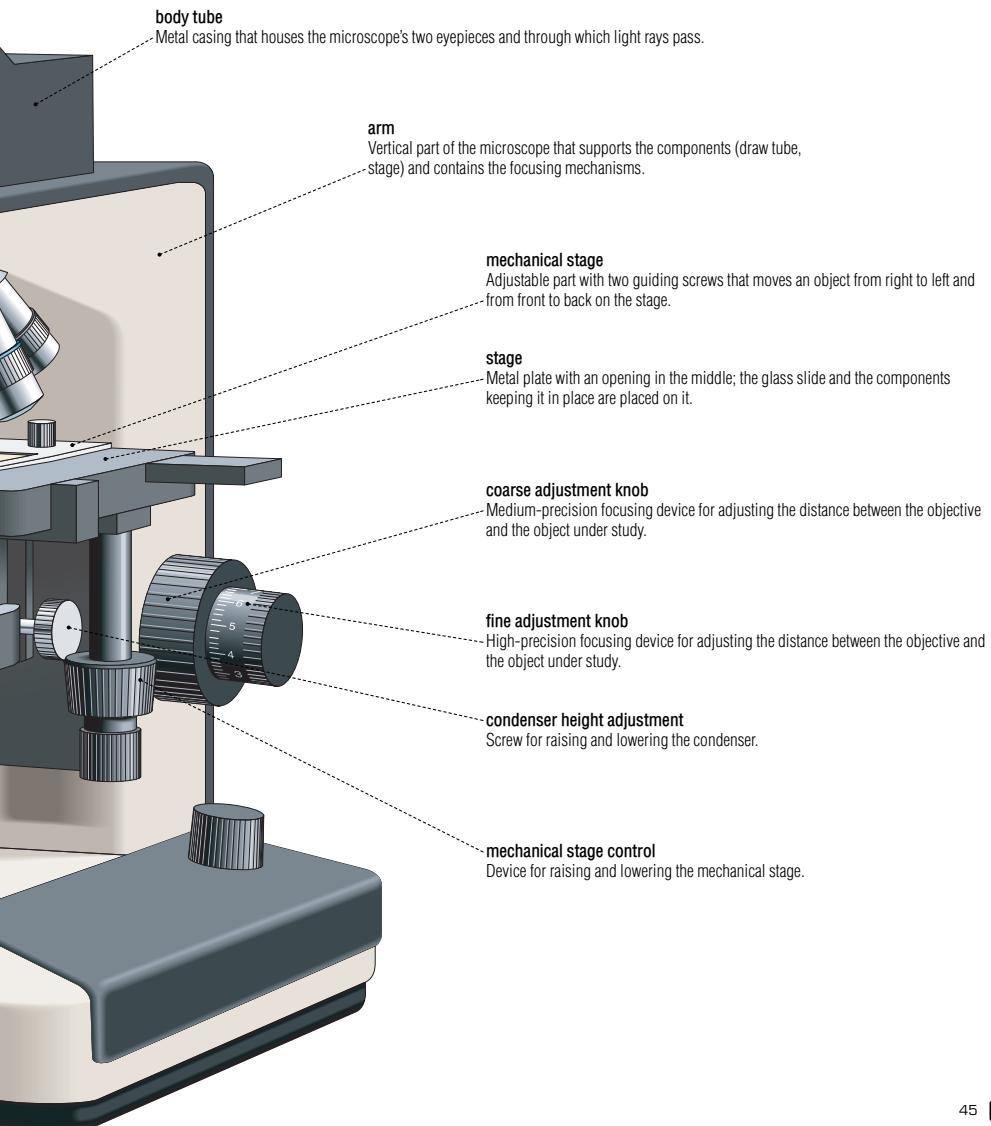
lamp

Electric device that produces a light beam to illuminate the object under study.

condenser

Optical system that usually has two lenses to concentrate the light emitted from the lamp onto the object under study.

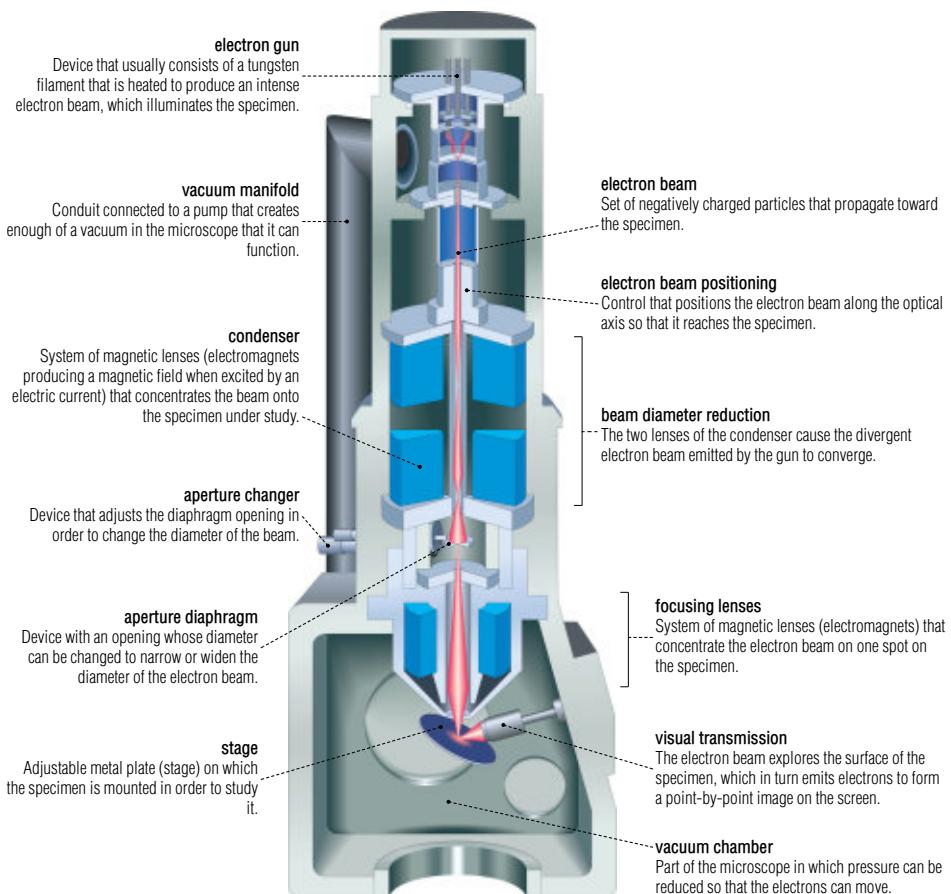


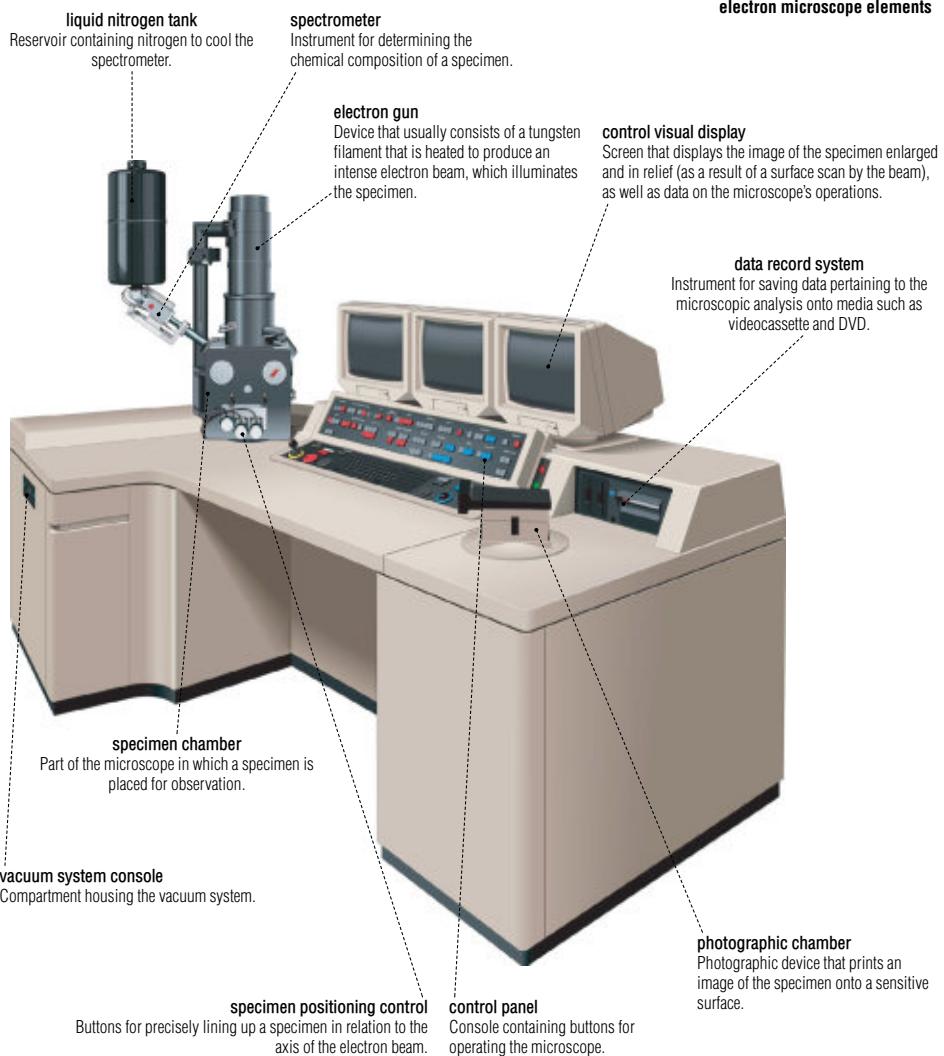


magnifying glass and microscopes

cross section of an electron microscope

Electron microscope: it uses an electron beam (as opposed to light) to provide magnification that is markedly superior to that of an optical microscope.

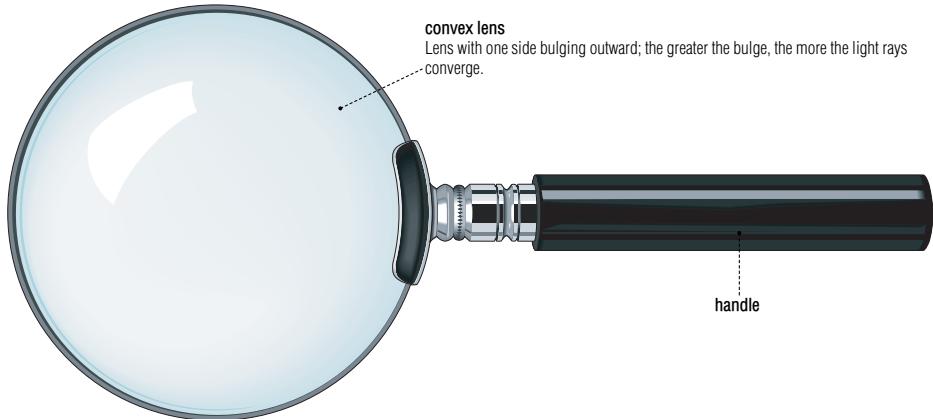




magnifying glass and microscopes

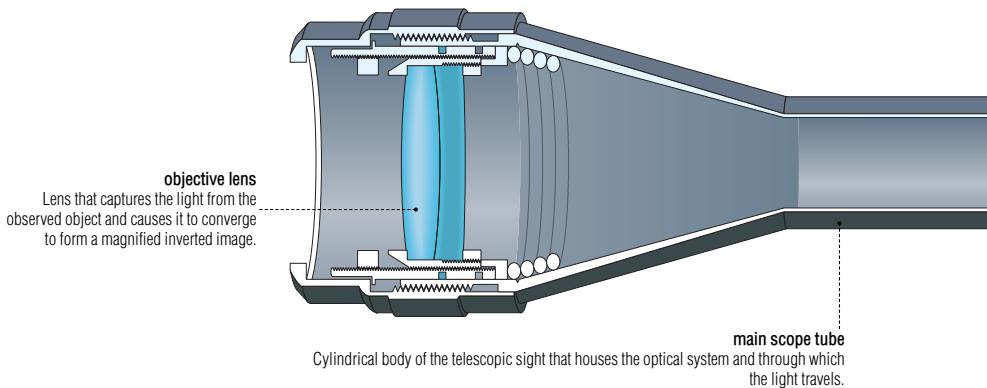
magnifying glass

Converging lens that magnifies the image of an object.



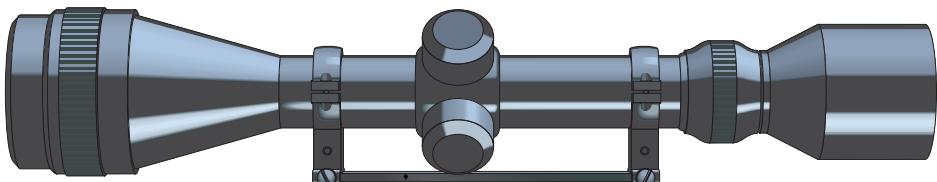
telescopic sight

Optical instrument mounted on a rifle or a measuring device to increase accuracy.



objective lens
Lens that captures the light from the observed object and causes it to converge to form a magnified inverted image.

main scope tube
Cylindrical body of the telescopic sight that houses the optical system and through which the light travels.

**dovetail**

Device for mounting the telescopic sight onto a device or firearm.

elevation adjustment

Button for positioning the sight vertically to offset any divergence of the target from the reticle.

erecting lenses

Lens system that returns the inverted image formed on the objective lens.

turret cap

Part covering and protecting an adjustment button.

field lens

Lens placed between the objective and the eyepiece to widen the field of vision.

eyepiece

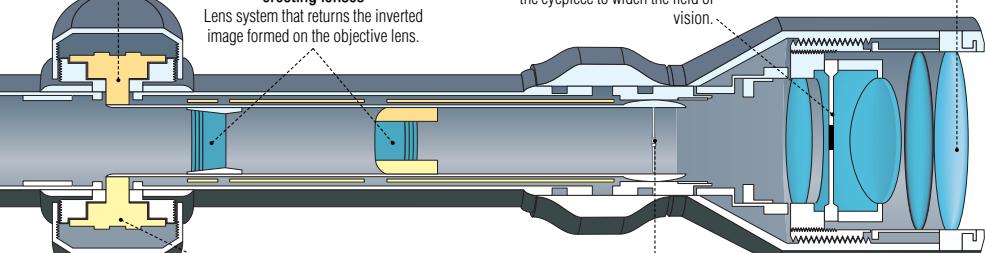
Optical disk or system of disks through which the eye sees the image produced by the lens.

reticle

Optical system made up of two fine crossed wires to create a precise point as a sighting reference.

winding adjustment

Button for positioning the sight horizontally to offset any divergence of the target from the reticle.



reflecting telescope

Optical instrument that uses an objective mirror to observe celestial bodies.



refracting telescope

Optical instrument that uses an objective lens to observe celestial bodies.

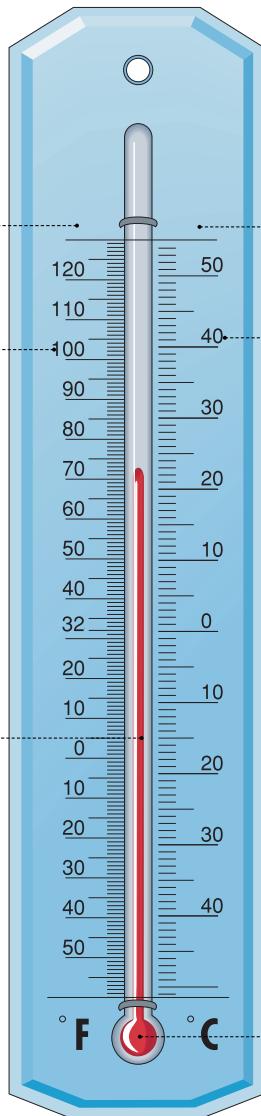


measure of temperature

Temperature: physical quantity corresponding to the level of heat or cold, which is measured by means of a thermometer.

thermometer

Instrument for measuring temperature by means of a substance (usually a liquid or a gas) contained in a graduated tube.



Fahrenheit scale

Temperature scale that is used in some English-speaking countries, on which the freezing point of water is at 32 and the boiling point at 212.

F degrees

Symbol representing a unit of measurement on the Fahrenheit scale (Fahrenheit degree).

Celsius scale

Temperature scale that is based on a graduation from 0 (freezing point of water) to 100 (boiling point of water); it was formerly called the centigrade scale.

C degrees

Symbol representing a unit of measurement on the Celsius scale (Celsius degree).

alcohol column

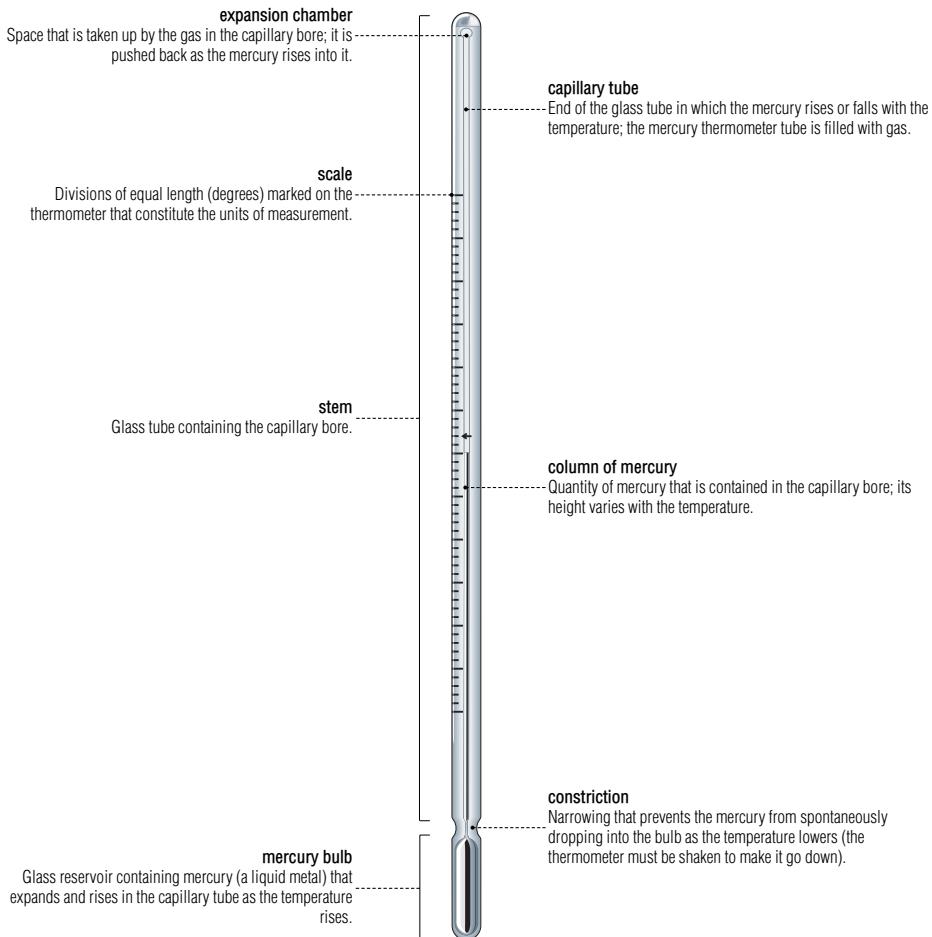
Quantity of alcohol that is contained in the glass tube; its height varies with the temperature.

alcohol bulb

Glass reservoir containing colored alcohol (methanol, ethanol) that expands and rises in the capillary bore as the temperature rises.

clinical thermometer

More precise than the alcohol thermometer, it is used to take the temperature of the human body; it is graduated from 94°F to 108°F.



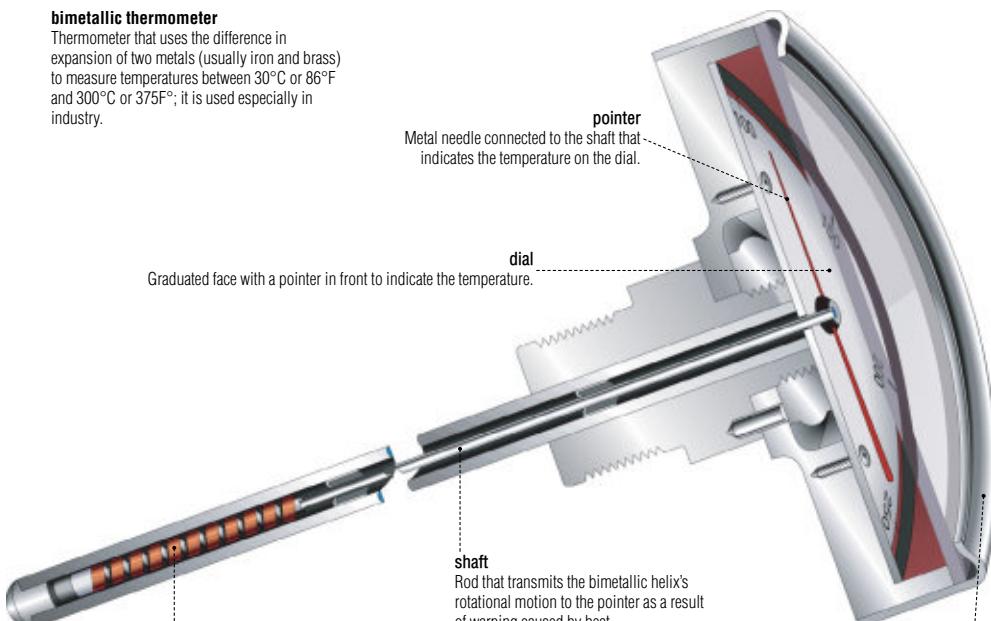
measure of temperature

digital thermometer

Thermometer that indicates the temperature in digits on a liquid crystal display screen.

**bimetallic thermometer**

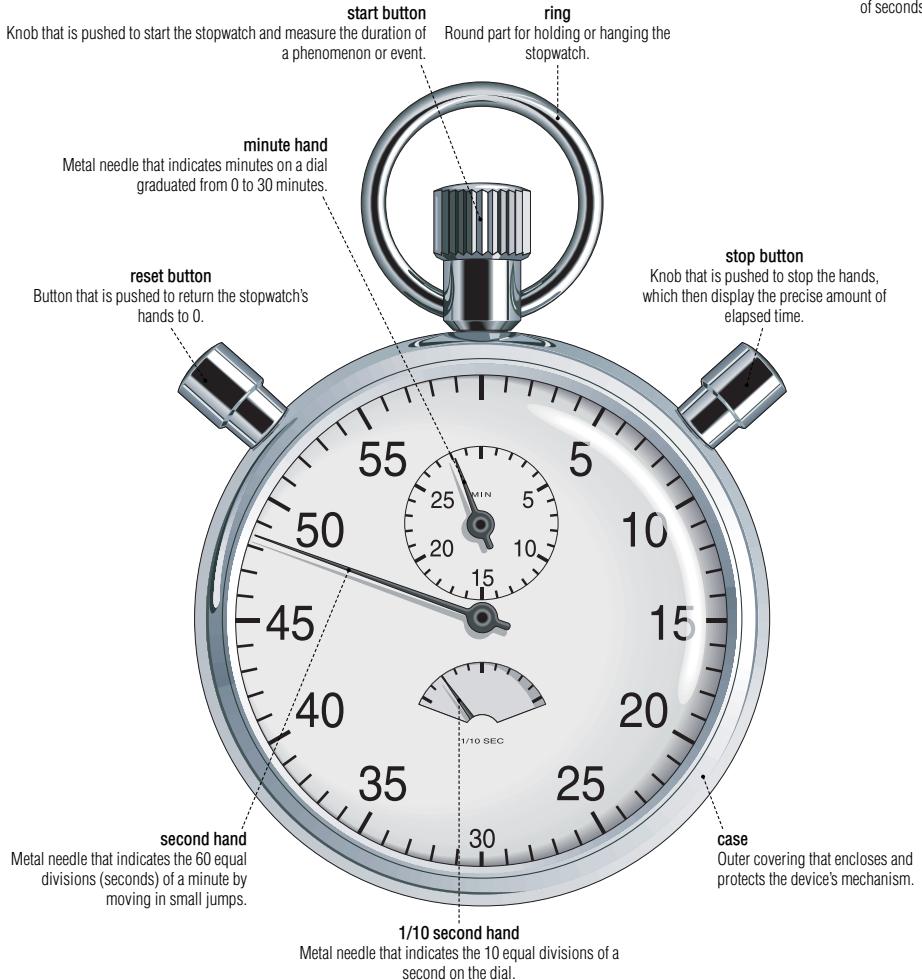
Thermometer that uses the difference in expansion of two metals (usually iron and brass) to measure temperatures between 30°C or 86°F and 300°C or 375°F; it is used especially in industry.



Time: physical quantity corresponding to a phenomenon or an event that is measured with devices such as watches and stopwatches.

stopwatch

Instrument that precisely measures time in minutes, seconds and fractions of seconds.



measure of time

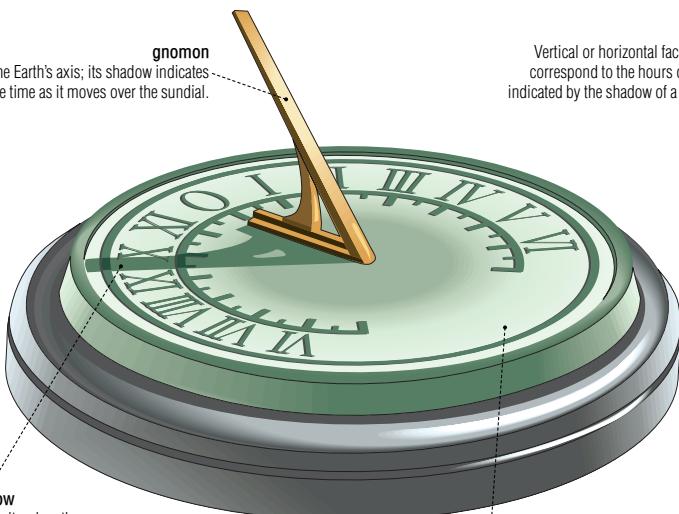
mechanical watch

Set of geared wheels that reduce the force transmitted by a spiral spring to cause the watch's hands to rotate.



sundial

Vertical or horizontal face with divisions that correspond to the hours of the day, which are indicated by the shadow of a gnomon cast by the Sun.

**shadow**

Dark area that results when the gnomon blocks the sunlight and indicates the time in accordance with the position of the Sun.

dial

Face marked with numbers over which shadows are cast by the gnomon to indicate the approximate time of day.

analog watch

The time is displayed by hands, which move around the dial.

digital watch

The time is read from letters and numbers that appear on a clear background.

**liquid crystal display**

dial

Graduated face over which the hands move to indicate the time.

**strap**

Leather, fabric, plastic or metal bracelet with a clasp; it is used to hold a watch on the wrist.

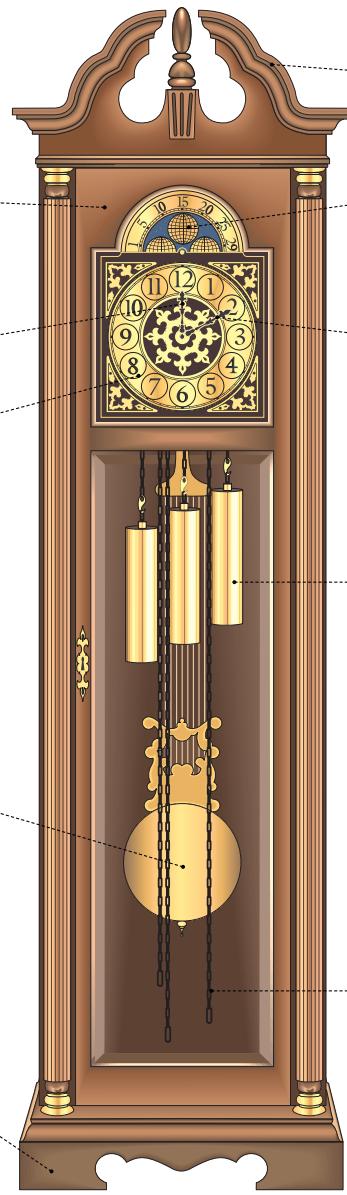
crown

Knob with sprockets that is connected to the winder; it is used to manually wind the watch and set its time.

measure of time

grandfather clock

Clock with a pendulum that is operated by weights and housed in a tall (usually over 2 m high) straight body, which stands upright on the floor.

**pediment**

Set of decorative moldings that surmount the clock.

Moon dial

Face divided into 29 1/2 days that is represented by a moon whose movement indicates the phases of the Moon: first quarter, full moon, last quarter, new moon.

minute hand

Metal needle that points at the 60 minutes of an hour on the dial.

body

Usually wooden box that houses and protects the clock's mechanism.

hour hand

Metal needle that points at the 24 hours of a day on the dial.

dial

Graduated face over which the hands move to indicate the time.

weight

Heavy body that hangs from the main wheel; its descent provides the necessary energy for the clock's mechanism.

pendulum

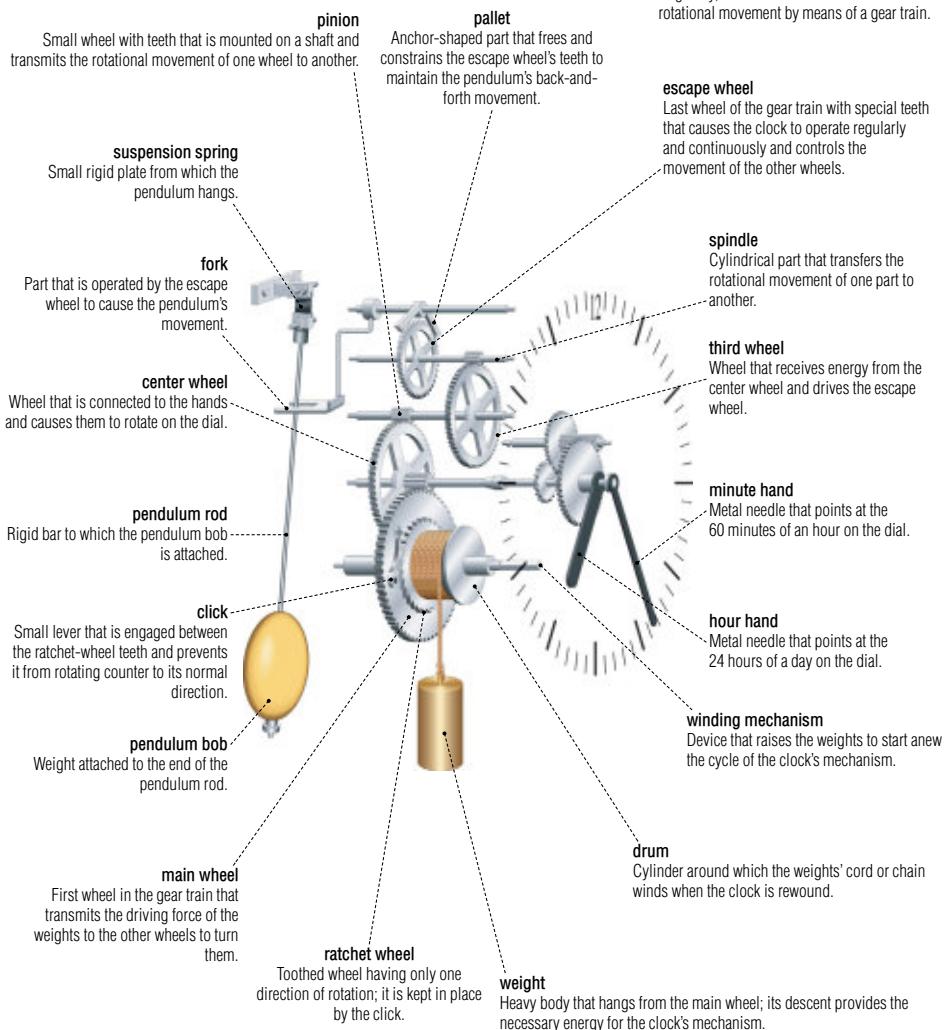
Unit whose regular swinging motion controls the workings of the clock's mechanism.

chain

Series of interlaced rings to which weights are attached.

plinth

Base that supports the clock and makes it stable.

**weight-driven clock mechanism**

This clock is operated by weights that, under gravity, drive the hands of the clock in their rotational movement by means of a gear train.

escape wheel

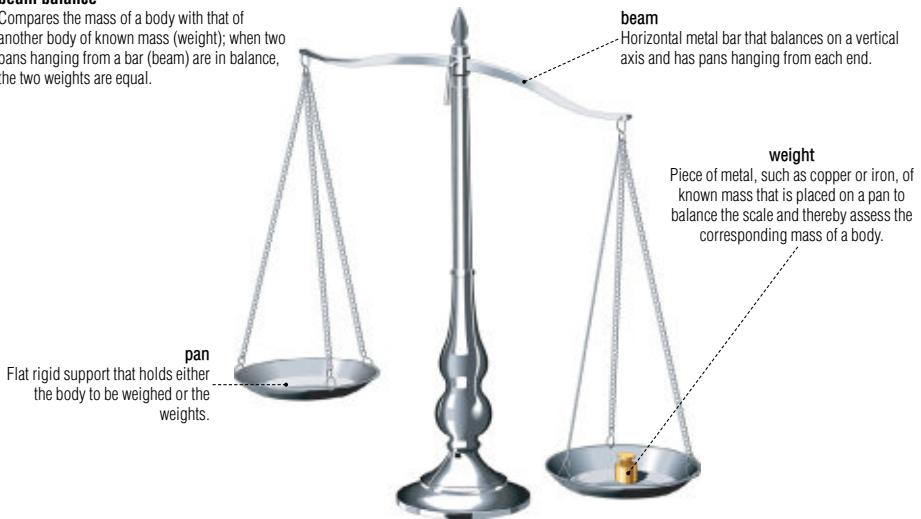
Last wheel of the gear train with special teeth that causes the clock to operate regularly and continuously and controls the movement of the other wheels.

measure of weight

Mass: physical quantity that characterizes an amount of matter (mass) that is measured by means of a scale.

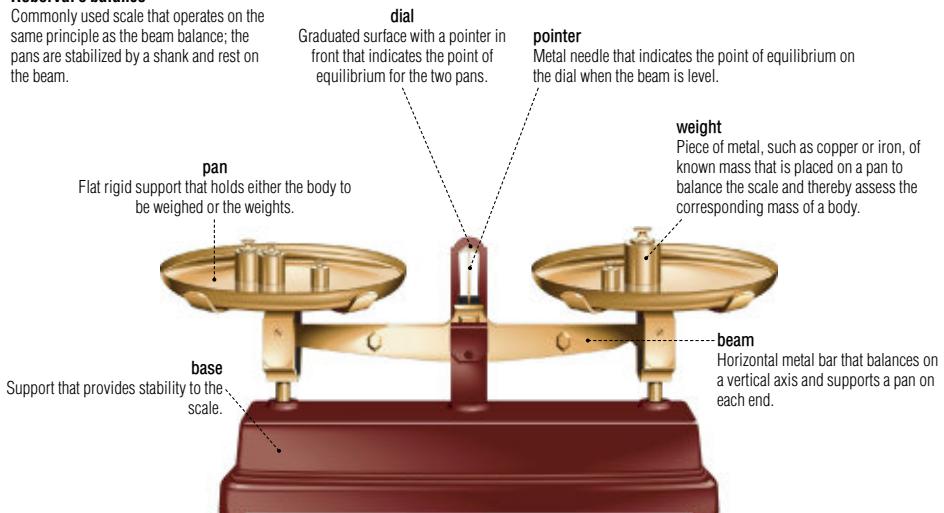
beam balance

Compares the mass of a body with that of another body of known mass (weight); when two pans hanging from a bar (beam) are in balance, the two weights are equal.



Roberval's balance

Commonly used scale that operates on the same principle as the beam balance; the pans are stabilized by a shank and rest on the beam.



steelyard

Scale used for weighing loads that has a beam with arms of different lengths; the shorter arm supports the pan and the longer arm supports the weights that slide to attain a balance.

notch

Groove in which a sliding weight catches so that a precise reading on the graduated scale can be taken.

sliding weight

Sliding part that is moved along the beams until a balance between the two masses is attained.

pan hook

Curved part from which the pan is hung by means of rods.

magnetic damping system

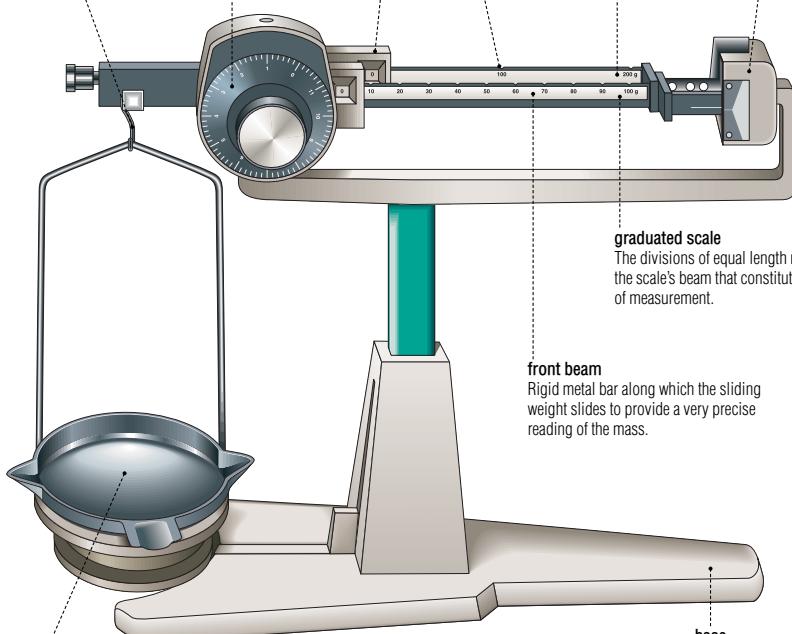
Device made up of magnets that reduce the beams' oscillations when the weights are moved to provide a quick reading of the mass.

rear beam

Rigid metal bar along which the sliding weight slides to provide a relatively precise reading of the mass.

vernier

Small graduated dial that slides along the beams and provides a very precise reading of the mass.

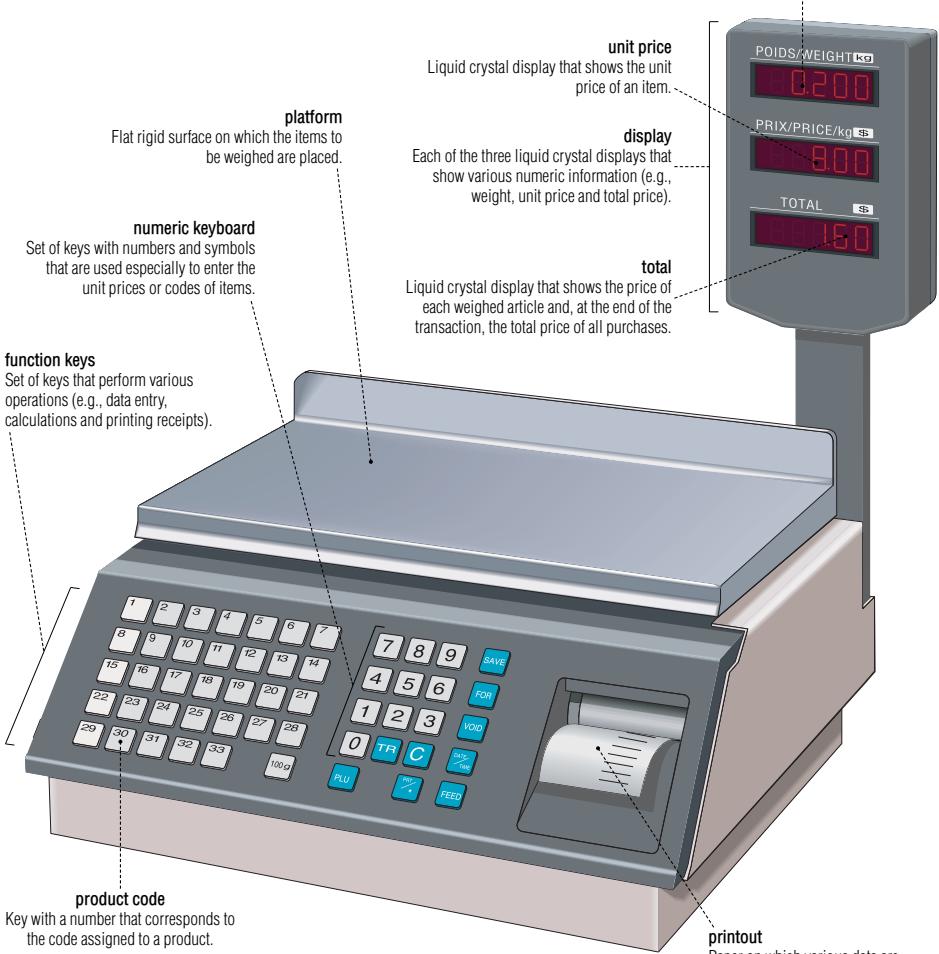


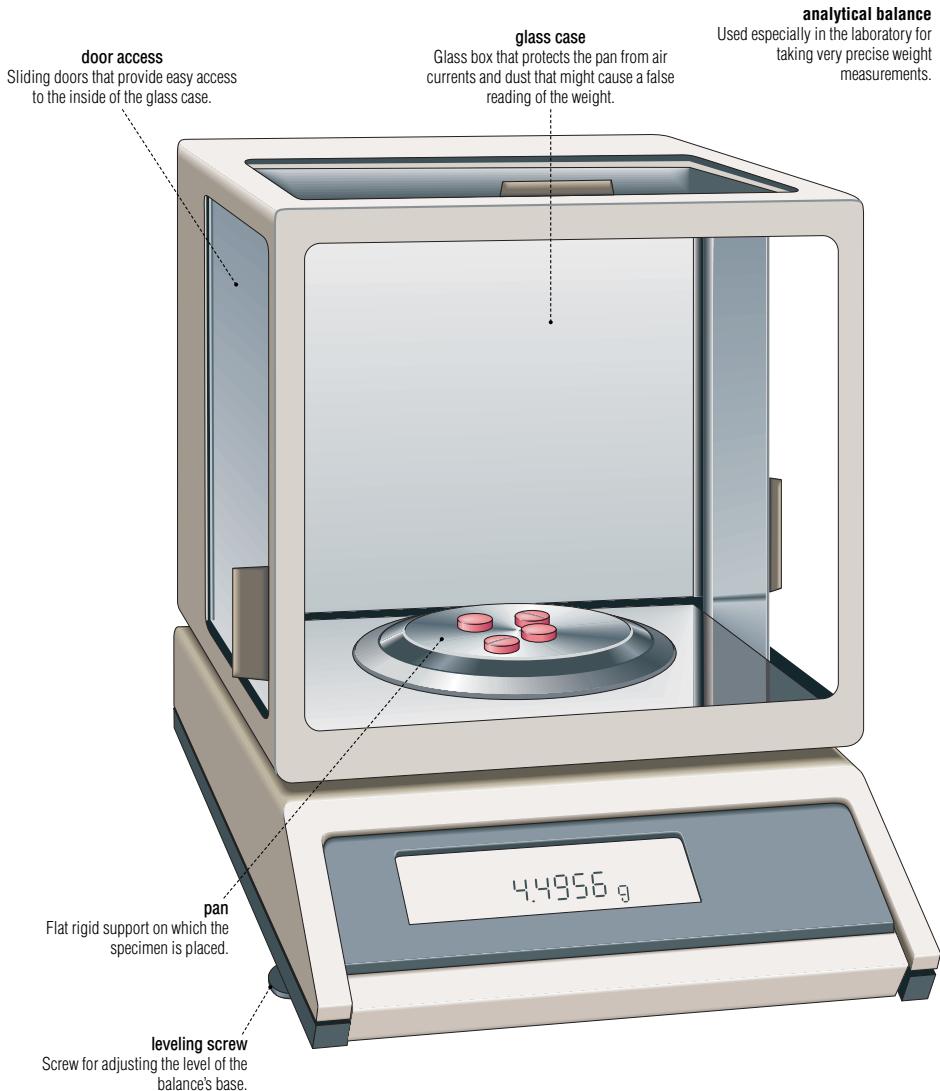
Flat rigid stand on which the body to be weighed is placed.

measure of weight

electronic scale

Commercial scale that weighs and calculates the price of a quantity of merchandise and displays these elements.





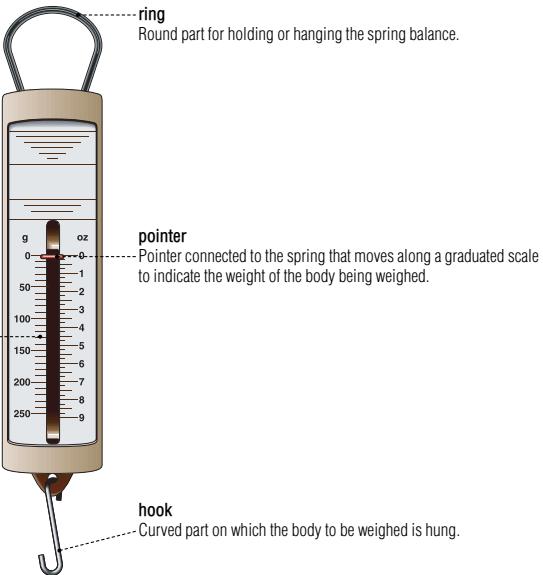
MEASURING DEVICES

measure of weight

spring balance

Scale made up of a hook attached to a spring that stretches in proportion to the weight of the object being weighed.

graduated scale
The divisions of equal length that are marked on the spring balance and constitute the units of measurement.



bathroom scale

Scale used for weighing a person; it has a spring mechanism that compresses in proportion to the weight.

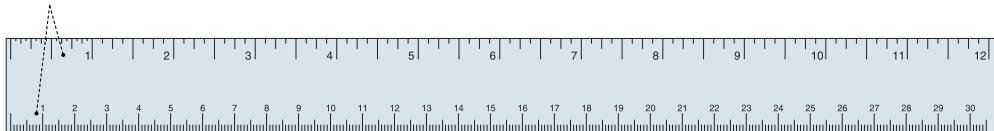


measure of length

Length: the longer dimension of an object as opposed to its width.

scale

The divisions of equal length that are marked on the ruler and constitute the units of measurement.

**ruler**

Instrument for measuring length.

measure of distance

Distance: interval separating two points in space.

pedometer

Device that counts the number of steps taken by a walker or runner to measure the distance traveled.



distance traveled
Number of steps taken by the walker or runner converted into miles.

step setting
Button for adjusting the average length of a step in the walk or run.

clip
Metal fastener for attaching the pedometer to a belt or article of clothing.

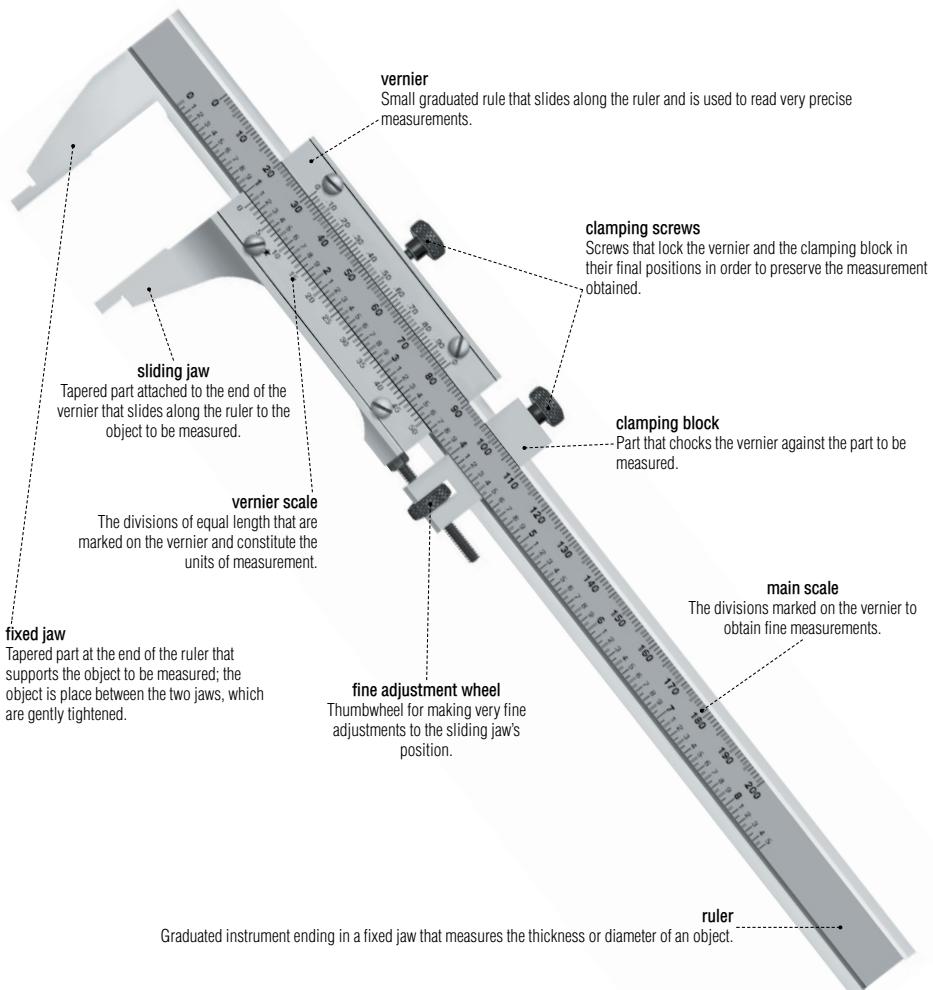
case
Outer covering that encloses and protects the device's mechanism.

measure of thickness

Thickness: dimension corresponding to the distance between two surfaces of the same body.

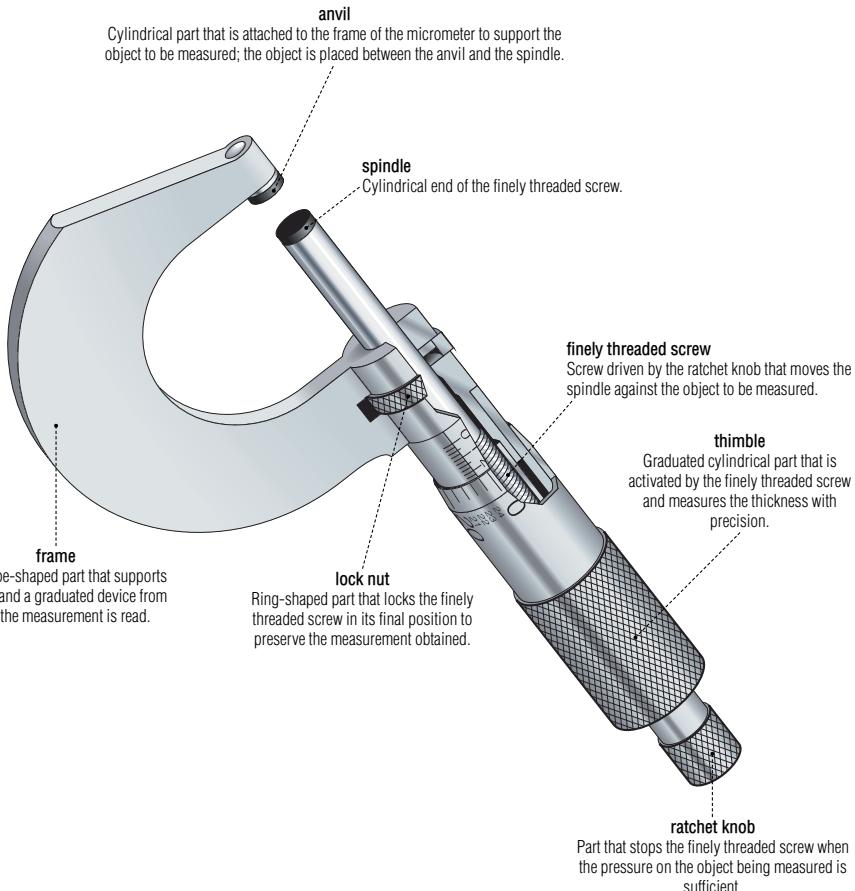
vernier caliper

Precision instrument for measuring the thickness and diameter of mechanical parts.



micrometer caliper

Instrument that measures the thickness or the diameter of relatively small parts; it produces finer results than a vernier caliper.



measure of angles

Angle: figure formed by two intersecting lines or planes; it is measured in degrees.

theodolite

Sighting instrument that is used especially in astronomy, geodesy and navigation for measuring horizontal and vertical angles.

optical sight

Device with an eyepiece that precisely aims the telescope at the target whose angles are to be measured.

alidade

Part of the theodolite that rotates on a vertical axle to measure angles by means of the telescope.

adjustment for vertical-circle image

Knob that adjusts the sharpness of the image of the vertical circle (graduated from 0° to 360°) in order to read the angles on the vertical axis.

micrometer screw

Knob that adjusts the micrometer to give a very precise reading of the circles' measurements.

adjustment for horizontal-circle image

Knob that adjusts the sharpness of the image of the horizontal circle (graduated from 0° to 360°) in order to read the angles on the horizontal axis.

horizontal clamp

Knob that locks the alidade to prevent it from rotating.

leveling head level

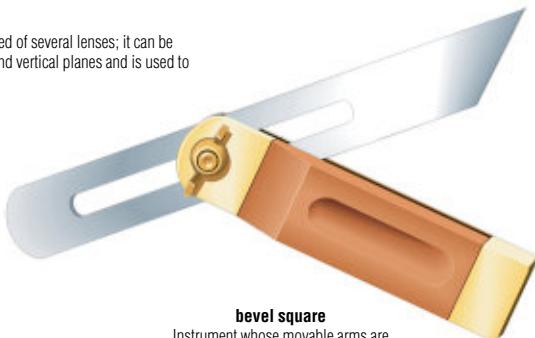
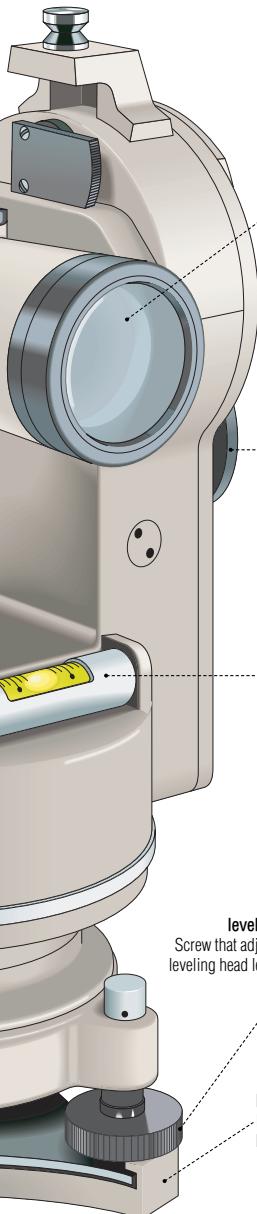
Transparent tube that contains liquid and an air bubble; it serves as a guide for positioning the leveling head on the horizontal axis.

leveling head

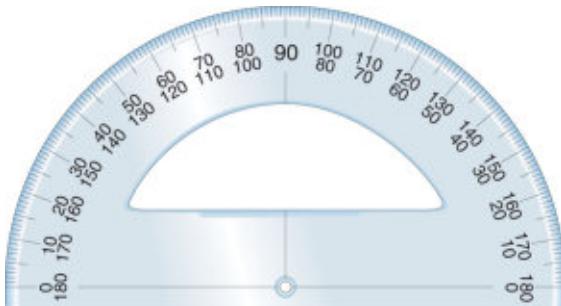
Platform serving as a support for the theodolite.

leveling head locking knob

Knob that locks the alidade to the leveling head.

**alidade level**

Transparent tube that contains liquid and an air bubble; it serves as a guide for positioning the alidade on the vertical axis.

**leveling screw**

Screw that adjusts the theodolite's leveling head level on the horizontal plane.

base plate

Plate to which the leveling head is attached by means of three leveling screws.

international system of units

Decimal system established by the 11th General Conference on Weights and Measures (GCWM) in 1960 and used by many countries.

measurement of electric potential difference



volt

measurement of frequency



hertz

Frequency of a periodic phenomenon whose period is 1 second.

measurement of electric charge



coulomb

Amount of electricity carried in 1 second by a current of 1 ampere.

measurement of energy



joule

Amount of energy released by the force of 1 newton acting through a distance of 1 meter.

measurement of power



watt

Energy transfer of 1 joule during 1 second.

measurement of force



newton

Force required to impart an acceleration of 1 m/s^2 to a body having a mass of 1 kg.

measurement of electric resistance



ohm

Electrical resistance between two points of a conductor carrying a current of 1 ampere when the difference in potential between them is 1 volt.

measurement of electric current



ampere

Constant current of 1 joule per second in a conductor.

measurement of length**m****meter**

Distance traveled by light in a vacuum in $1/299,792,458$ of a second.

measurement of mass**kg****kilogram**

Mass of a platinum prototype that was accepted as the international reference in 1889; it is stored at the International Bureau of Weights and Measures.

measurement of Celsius temperature**°C****degree Celsius**

Division into 100 parts of the difference between the freezing point of water (0°C) and its boiling point (100°C) at standard atmospheric pressure.

measurement of thermodynamic temperature**K****kelvin**

Zero degrees Kelvin is equal to minus 273.16°C .

measurement of pressure**Pa****pascal**

Uniform pressure exerted on a flat surface of 1 m^2 with a force of 1 newton.

measurement of amount of substance**mol****mole**

Quantity of matter equal to the number of atoms in 0.012 kg of carbon 12.

measurement of radioactivity**Bq****becquerel**

Radioactivity of a substance in which one atom disintegrates per second.

measurement of luminous intensity**cd****candela**

Unit of light intensity equivalent to a radiant intensity of $1/683$ watts per steradian (solid angle).

mathematics

The science that uses deductive reasoning to study the properties of abstract entities such as numbers, space and functions and the relations between them.

**minus/negative**

Sign denoting that a number is to be subtracted from another; the result is a difference.

**plus/positive**

Sign denoting that a number is to be added to another; the result is a sum.

**multipled by**

Sign denoting that a number is to be multiplied by another; the result is a product.

**divided by**

Sign denoting that a number (dividend) is to be divided by another (divisor); the result is a quotient.

**equals**

Sign denoting the result of an operation.

**is not equal to**

Sign denoting that the result of an operation is not close to the same value as the one on the right.

**is approximately equal to**

Sign denoting that the result of an operation is close to the same value as the one on the right.

**is equivalent to**

Sign denoting that the value on the left is the same magnitude as the one on the right.

**is identical with**

Binary sign denoting that the result of the operation noted on the left has the same value as the operation noted on the right.

**is not identical with**

Binary sign denoting that the result of the operation noted on the left does not have the same value as the operation noted on the right.

**empty set**

Sign denoting that a set contains no elements.

**union of two sets**

Binary sign denoting that a set is composed of the sum of the elements of two sets.

**intersection of two sets**

Binary sign denoting that two sets M and N have elements in common.

**is included in/is a subset of**

Binary sign denoting that a set A on the left is part of the set B on the right.

**plus or minus**

Sign denoting that the number that follows denotes an order of magnitude.

**is less than or equal to**

Sign denoting that the result of an operation is equal to or of smaller magnitude than the number that follows.

**is greater than**

Sign denoting that the value on the left is of greater magnitude than the number that follows.

**is greater than or equal to**

Sign denoting that the result of an operation is equal to or of greater magnitude than the number that follows.

**is less than**

Sign denoting that the value on the left is of smaller magnitude than the number that follows.

**percent**

Sign denoting that the number preceding it is a fraction of 100.

**is an element of**

Binary sign denoting that the element on the left is included in the set on the right.

**is not an element of**

Binary sign denoting that the element on the left is not included in the set on the right.

**sum**

Sign indicating that several values are to be added together (their sum).

**square root of**

Sign denoting that, when a number is multiplied by itself, the result is the number that appears below the bar.

**fraction**

Sign denoting that the number on the left of the slash (numerator) is one part of the number on the right of the slash (denominator).

**infinity**

Symbol denoting that a value has no upper limit.

**integral**

Result of the integral calculation used especially to determine an area and to resolve a differential equation.

**factorial**

Product of all positive whole numbers less than and equal to a given number. For example, the factorial of 4 is: $4! = 1 \times 2 \times 3 \times 4 = 24$.

mathematics

Roman numerals

Uppercase letters that represented numbers in ancient Rome; they are still seen today in uses such as clock and watch dials and pagination.



one

Letter whose value is 1 unit.



five

Letter whose value is 5 units.



ten

Letter whose value is 10 units.



fifty

Letter whose value is 50 units.



one hundred

Letter whose value is 100 units.



five hundred

Letter whose value is 500 units.



one thousand

Letter whose value is 1,000 units.

biology

The scientific study of living organisms (humans, animals and plants) from the point of view of their structure and how they function and reproduce.



male

Symbol denoting that a being has male reproductive organs.



female

Symbol denoting that a being has female reproductive organs.



blood factor positive

Individuals are Rh positive when their red blood cells carry an Rh molecule (antigen); the Rh factor is positive in about 85% of the population.



blood factor negative

Individuals not carrying the Rh molecule (antigen) are Rh negative; the Rh factor plays an important role in pregnancy (the parents' factors must be compatible).



death

Symbol placed before a date denoting a person's year of death.



birth

Symbol placed before a date denoting a person's year of birth.

Mathematical discipline that studies the relations between points, straight lines, curves, surfaces and volumes.

**degree**

Symbol placed in superscript after a number to denote the opening of an angle or the length of an arc, or in front of an uppercase letter to identify a scale of measurement.

**minute**

Symbol placed in superscript after a number that denotes degrees in sixtieths of a measure.

**second**

Symbol placed in superscript after a number that denotes degrees in sixtieths of a minute.

**pi**

Constant that represents the ratio of a circle's circumference to its diameter; its value is approximately 3.1416.

**perpendicular**

Symbol denoting that a straight line meets another at a right angle.

**is parallel to**

Symbol denoting that two straight lines remain at a constant distance from one another.

**is not parallel to**

Symbol denoting that two straight lines do not remain at a constant distance from one other.

**right angle**

Angle formed by two lines or two perpendicular planes that measures 90°.

**obtuse angle**

Angle between 90° and 180°.

**acute angle**

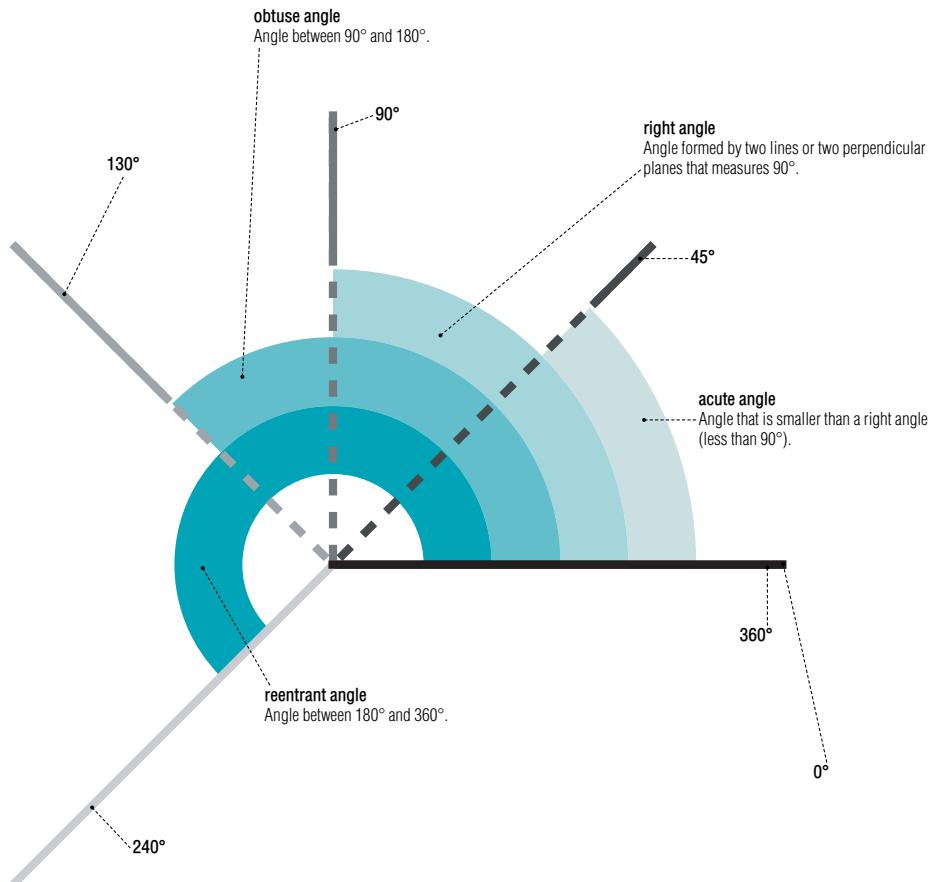
Angle that is smaller than a right angle (less than 90°).

geometrical shapes

Drawings that represent various geometric forms such as straight lines, circles and polygons.

examples of angles

Angle: figure formed by two intersecting lines or planes; it is measured in degrees.



plane surfaces

Set of points on a plane that describes an area of space.

parts of a circle

Circle: closed plane curve; all its points are the same distance from a fixed point (center).

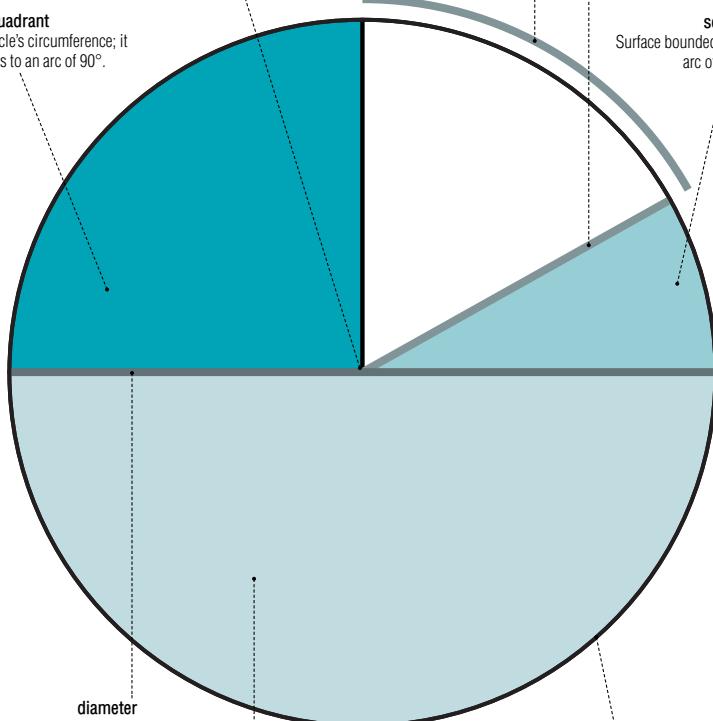
center
Point located at the same distance from every point on the circle's circumference.

quadrant
Quarter of a circle's circumference; it corresponds to an arc of 90° .

arc
Section of a circle between two points on the circle.

radius
Line that joins a point on a circle's circumference to its center; it is one half of the diameter.

sector
Surface bounded by two radii and an arc of a circle.



diameter
Line that connects two points on a circle's circumference and passes through its center.

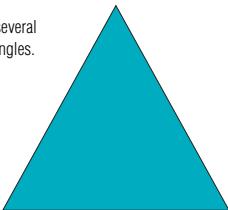
semicircle
A half circle that is delimited by its diameter.

circumference
Length of a circle that corresponds to the product of its diameter and pi.

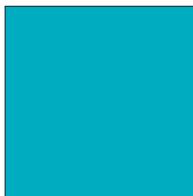
geometrical shapes

polygons

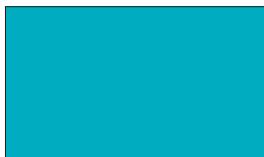
Geometric plane figures with several sides and a number of equal angles.

**triangle**

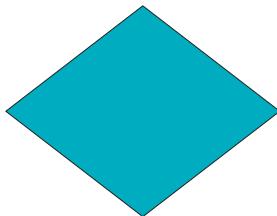
Three-sided polygon; triangles are scalene (no side is equal to any other) isosceles (two sides equal) or equilateral (all sides equal).

**square**

Equilateral rectangle with four right angles.

**rectangle**

Quadrilateral whose opposite sides are equal in length; the sides meet at right angles.

**rhombus**

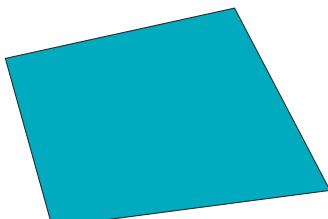
Equilateral parallelogram.

**trapezoid**

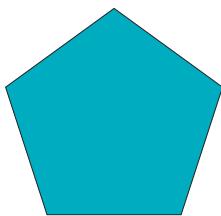
Quadrilateral with two sides (bases) that are parallel. It is isosceles when it has two sides that are not parallel and equal, and rectangle when two of its sides form a right angle.

**parallelogram**

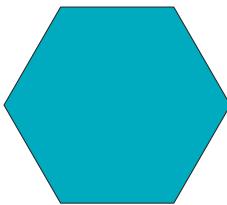
Trapezoid whose opposite sides are parallel and of equal length; the sides do not meet at right angles.

**quadrilateral**

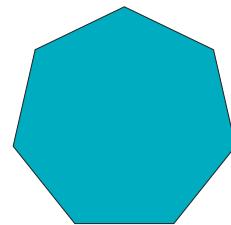
Any plane figure with four sides and four angles.

**regular pentagon**

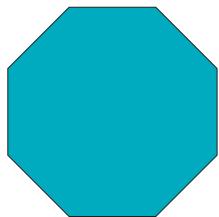
Polygon with five (penta = five) sides and equal angles.

**regular hexagon**

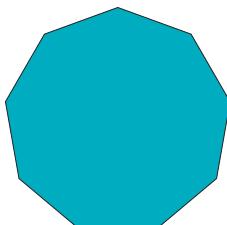
Polygon with six (hexa = six) sides and equal angles.

**regular heptagon**

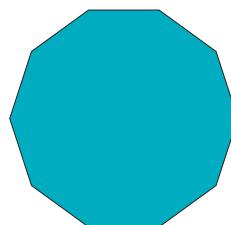
Polygon with seven (hepta = seven) sides and equal angles.

**regular octagon**

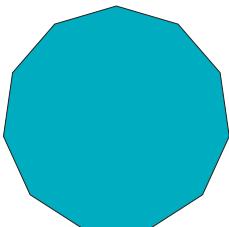
Polygon with eight (octo = eight) sides and equal angles.

**regular nonagon**

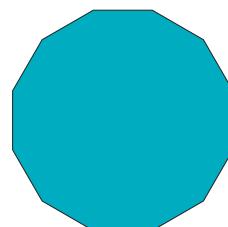
Polygon with nine (nona = nine) sides and equal angles.

**regular decagon**

Polygon with 10 (deca = ten) sides and equal angles.

**regular hendecagon**

Polygon with 11 (hendeca = eleven) sides and equal angles.

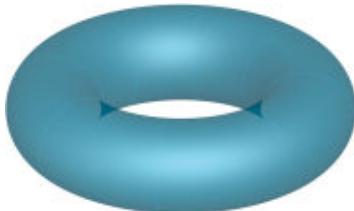
**regular dodecagon**

Polygon with 12 (dodeca = twelve) sides and equal angles.

geometrical shapes

solids

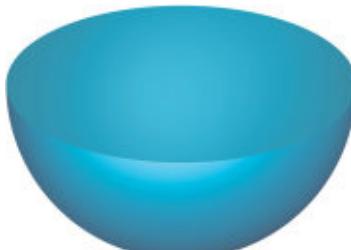
Geometric shapes in three dimensions that are delimited by surfaces.

**torus**

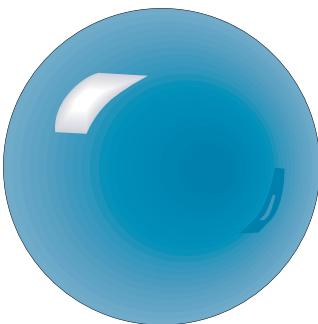
Volume or solid generated by the rotation of a circle at an equal distance from its center of rotation.

**helix**

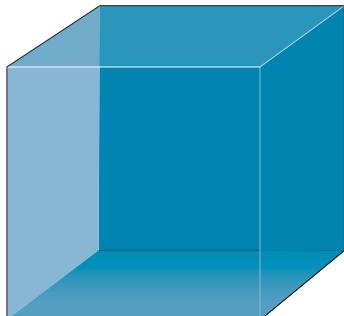
Volume or solid of spiral shape that turns toward the left at a constant angle.

**hemisphere**

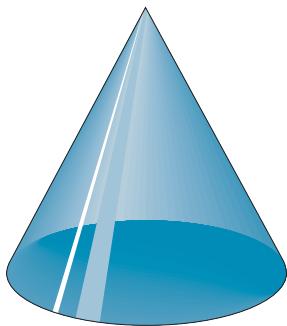
Half sphere cut along its diameter.

**sphere**

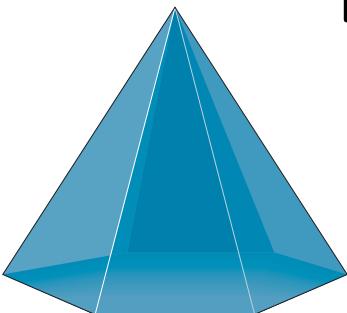
Volume with all the points on its surface the same distance from its center; the solid thus delimited is a round ball.

**cube**

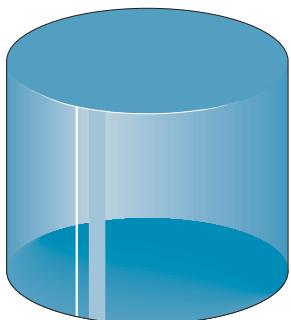
Volume or solid with six square sides of equal area and six equal edges; it has eight vertices.

**cone**

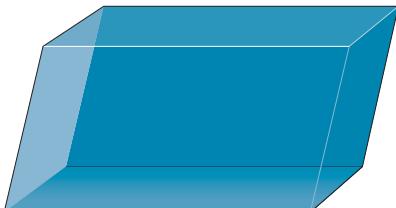
Volume or solid generated by the rotation of a straight line (generatrix) along a circular line (directrix) from a fixed point (vertex).

**pyramid**

Volume or solid generated by straight lines (edges) connecting the angles of a polygon (base) to the vertex and whose sides form triangles.

**cylinder**

Volume or solid generated by the rotation of a straight line (generatrix) moving along a curved line (directrix).

**parallelepiped**

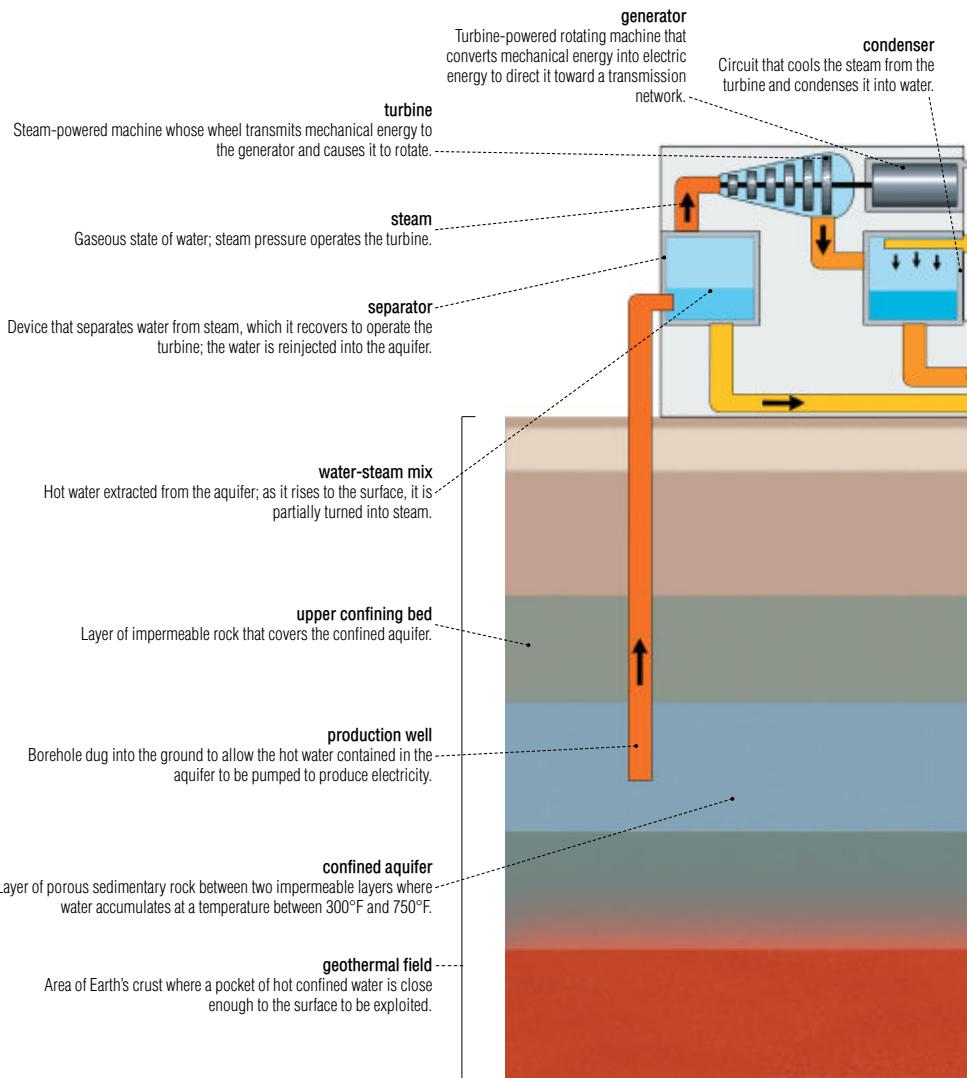
Volume or solid with six sides (parallelograms) that are parallel in pairs.

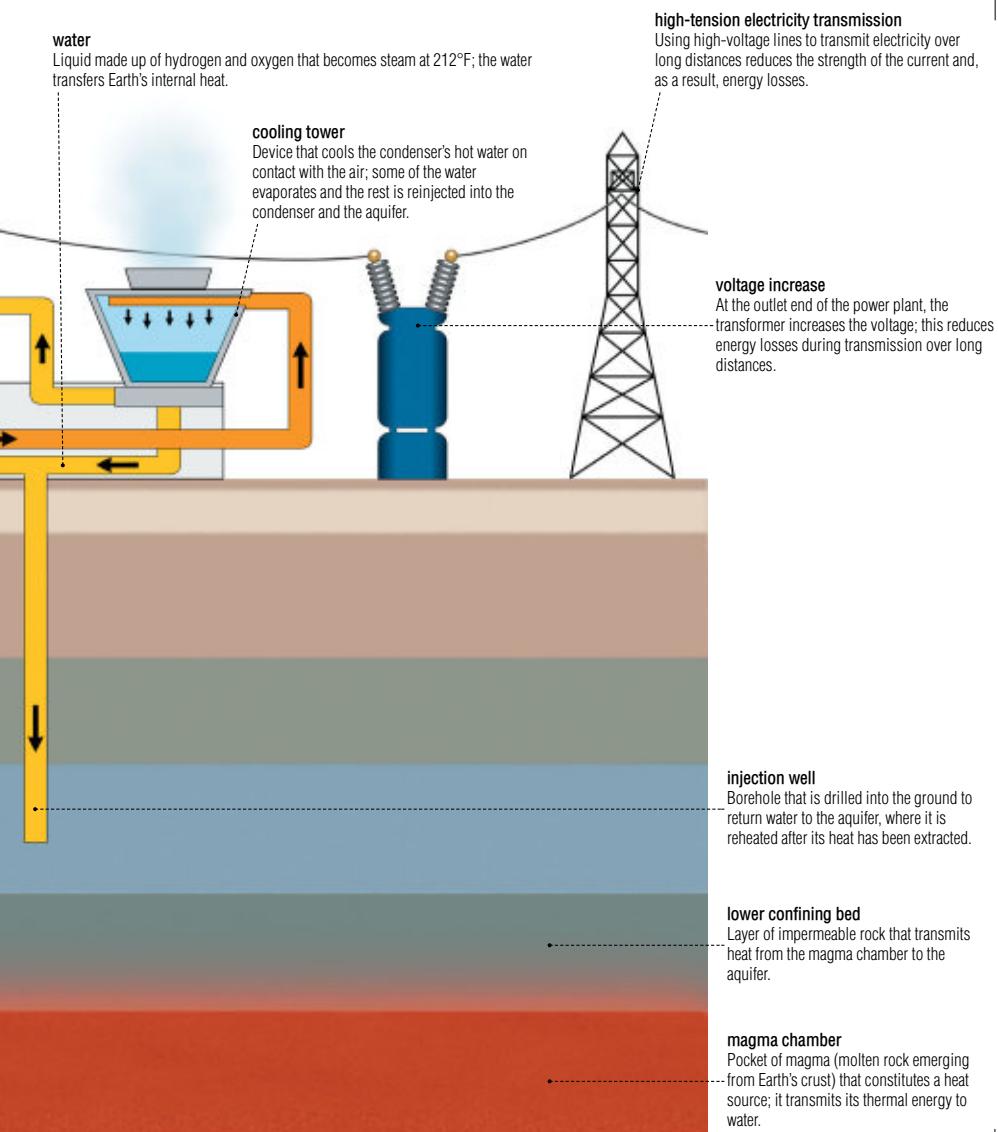
**regular octahedron**

Volume or solid with eight triangular sides of equal area; it has six vertices and 12 edges.

production of electricity from geothermal energy

Hot water contained in the ground near a volcano, geyser or thermal source is piped to the surface by drilling to extract steam and produce electricity.



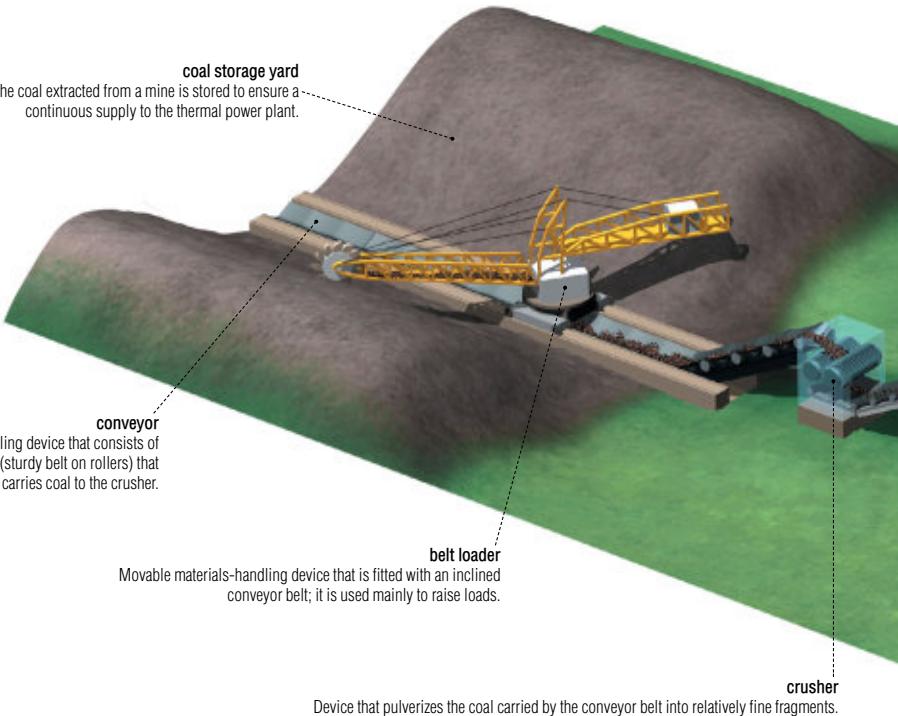


thermal energy

Energy that is produced by turning water into steam through the burning of fuel (e.g., petroleum and coal) or through nuclear reaction.

production of electricity from thermal energy

The heat that is given off by burning combustible fuels in the thermal power plant converts water into steam; the steam turns a turbo-alternator unit to produce electricity.



pulverizer

Device that pulverizes coal into a very fine powder so that it burns more easily in the steam generator.

steam generator

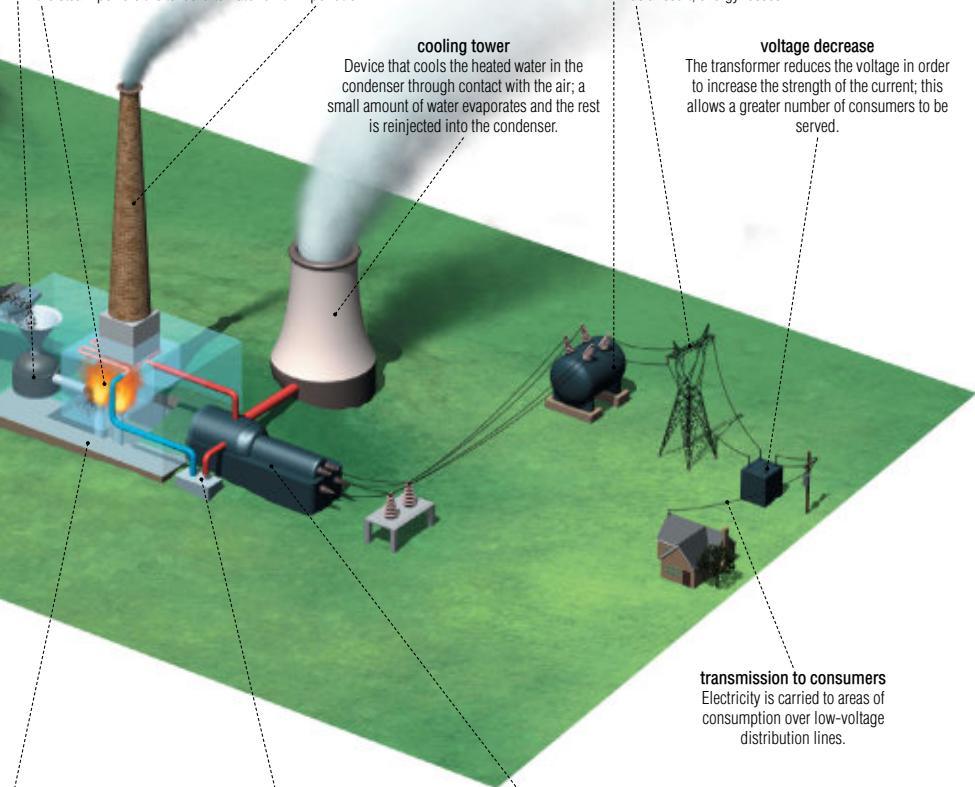
Device that uses the heat produced from burning coal to convert water into steam; the steam powers the turbo-alternator unit.

stack

Pipe through which gases produced by burning coal are discharged; these gases are first partially cleaned to reduce pollution.

cooling tower

Device that cools the heated water in the condenser through contact with the air; a small amount of water evaporates and the rest is reinjected into the condenser.

**coal-fired thermal power plant**

Plant that produces electricity from thermal energy by burning coal.

condenser

Circuit that cools the steam from the turbine and condenses it into water, which is reintroduced into the steam generator.

voltage increase

At the outlet end of the power plant, the transformer increases the voltage; this reduces energy losses during transmission over long distances.

high-tension electricity transmission

Using high-voltage lines to transmit electricity over long distances reduces the strength of the current and, as a result, energy losses.

voltage decrease

The transformer reduces the voltage in order to increase the strength of the current; this allows a greater number of consumers to be served.

transmission to consumers

Electricity is carried to areas of consumption over low-voltage distribution lines.

turbo-alternator unit

Device with a turbine that transmits the water's mechanical energy to the alternator's rotor to make it turn to produce electricity.

coal mine

The underground or open-pit facilities that are set up around a coal deposit in order to extract it.

open-pit mine

Type of mining that is used for shallow deposits; coal or ore is extracted by digging a succession of benches from the surface of the ground downward.

bench

The levels of a quarry that are arranged like steps of a staircase and from which coal or ore is extracted.

face

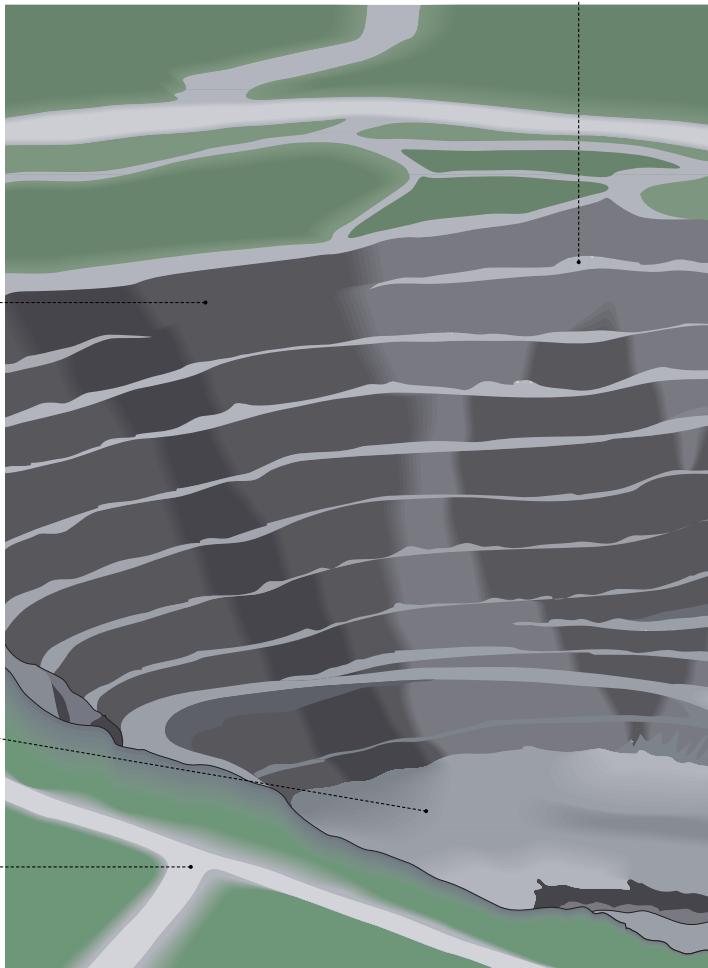
Vertical surface created by dynamiting a deposit to extract its ore.

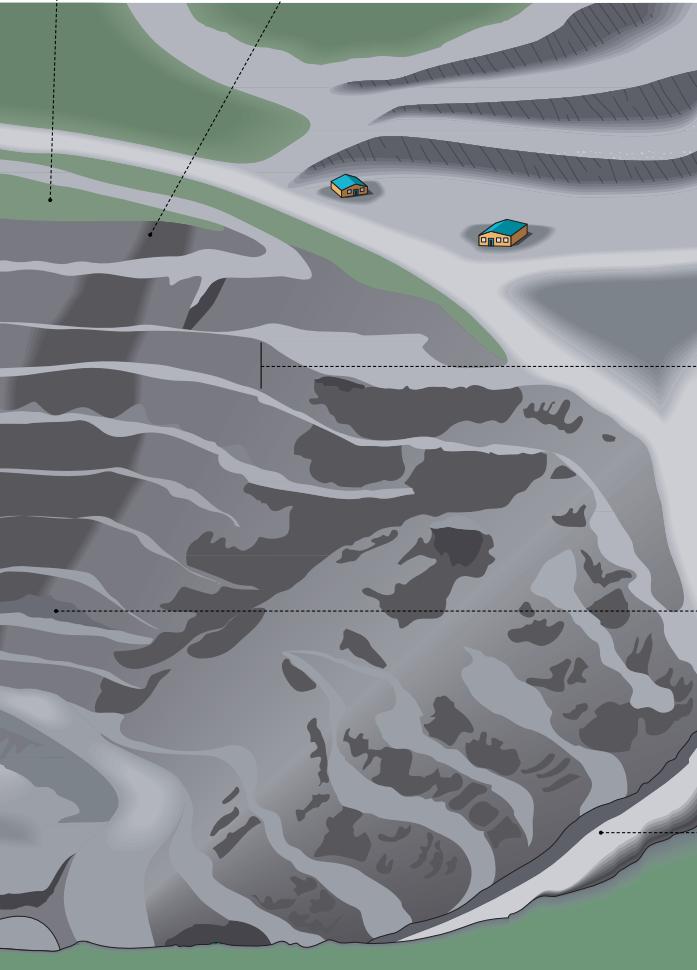
crater

Depression that forms the bottom of the quarry; it is a result of the extraction of deposits.

haulage road

Access road leading to the quarry; it is used to haul coal to the treatment plant.



**ground surface**

The land that covers the deposit.

overburden

Part of the ground that covers the ore beds; it is removed to reach the deposit.

bench height

Vertical distance between the horizontal planes of two benches.

ore

Solid fossil fuel that is black and contains a large amount of carbon.

ramp

Roadway between two benches; it is inclined so that motorized vehicles can remove the ore extracted from the various levels.

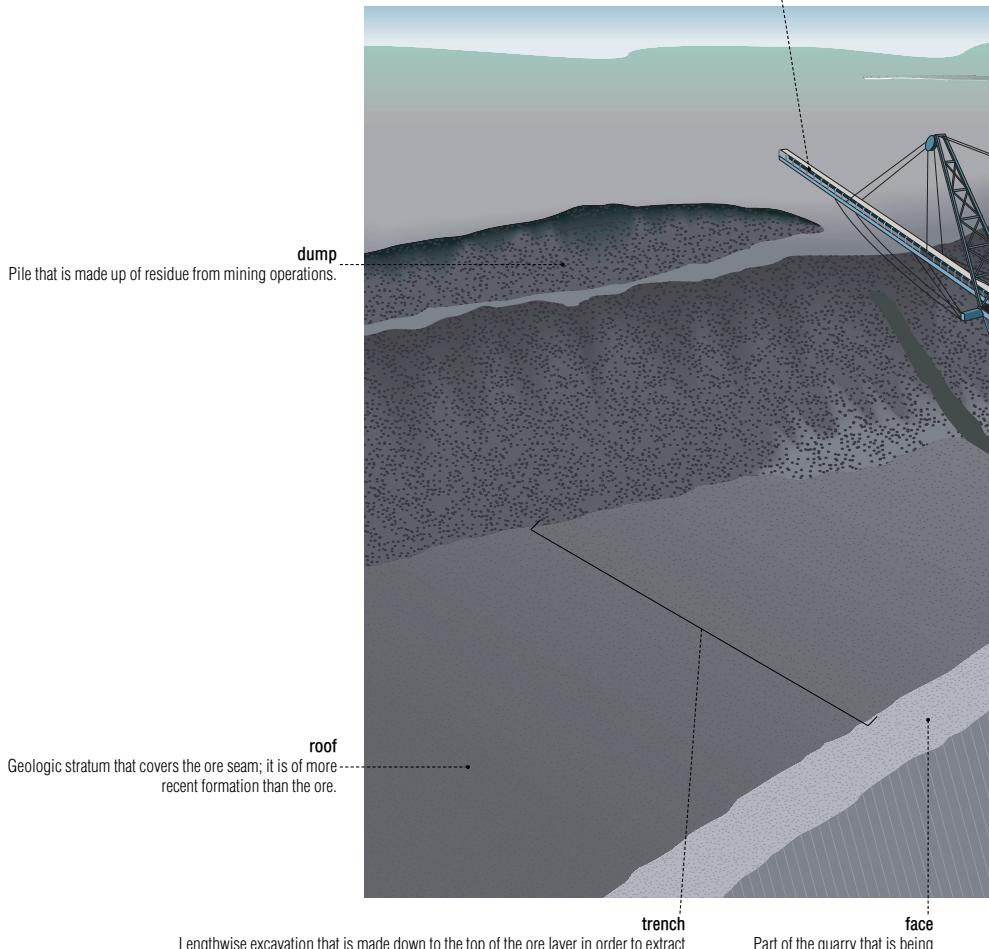
coal mine

strip mine

Type of mining that is used especially for large shallow deposits; coal or ore is extracted by digging a trench in the ground surface.

conveyor

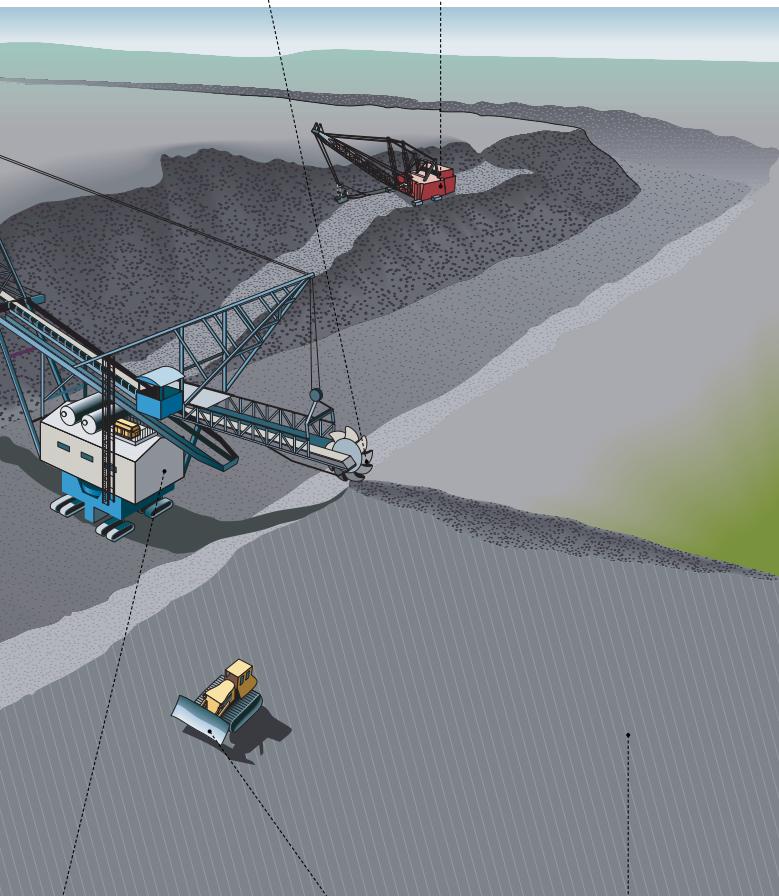
Materials-handling device that consists of a conveyor belt (sturdy belt on rollers) that is used to transport coal extracted from the mine.



roof
Geologic stratum that covers the ore seam; it is of more recent formation than the ore.

trench
Lengthwise excavation that is made down to the top of the ore layer in order to extract its coal.

face
Part of the quarry that is being excavated and from which ore is progressively extracted.

**bucket wheel excavator**

Earthmover that consists of a wheel fitted with buckets (scoops); it is used to dig into rock to extract materials, which are then dumped onto a conveyor.

mechanical shovel

Earthmover that consists of a movable cab with an articulated arm fitted with a bucket (scoop); it is used for digging and handling loads.

belt loader

Movable materials-handling device that is fitted with an inclined conveyor belt; it is used mainly to raise loads.

bulldozer

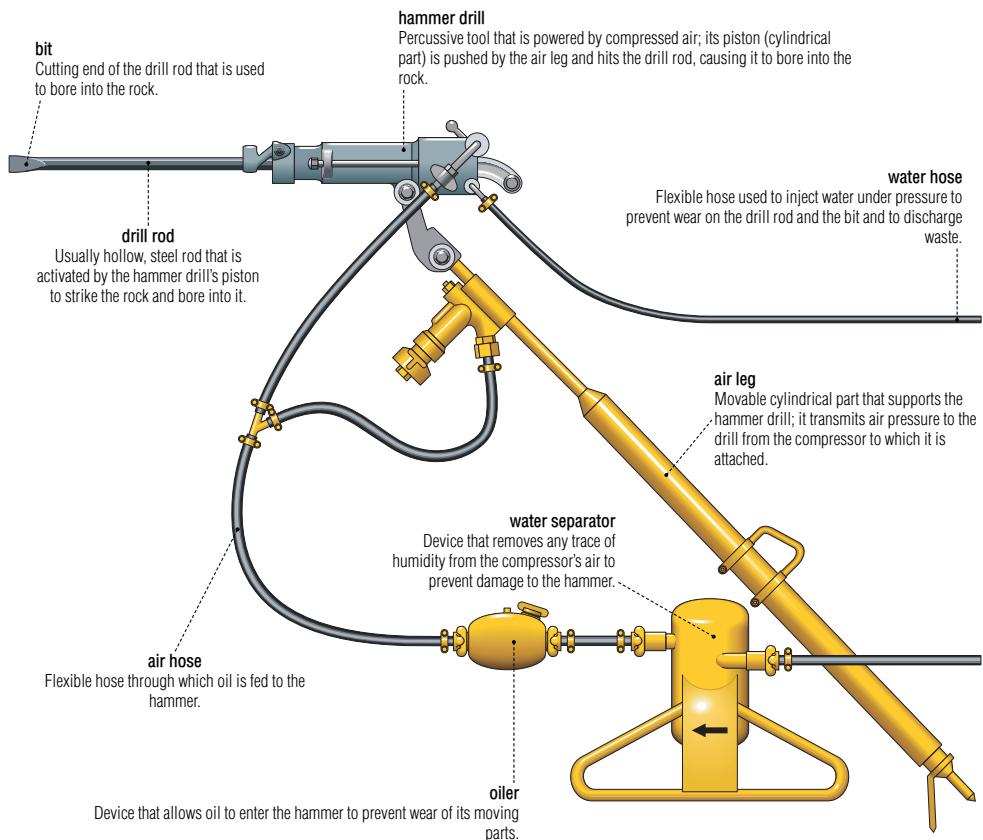
Excavation machine for pushing materials; it is made up of a crawler tractor, a blade and often a ripper.

overburden

Part of the ground that covers the ore beds; it is removed to reach the deposit.

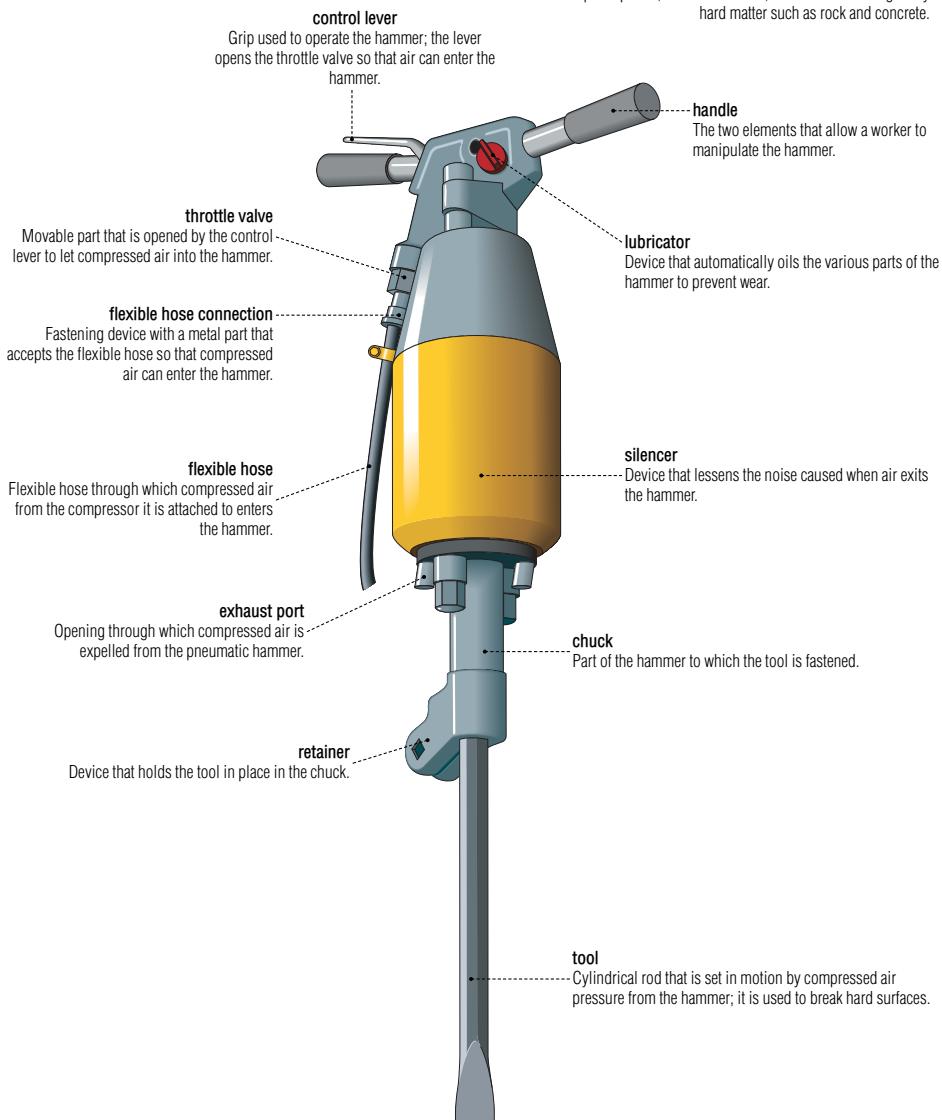
jackleg drill

Percussive tool that is powered by compressed air; it is used to bore holes into hard rock. The air leg makes the job easier for the drill operator.



pneumatic hammer

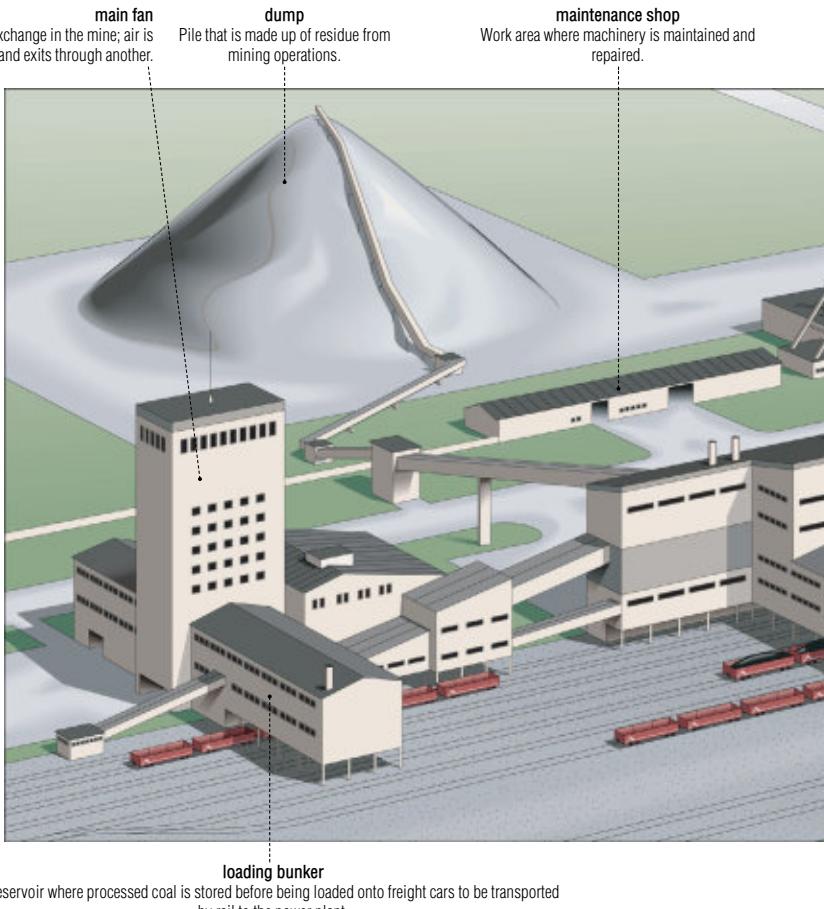
Percussive tool that is powered by compressed air; with the help of a piston, it activates a tool, which breaks through very hard matter such as rock and concrete.

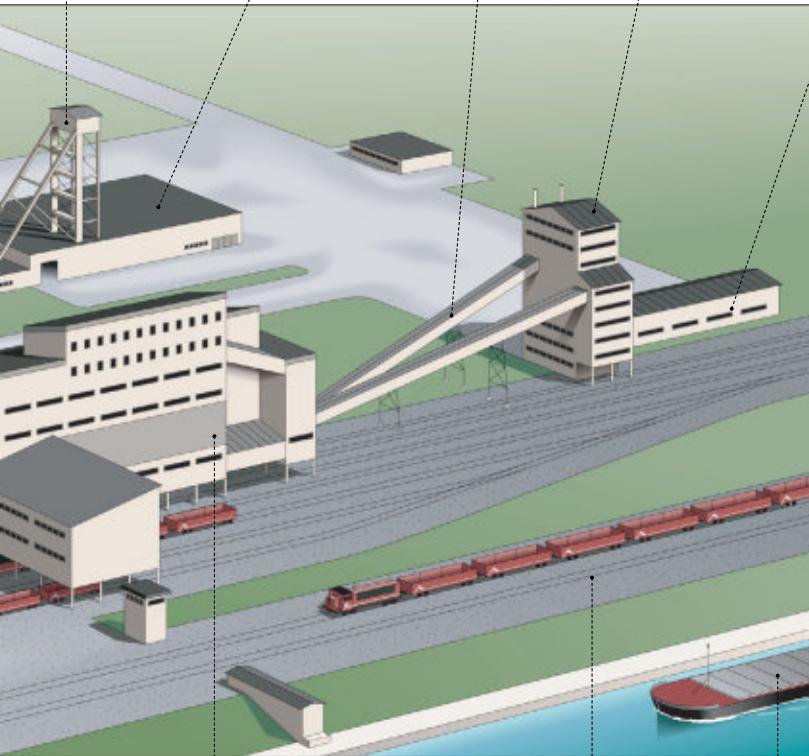


coal mine

pithead

The surface facilities needed for underground mining (including extraction machinery, storage areas and offices).



**headframe**

Opening at the top of the shaft that connects the aboveground facilities (including ventilation fans and hoists) to the underground areas being mined.

miners' changing-room

Area with sanitary facilities (showers, toilets) where miners can go mainly to change their clothes.

conveyor

Materials-handling device that consists of a conveyor belt (sturdy belt on rollers); it is used to carry coal to the treatment plant.

winding tower

Building that houses the shaft's hoisting equipment (including motors and hoisting cables); it provides communication between the surface and the mine galleries.

hoist room

Area that houses the hoist (cylinder) on which the hoisting cables are wound; it controls movement of the elevators and skip hoists in the shaft.

treatment plant

Place where all processing activities (including crushing and washing) are carried out to prepare the coal for market.

rail track

The tracks formed of two parallel rails on which trains travel to transport coal.

maritime transport

Means of transport that uses barges to transport coal over water.

coal mine

underground mine

Property in which excavations are carried out to extract deeply embedded (between 30 and 11,500 ft) coal for industrial mining.

headframe

Opening at the top of the shaft that connects the aboveground facilities (including ventilation fans and hoists) to the underground areas being mined.

vertical shaft

Shaft that is dug perpendicular to the surface; it serves various levels and is used mainly to transport personnel, equipment and ore.

elevator

Power lift fitted with a cab that transports coal or miners between the various levels.

pillar

Mass of ore that is left unmined at regular intervals in an excavation (chamber); it provides stability for the upper layers.

room

Cavity that remains after the ore is extracted; pillars support its roof.

chute

Vertical or inclined passageway through which ore, equipment, personnel and air move from one level of the mine to the other.

cross cut

Horizontal passageway that cuts through the ore bed perpendicularly; it provides communication between the passageways and helps to ventilate the mine.

manway

Passageway allowing workers to move around in the mine.

drift

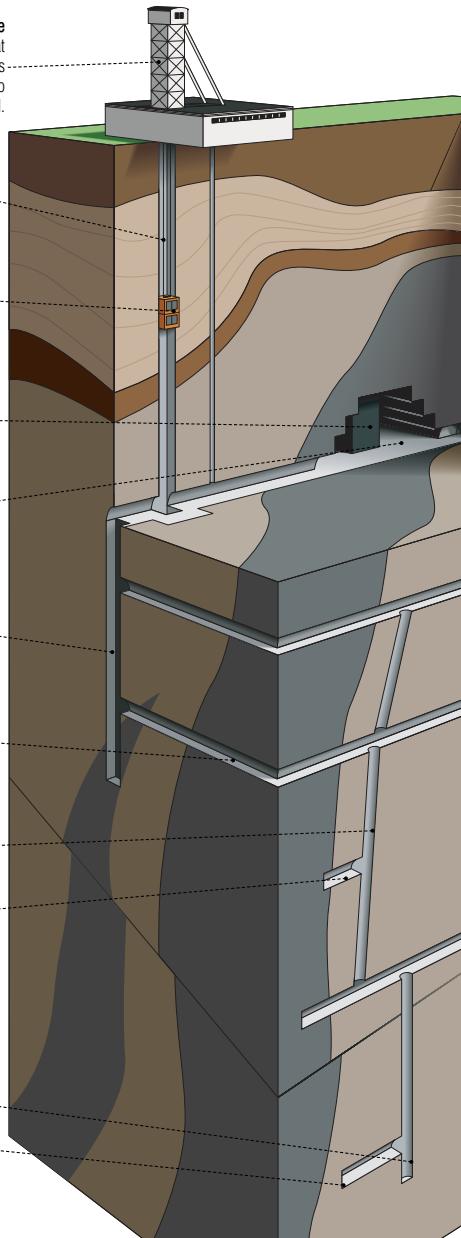
Passageway dug horizontally along the grade line of the ore seam; it can also be dug into the ore vertically.

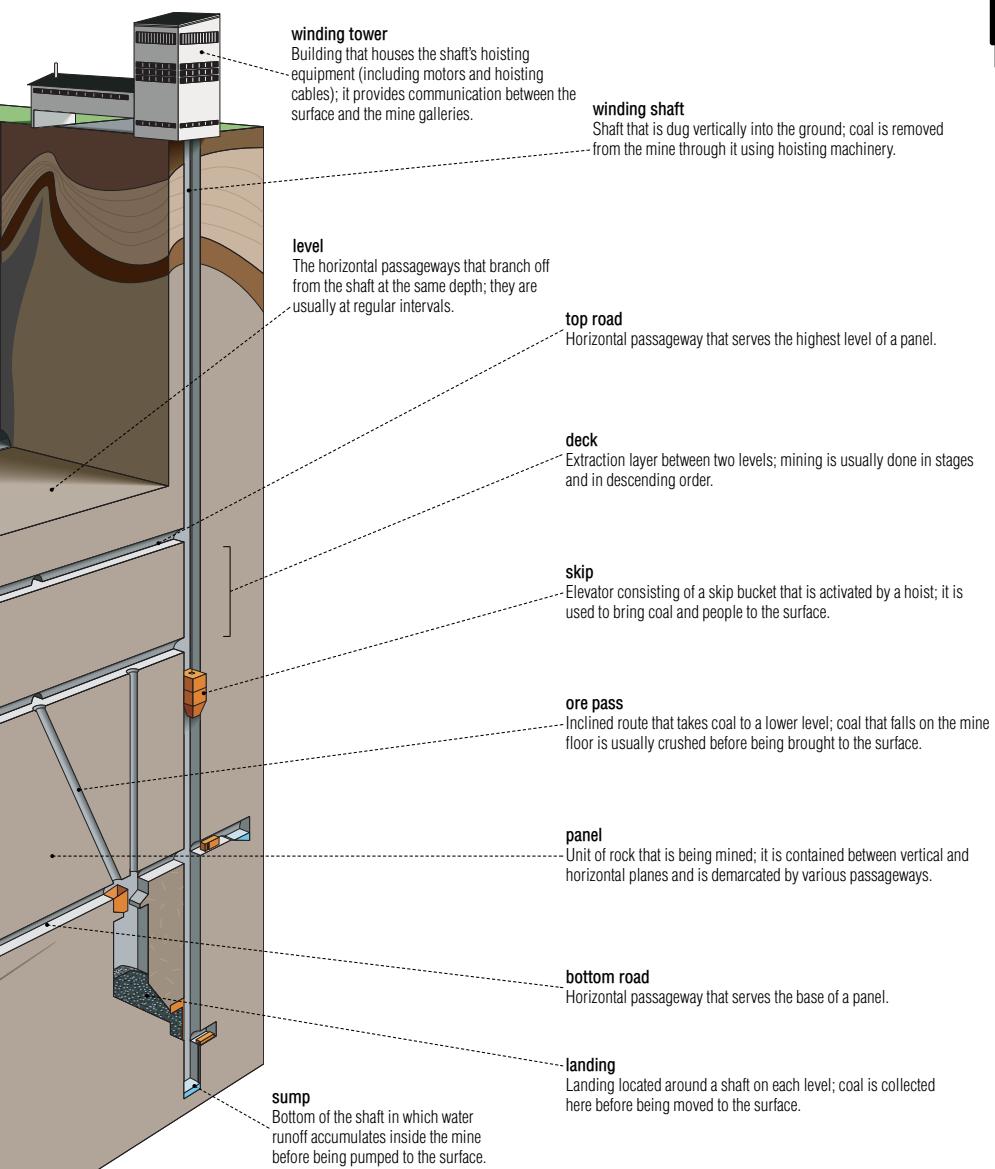
winze

Vertical or inclined passageway that connects two levels; it is dug downward from inside the mine and not from the surface.

face

Opening that is dug laterally into the rock as coal is extracted.



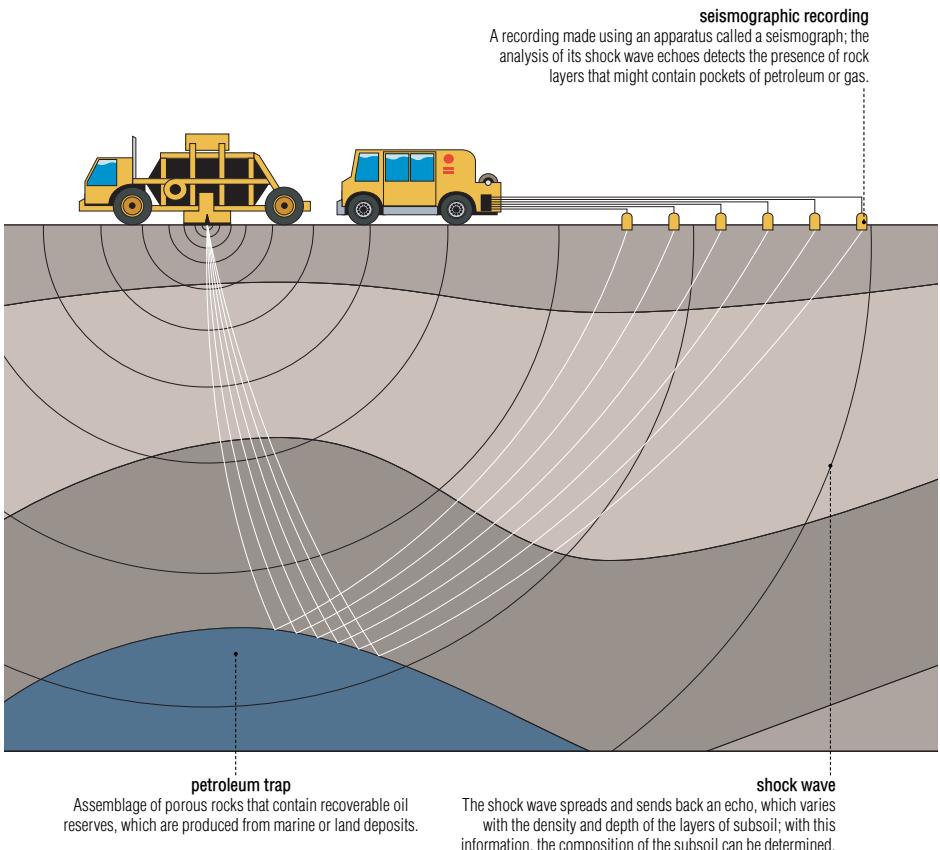


oil

Flammable, relatively viscous oily liquid that is used as an energy source; it is made up of various hydrocarbons resulting from the decomposition of plant life over millions of years.

surface prospecting

Searching for potential oil deposits by studying the structure of the subsoil using a seismograph.

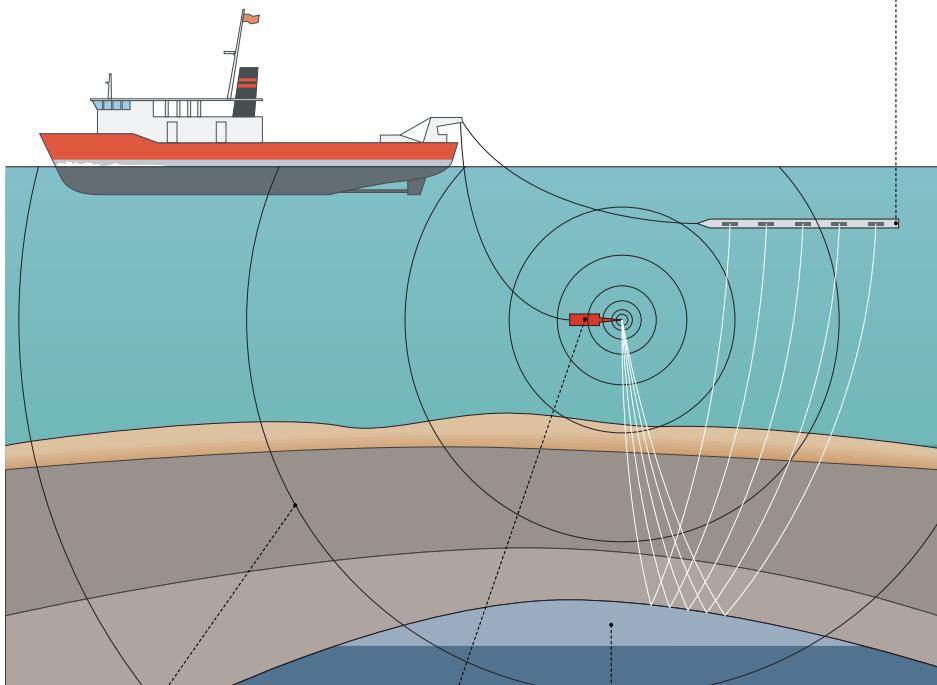


offshore prospecting

Vibrations from an exploding charge in the sea are used to locate oil deposits; prospecting offshore is more difficult than on land.

seismographic recording

A recording made using an apparatus called a seismograph; the analysis of its shock wave echoes detects the presence of rock layers that might contain pockets of petroleum or gas.

**shock wave**

The shock wave spreads and sends back an echo, which varies with the density and depth of the layers of subsoil; with this information, the composition of the subsoil can be determined.

blasting charge

Quantity of explosives (substances capable of discharging high-temperature gases over a very short time period) that produce shock waves when detonated.

petroleum trap

Assemblage of porous rocks that contain recoverable oil reserves, which are produced from marine or land deposits.

oil

drilling rig

All the drilling machinery and devices that are used to excavate and extract oil from the ground.

Metal structure erected over an oil well; tools for drilling through rock are raised and lowered through it.

mud injection hose

Flexible hose that introduces the drilling mud into the swivel.

swivel

Piece attached to the lifting hook and the kelly; it is used to introduce mud into the drill pipe to cool and lubricate the bit.

drilling drawworks

Device that consists of a cylinder on which hoisting cables are wound; it is used to lower the drill pipes and bit into the well and to lift them out.

substructure

Metal infrastructure that supports the derrick, engines and auxiliary equipment.

vibrating mudscreen

Perforated vibrating tray that is used to filter mud as it exits the well to remove debris and recycle the mud.

drill pipe

Hollow steel rods that are joined together according to the depth of the excavation; their rotation activates the bit.

drill collar

Heavy steel tube immediately above the bit that applies a certain weight to the bit to help it cut into the rock.

bit

Rotating drill bit with toothed steel or diamond wheels; it bores into rock to break it up and drill a hole.

oil

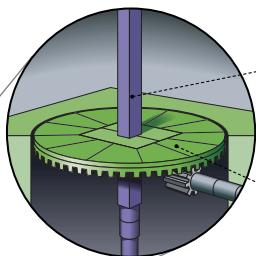
Flammable, relatively viscous oily liquid that is used as an energy source; it is made up of various hydrocarbons resulting from the decomposition of plant life over millions of years.

gas

Mixture of gaseous hydrocarbons (mainly methane) that are found in underground deposits, which sometimes also contain crude oil; it is used mainly as a fuel.

rotary system

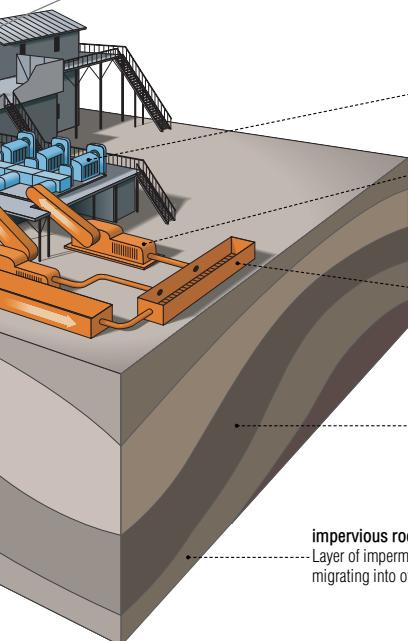
Drilling device in which a kelly is attached to a rotary table; with the help of powerful motors, it transmits the rotative movement to the kellys.

**kelly**

Special square rod that is screwed to the top of the drill pipes and driven by the rotary table.

rotary table

Circular table that is moved by powerful motors; it transmits its rotative movement to the drill pipes by means of the kelly.

**engine**

Device converting the combustion of fuel and air into mechanical energy.

mud pump

Device that circulates the mud in the drilling rig.

mud pit

Basin that contains mud (a mixture of water, clay and chemical products) used mainly to cool and lubricate the bit and to remove debris.

anticline

Geologic stratum that results from the convex folding of rock formations; large pools of oil often accumulate in it.

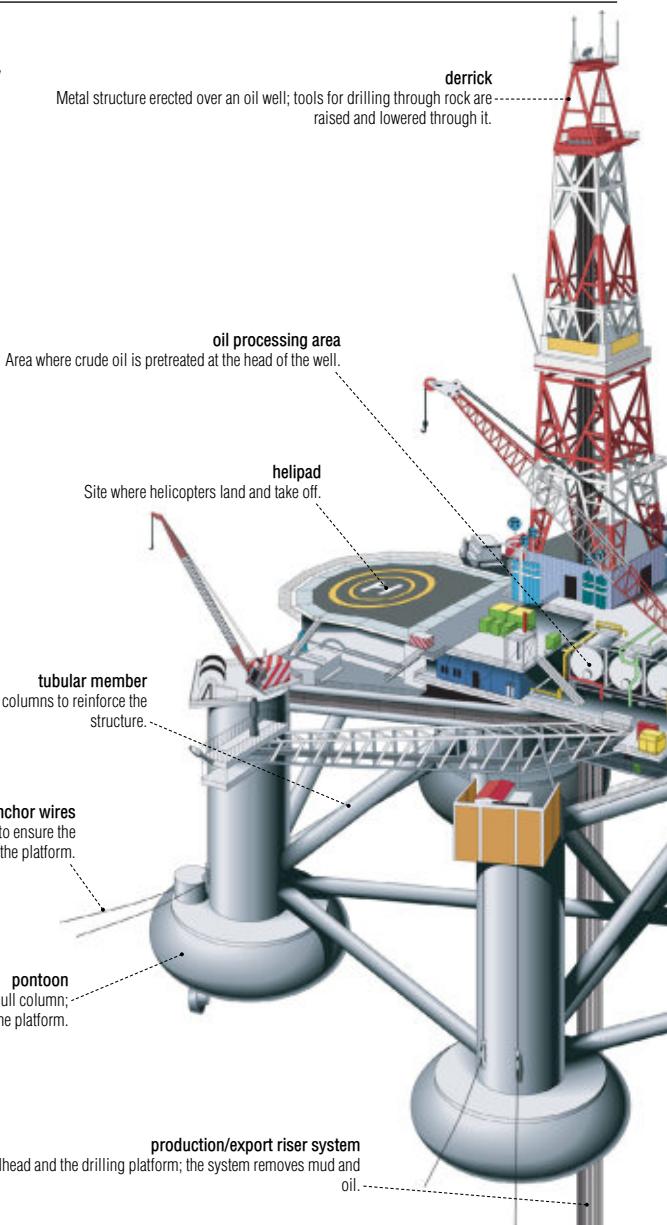
impervious rock

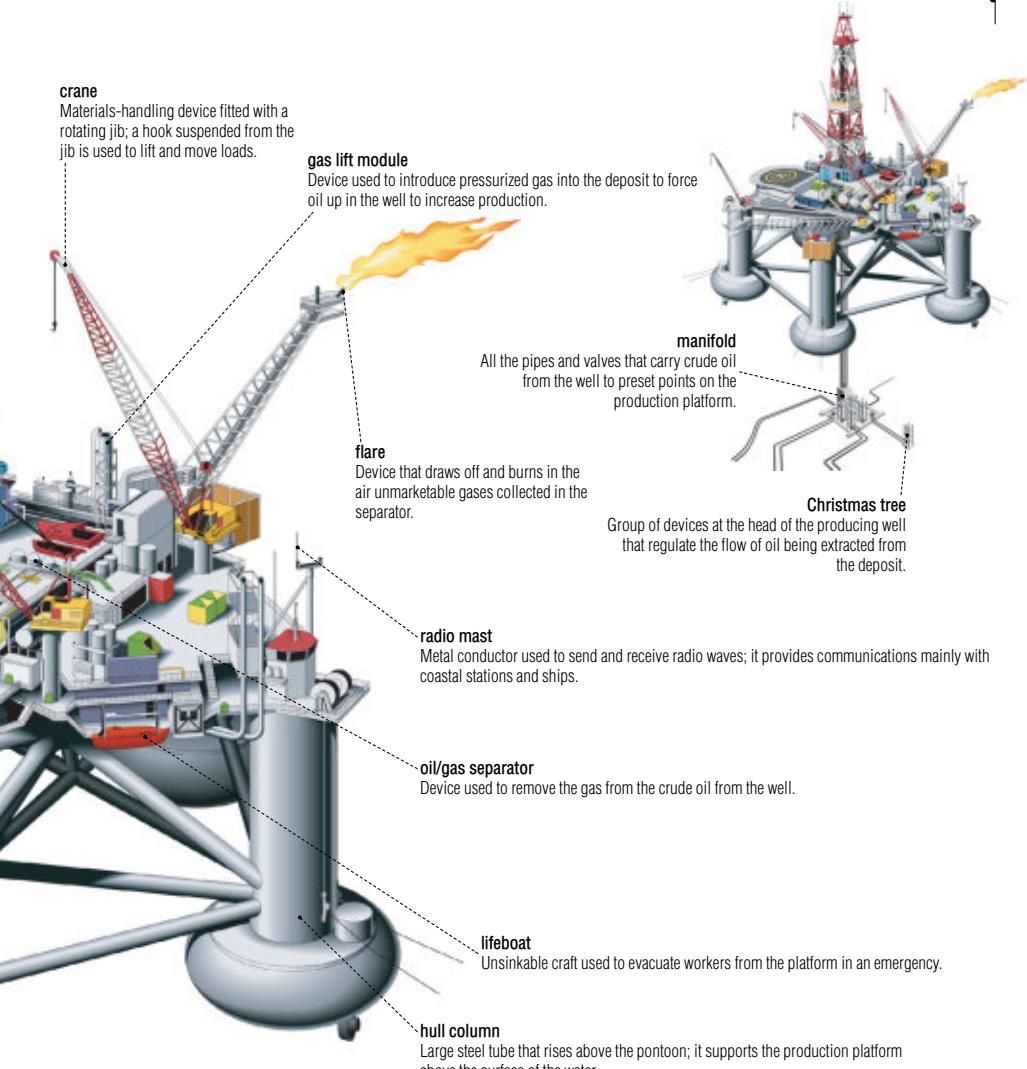
Layer of impermeable rock that covers and protects the oil deposit; it prevents hydrocarbons from migrating into other rocks.

oil

production platform

Facility used to extract underwater oil deposits; the separation and treatment of hydrocarbons are mainly done here.

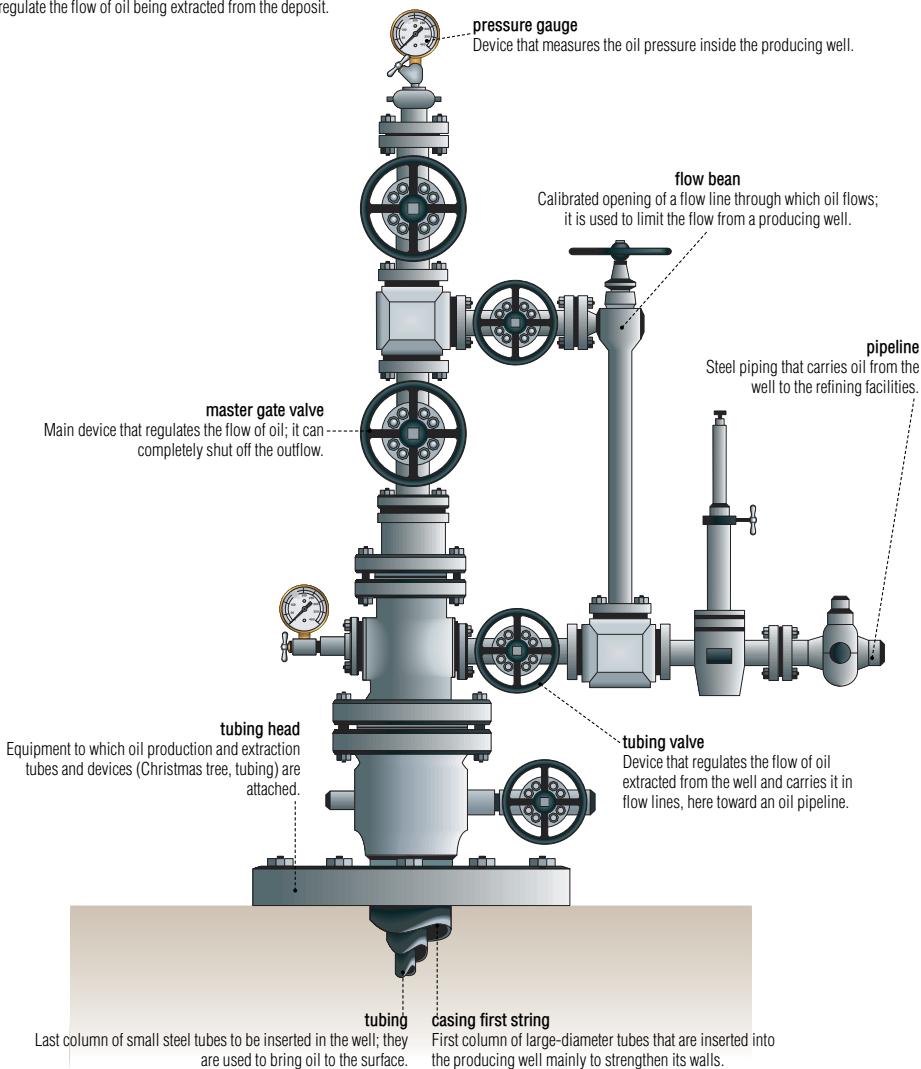




oil

Christmas tree

Group of devices at the head of the producing well that regulate the flow of oil being extracted from the deposit.



refinery products

Refining of crude oil yields hundreds of useful products.

**petrochemicals**

Chemical products derived from petroleum-based products; they are found in fertilizers, detergents, plastics and other products.

**jet fuel**

Aviation fuel used to power jet engines.

**gasoline**

Motor fuel that is used mainly by the automotive industry to power internal combustion engines.

**kerosene**

Fuel used for lighting and heating.

**stove oil**

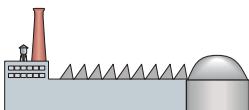
Fuel used mainly in home furnaces.

**diesel oil**

Fuel used mainly by the transportation industry to power diesel engines.

**heating oil**

Fuel used in home heating systems and industrial installations requiring little energy.

**bunker oil**

Fuel used in high-powered heating systems and electric power plants; it is also used to power large diesel engines.

**greases**

Pasty substances made of mineral oil and soap; they are used by industry to lubricate mechanical parts.

**lubricating oils**

Viscous substances that are used mainly to reduce friction between two moving surfaces.

**paraffins**

Water-insoluble substances that have various uses; these include candle making, packaging and pharmaceutical products.

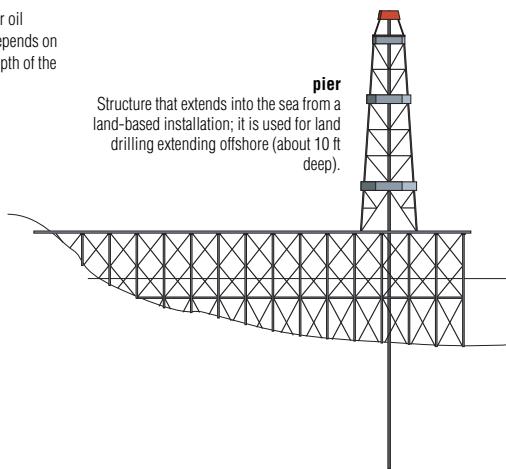
**asphalt**

Mixture of bitumen and other substances that is used mainly to pave roads.

oil

offshore drilling

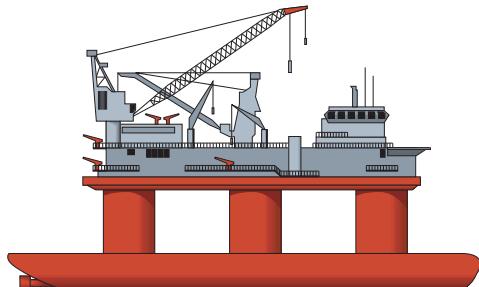
There are various types of underwater oil drilling installations; the one used depends on the location of the deposit and the depth of the water.



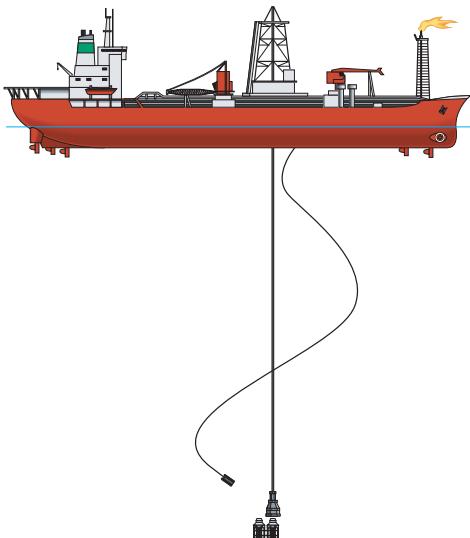
pier
Structure that extends into the sea from a land-based installation; it is used for land drilling extending offshore (about 10 ft deep).

emergency support vessel

Floating structure equipped with specialized equipment; it is used for rescue operations on drilling rigs.

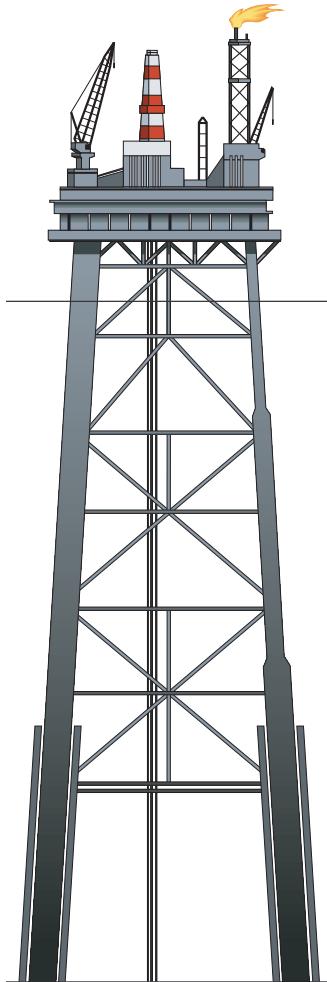
**drill ship**

Ship for drilling for oil in deep water (3,300 ft and more); it is more mobile but less stable than a semisubmersible or jack-up platform.

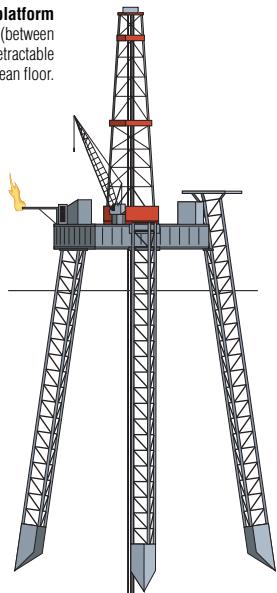


fixed platform

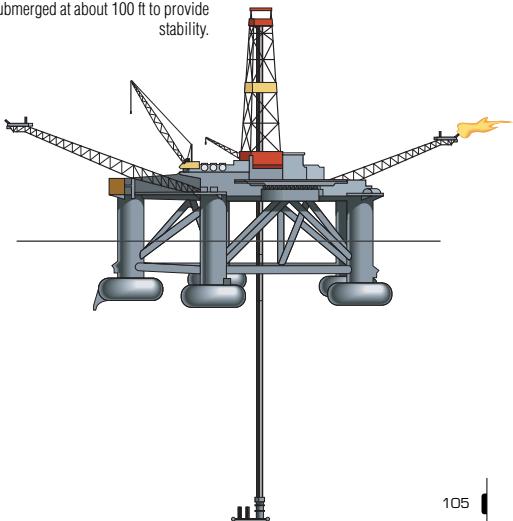
Structure that is mainly used at moderate depths (up to 1,300 ft); it rests on the seabed on pillars buried deep in the sea floor.

**jack-up platform**

Movable structure that is used in shallow water (between 65 and 330 ft); it is raised above sea level on retractable pillars resting on the ocean floor.

**semisubmersible platform**

Movable structure that is anchored to the seabed and used at depths of 350 to 1,650 ft; it is mounted on pontoons submerged at about 100 ft to provide stability.



oil

crude-oil pipeline

Continuous underground, aboveground or underwater oil pipeline that can be thousands of miles long (the Trans-Siberian pipeline is 3,800 mi long).

derrick

Metal structure erected over an oil well; tools for drilling through rock are raised and lowered through it.

offshore well

Hole dug in the sea floor to extract oil deposits; equipment such as the Christmas tree rests on the seabed.

Christmas tree

Group of devices at the head of the producing well that regulate the flow of oil being extracted from the deposit.

buffer tank

Large container that stores crude oil temporarily before it is pumped back into the pipeline.

central pumping station

Powerful pumping station that maintains the pressure required to move the oil along the pipeline to the next pumping station.

aboveground pipeline

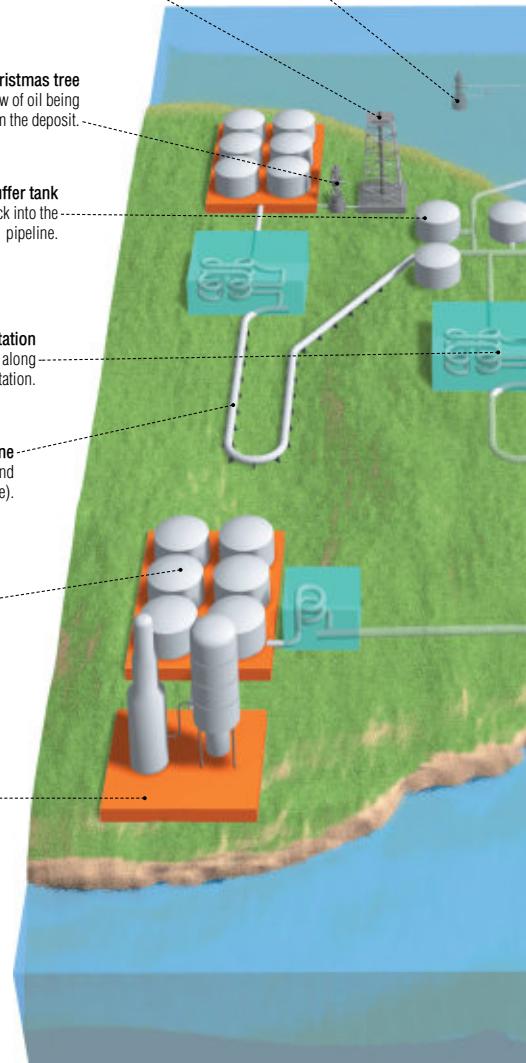
Oil pipeline that rests on aboveground supports to protect it from frozen ground (e.g., the Alaska pipeline).

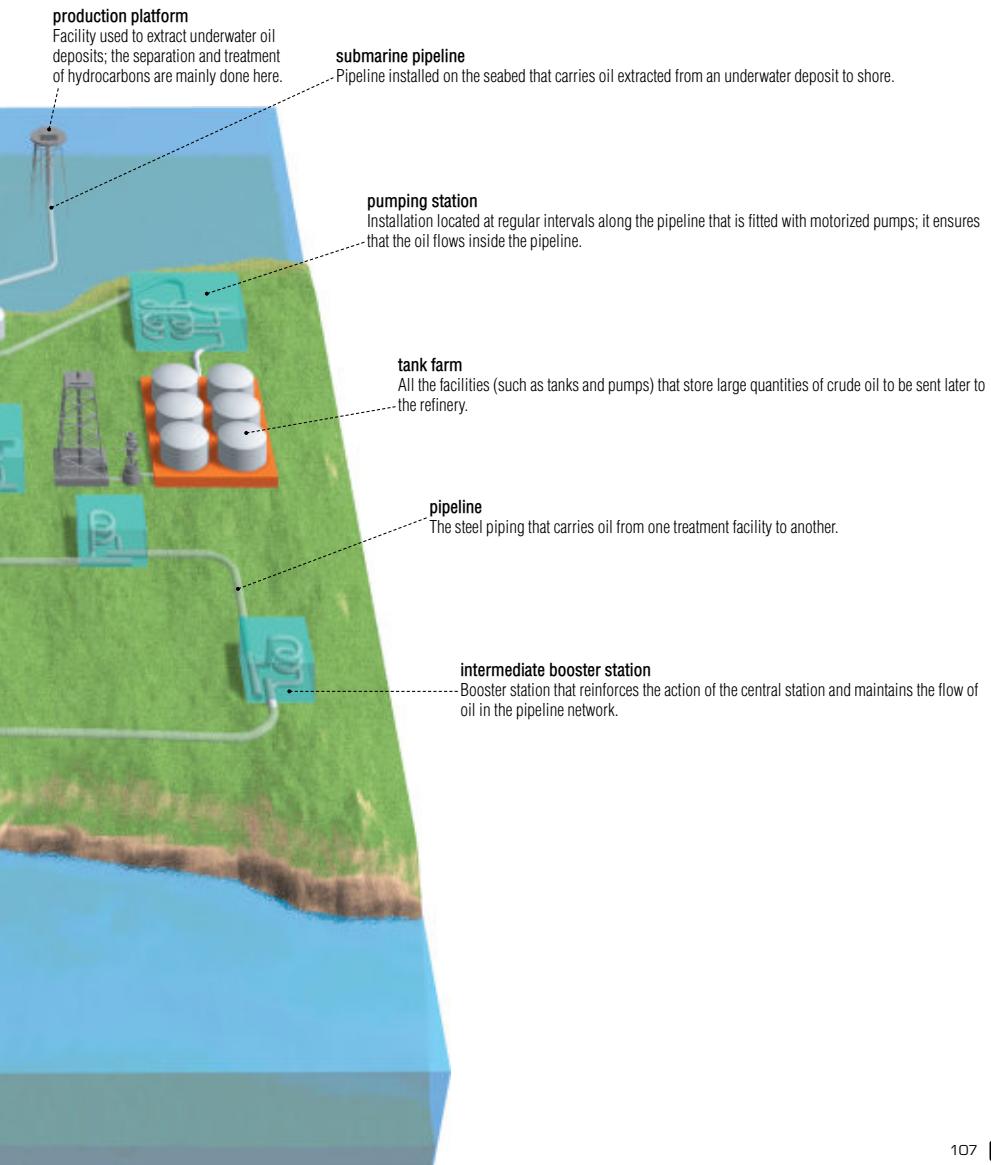
terminal

Facility located at the end of the pipeline that includes equipment such as tanks and pumps; it receives the crude oil before it is refined.

refinery

Plant in which crude oil is refined (separated and scrubbed) to obtain a broad range of finished products (including motor fuel and oils).





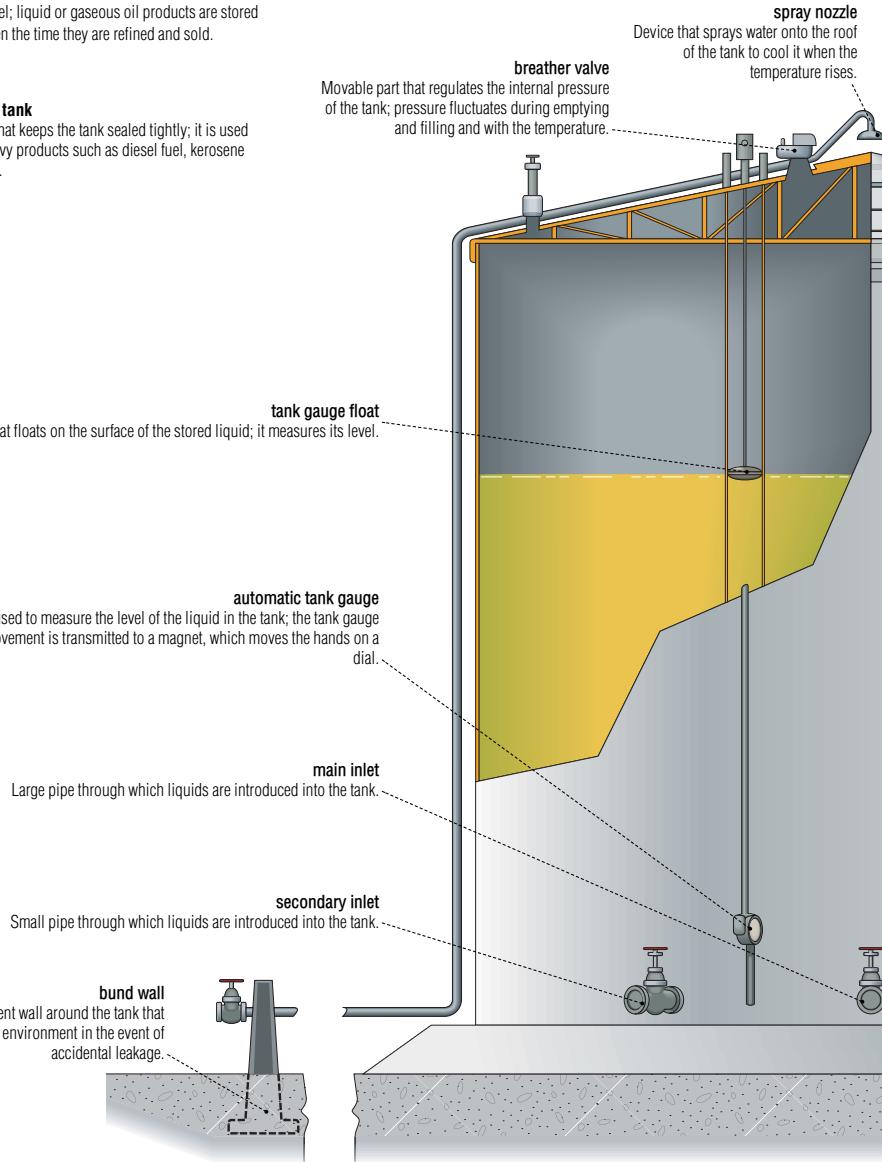
oil

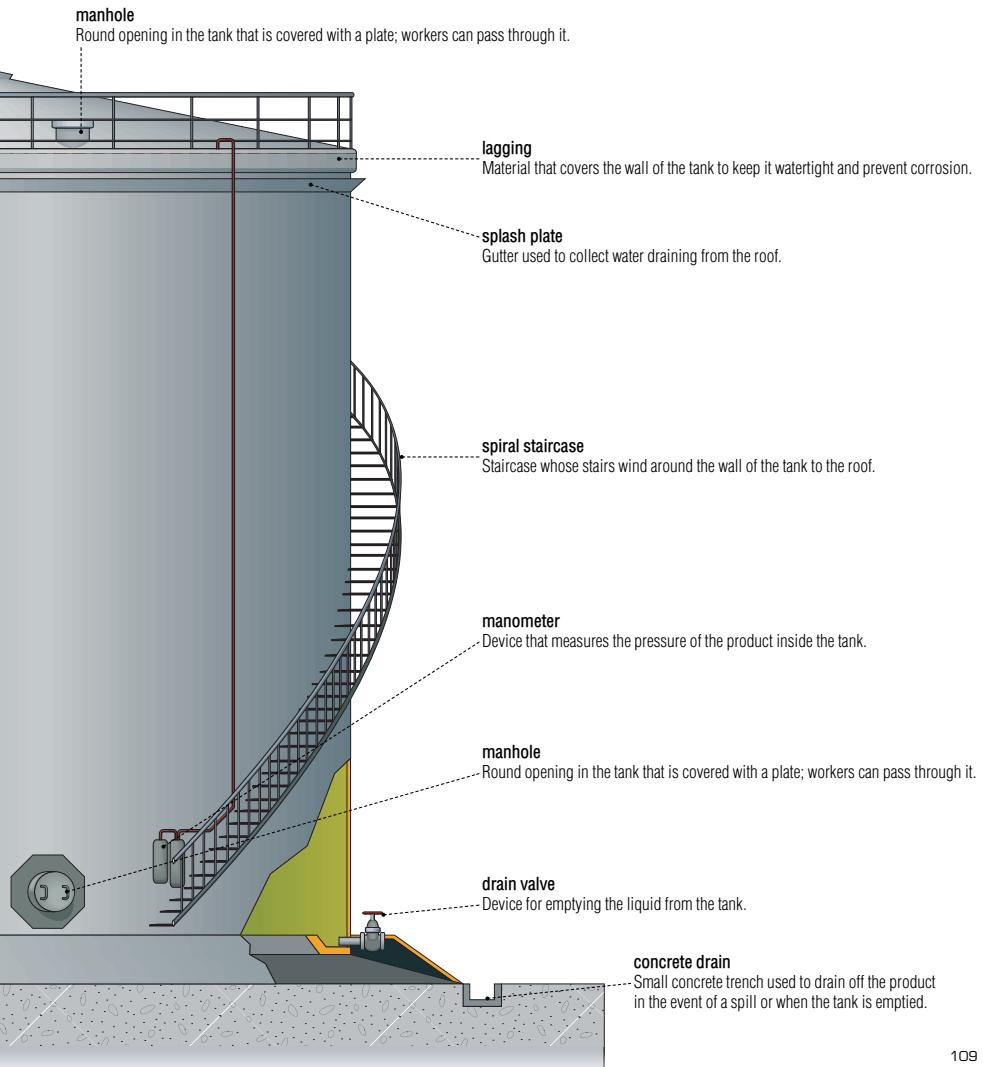
tanks

Large covered cylindrical containers that are usually made of steel; liquid or gaseous oil products are stored here between the time they are refined and sold.

fixed-roof tank

Fixed roof that keeps the tank sealed tightly; it is used to store heavy products such as diesel fuel, kerosene and asphalt.

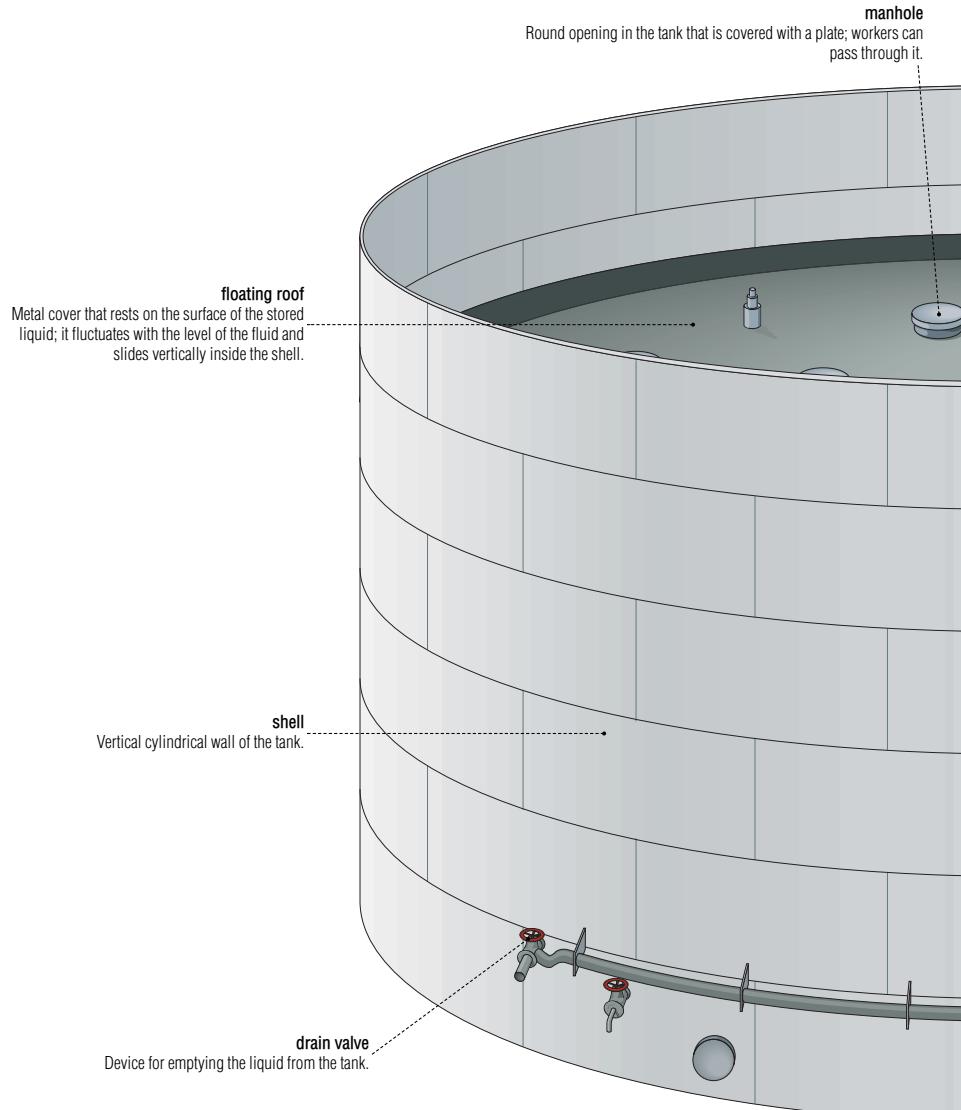


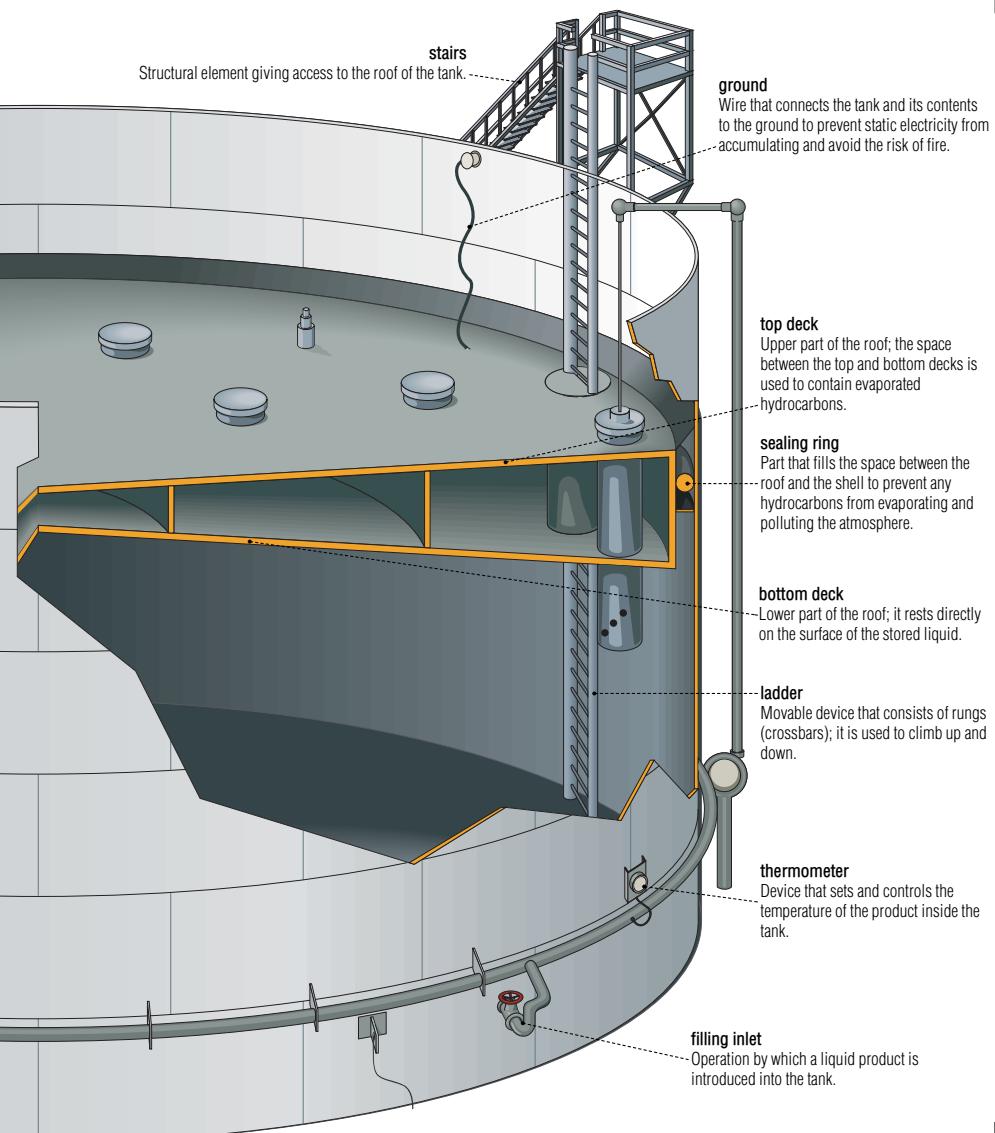


oil

floating-roof tank

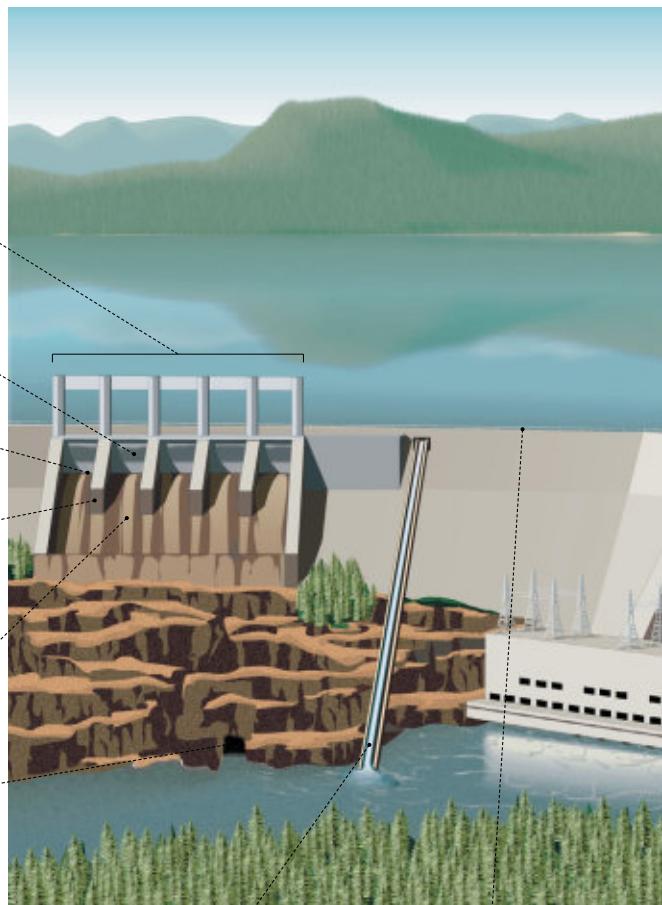
Tank whose floating roof rests directly on the surface of the liquid to minimize the evaporation of hydrocarbons; it is used to store the most volatile products.





hydroelectric complex

The reservoir structures and installations that use water power to produce electricity.



spillway

Channel that discharges excess water from the reservoir during flooding to avoid submerging the dam.

spillway gate

Movable vertical panel; it is opened to allow the reservoir's overflow to pass through.

crest of spillway

Cement crest over which the reservoir's overflow discharges when the spillway gates are opened.

training wall

Wall that separates the spillway chutes; it is used to direct the water flow.

spillway chute

Inclined surface along which discharged water flows out.

diversion tunnel

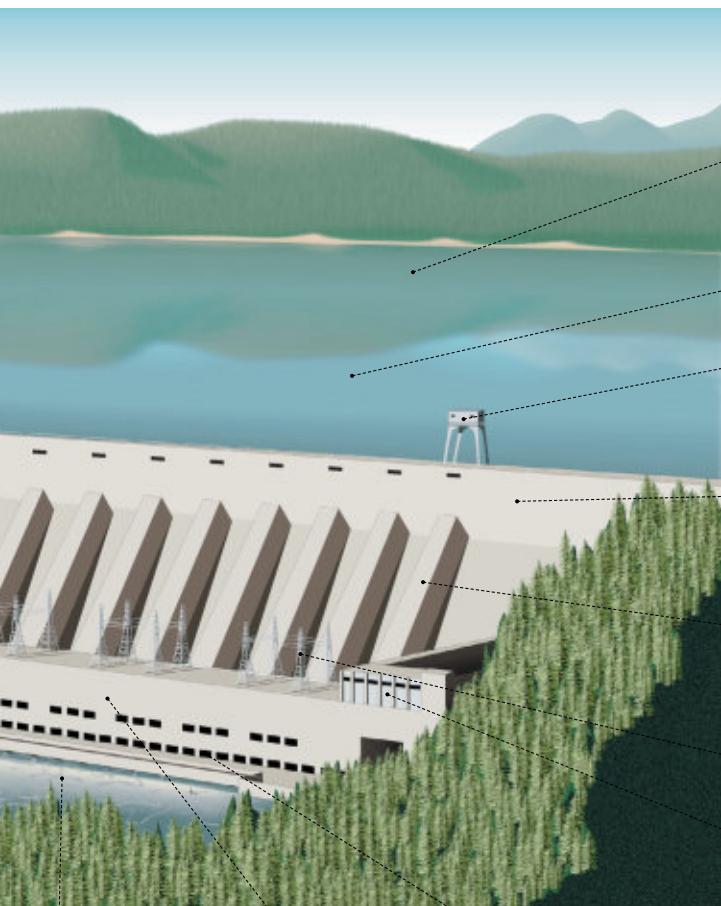
Underground conduit that diverts water during construction.

log chute

Structure that allows floating wood to travel from upstream to downstream of the dam.

top of dam

Upper part of the dam; it rises above the water level of the reservoir by several yards.

**afterbay**

Area of the watercourse where water is discharged after passing through the turbines.

power plant

Plant that uses an energy source, here water, and converts it into electricity.

machine hall

Area that houses the generator units used to produce electricity.

reservoir

Basin formed by the construction of a dam; it holds back a very large volume of water so that the flow rate can be controlled.

headbay

Part of the reservoir immediately in front of the dam where the current originates.

gantry crane

Hoisting device in the form of a bridge; it moves along rails.

dam

Barrier built across a watercourse in order to build up a supply of water for use as an energy source.

penstock

Channel that carries water under pressure to the power plant's turbines.

bushing

Device that allows the conductor to pass through the wall of the transformer and separates it from the latter.

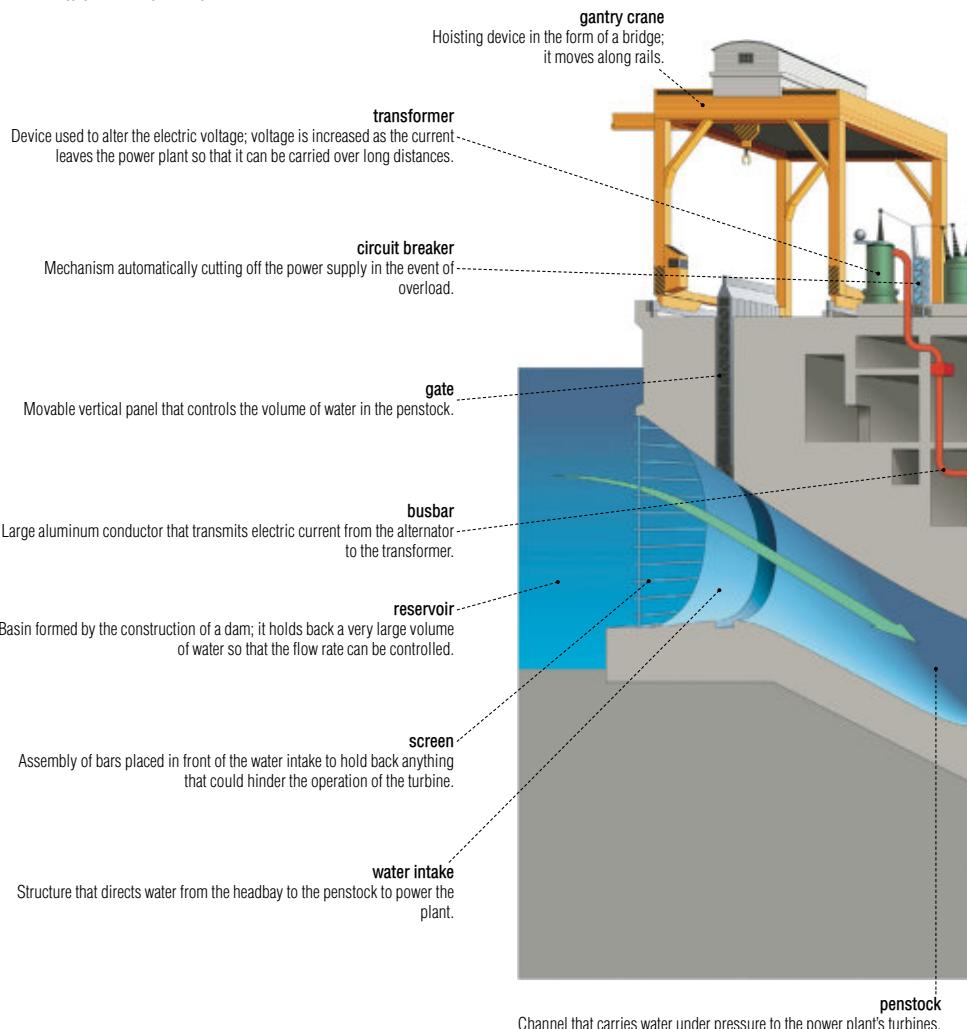
control room

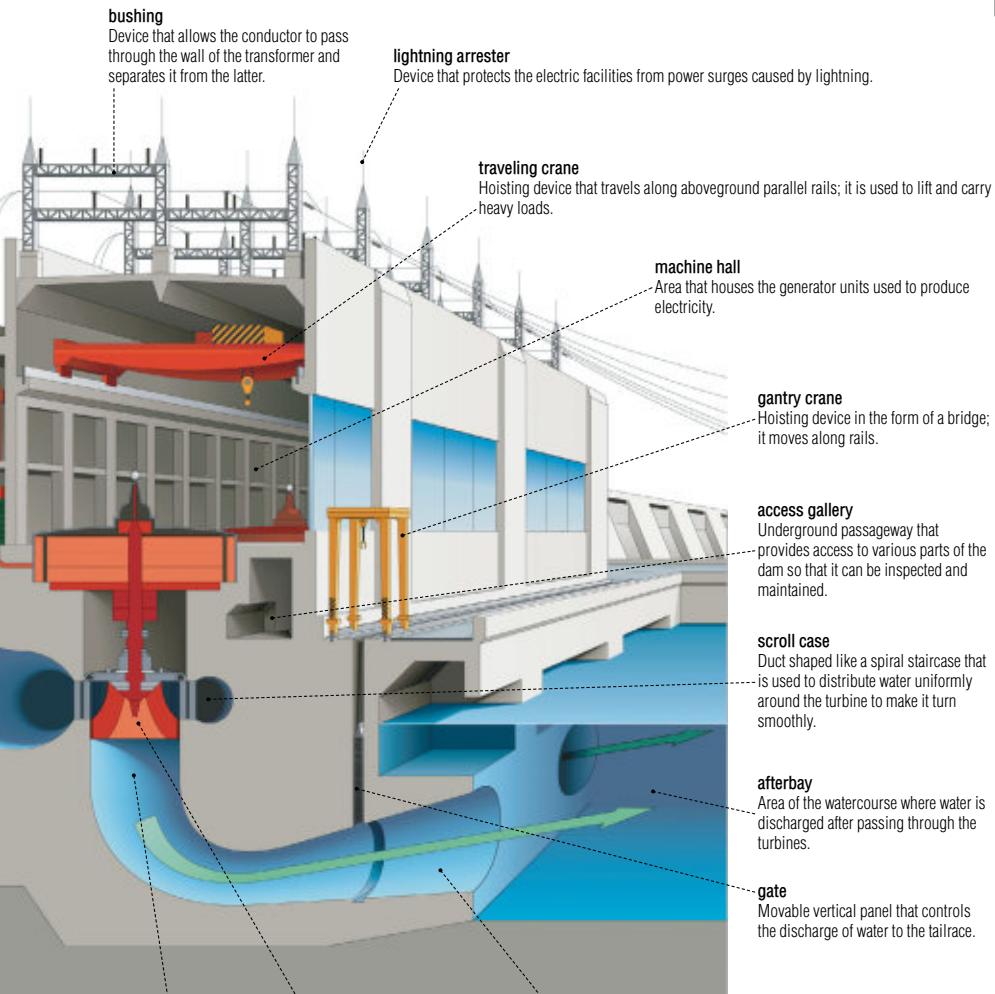
Area that contains the various control and monitoring devices required for the production of electricity.

hydroelectric complex

cross section of a hydroelectric power plant

Hydroelectric power plant: plant that produces electricity from energy generated by flowing water.





draft tube
Conduit at the base of the turbine that increases the runner's output by reducing the pressure of the water as it exits.

generator unit
Device with a turbine that transmits the water's mechanical energy to the generator's rotor to make it turn to produce electricity.

tailrace
Channel that discharges water toward the afterbay in order to return it to the watercourse.

generator unit

Device with a turbine that transmits the water's mechanical energy to the generator's rotor to make it turn to produce electricity.

Movable part of the generator that is made up of electromagnets; its rotation induces an electric current in the stator.

rotor

Unit that bears the thrust of the turbine and the weight of the rotating parts of the generator unit.

thrust bearing

Structure that covers the upper part of the turbine's runner.

turbine headcover

spiral case
Duct shaped like a spiral staircase that is used to distribute water uniformly around the turbine to make it turn smoothly.

spiral case

wicket gate
Movable panel that regulates the flow of water entering the turbine to ensure a constant rotational speed of the runner.

runner

Movable part of the turbine that transmits the movement of the water to the shaft to which it is attached to turn the rotor.

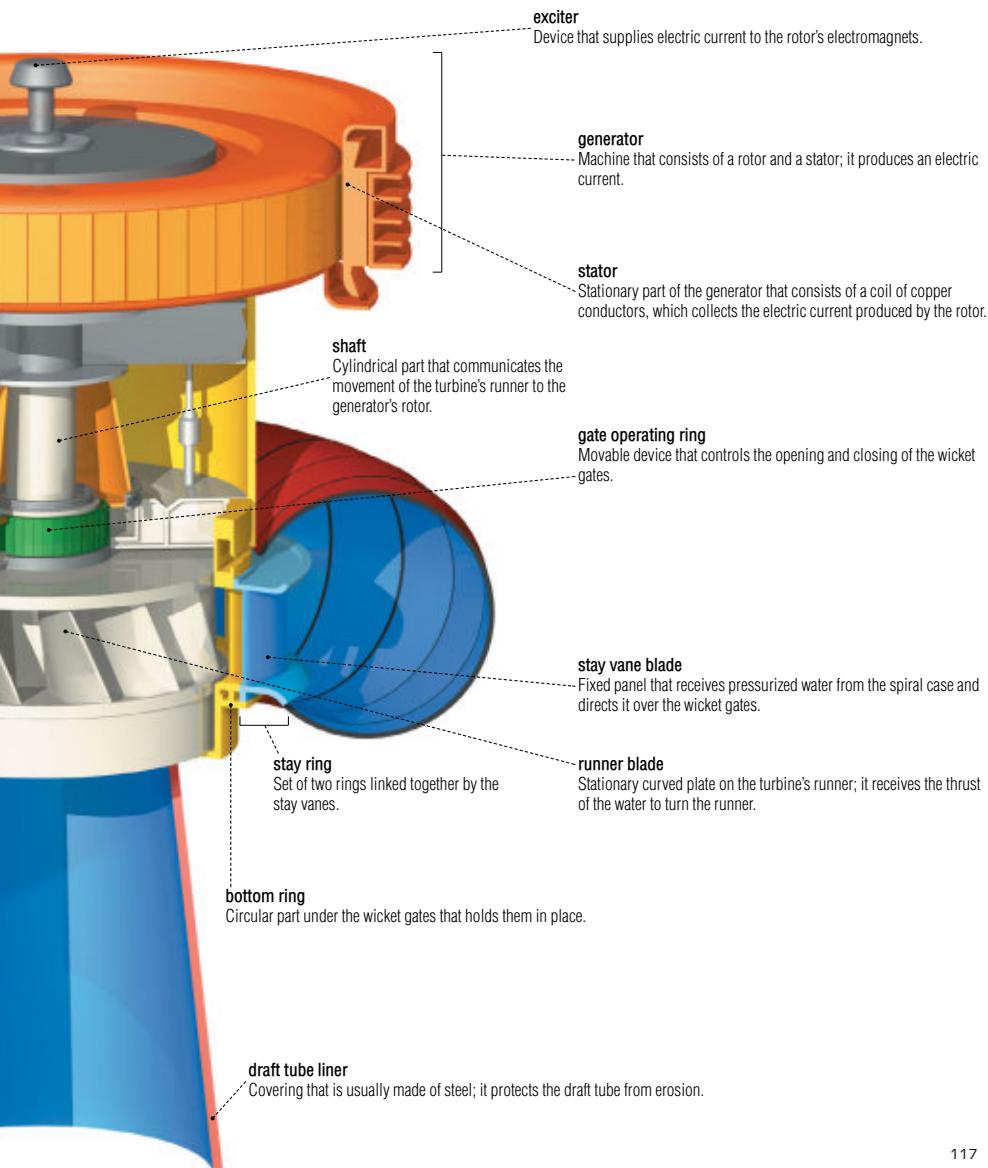


hydraulic turbine

Machine whose runner is powered by water; it transmits mechanical energy to the rotor to make it turn.

draft tube

Conduit at the base of the turbine that increases the runner's output by reducing the pressure of the water as it exits.



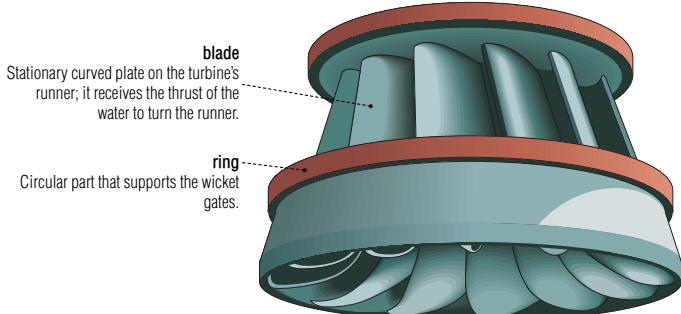
generator unit

runners

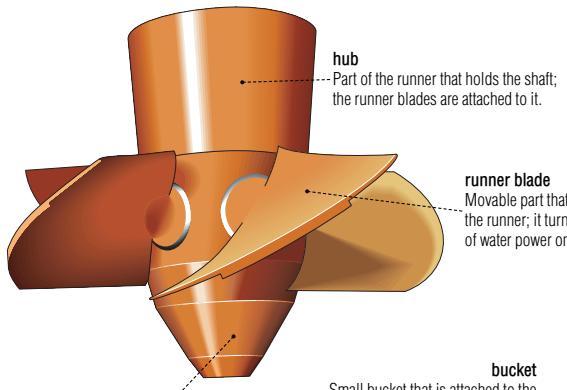
Movable parts of the turbine that transmit the movement of the water to the shaft to which they are attached to turn the rotor.

Francis runner

Most common type of runner that is suited to average heights of water (usually between 100 and 1,000 ft).

**Kaplan runner**

Type of runner that is suited to low heights of water (usually between 30 and 200 ft) and variable flow rates.

**Pelton runner**

Type of runner that is suited to high water sources (usually over 1,000 ft) and low flow rates.



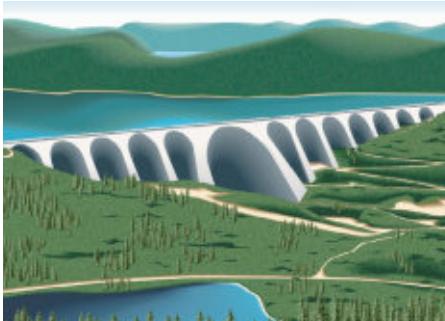
Disk housing all the turbine buckets that activates the runner.



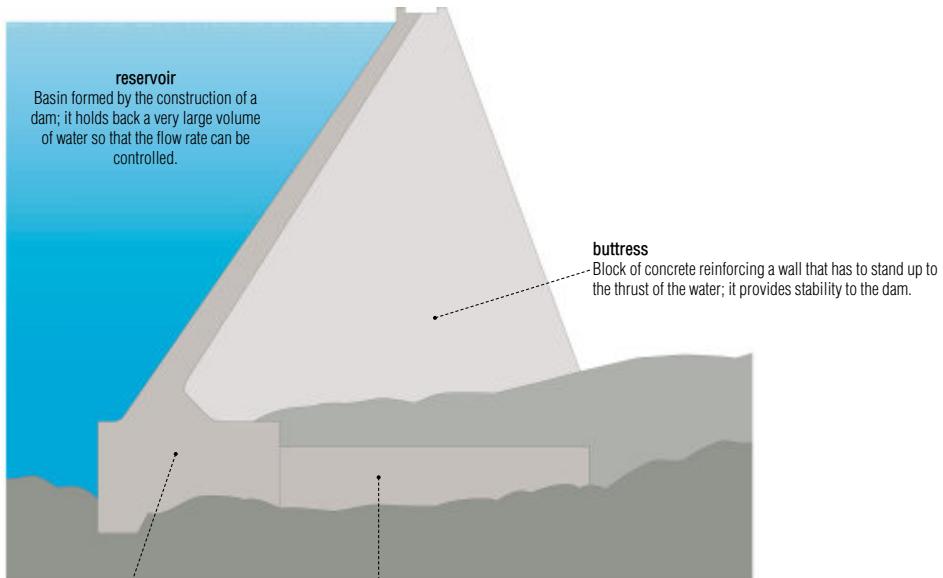
There are masonry dams, concrete dams and embankment dams; the choice depends on criteria such as the nature of the ground, the shape of the valley and the materials available.

buttress dam

Used mainly in wide valleys, it consists of an impermeable wall, which is shored up by a series of buttresses to transmit the thrust of the water to the foundation.



cross section of a buttress dam



foundation

Concrete structure that supports the weight of the dam and transmits it to the ground to provide stability to the dam.

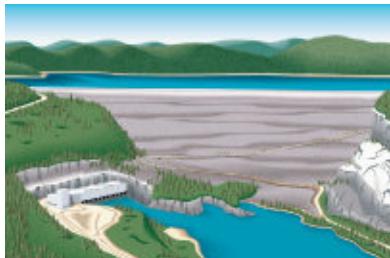
foundation blockage

Block of concrete that anchors the foundation in the ground to prevent movement.

examples of dams

embankment dam

Formed of mounds of earth or rocks, it is used mainly when the subsoil does not allow for construction of a concrete dam.

**cross section of an embankment dam****reservoir**

Basin formed by the construction of a dam; it holds back a very large volume of water so that the flow rate can be controlled.

pitching

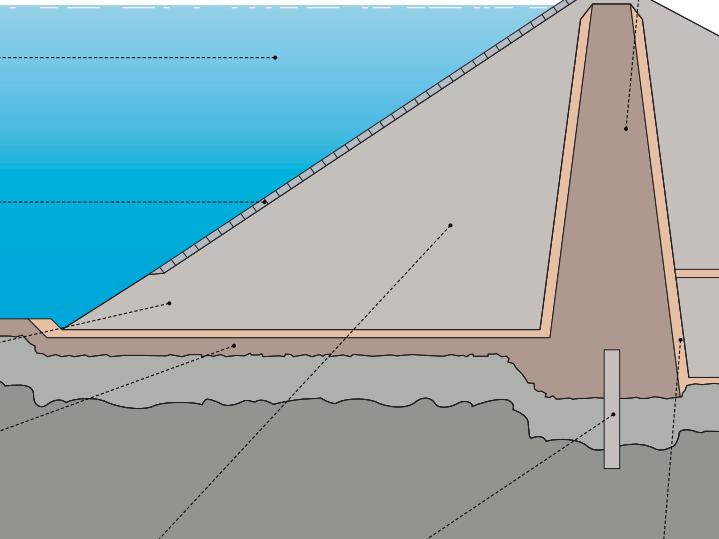
Layer of rock or concrete blocks that covers the upstream shoulder to prevent erosion.

upstream toe

Area where the upstream shoulder and the foundation of the dam meet.

upstream blanket

Impermeable layer that consists of compact clay; it rests on the bottom of the dam to prevent infiltration.



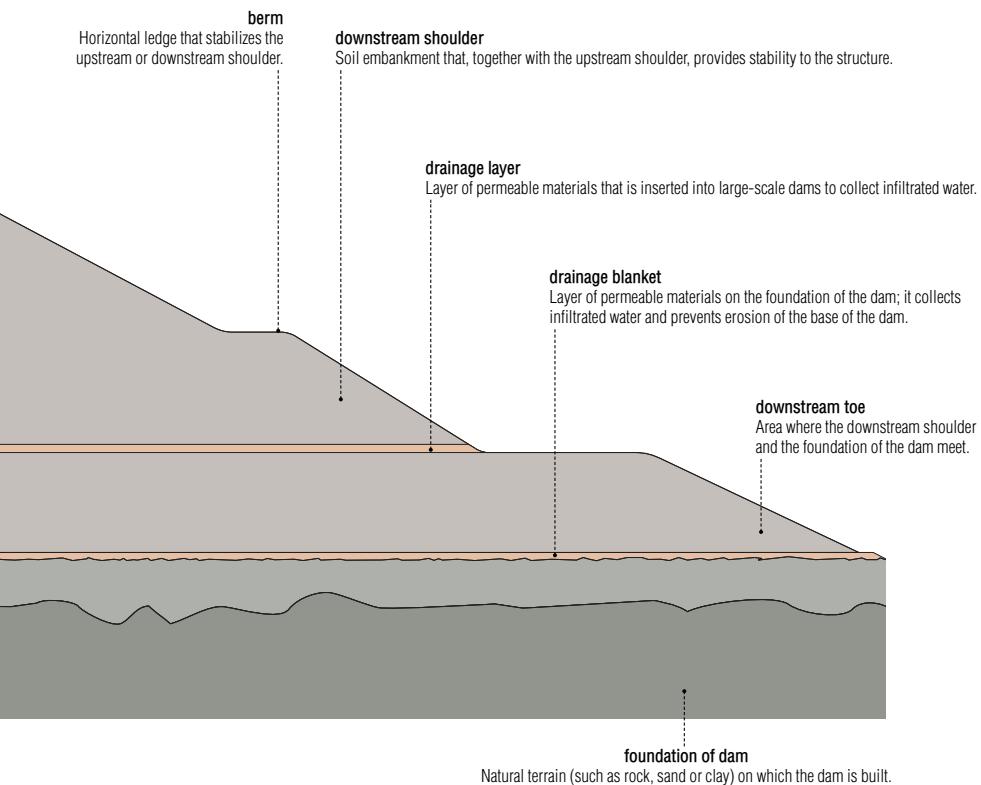
upstream shoulder
Soil embankment located on the reservoir side; its mass provides stability to the dam.

cut-off trench
Area of the foundation of the dam that is connected to the core; it contains impermeable materials to limit leakage and infiltration under the dam.

clay core
Central portion of the dam that is usually made of compact clay to make it watertight.

top of dam
Upper part of the dam; it rises above the water level of the reservoir by several yards.

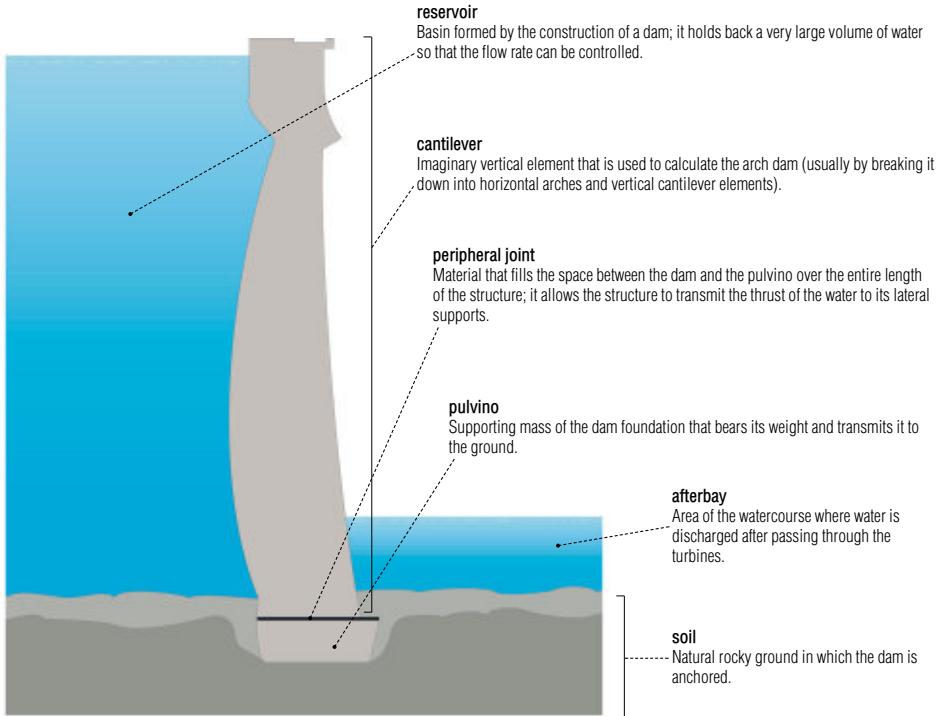
wave wall
Small wall located at the top of the upstream shoulder that protects the dam against waves.



examples of dams

arch dam

Its curvature allows most of the water's thrust to be transmitted to the usually narrow valley slopes supporting it.

**cross section of an arch dam**

**gravity dam**

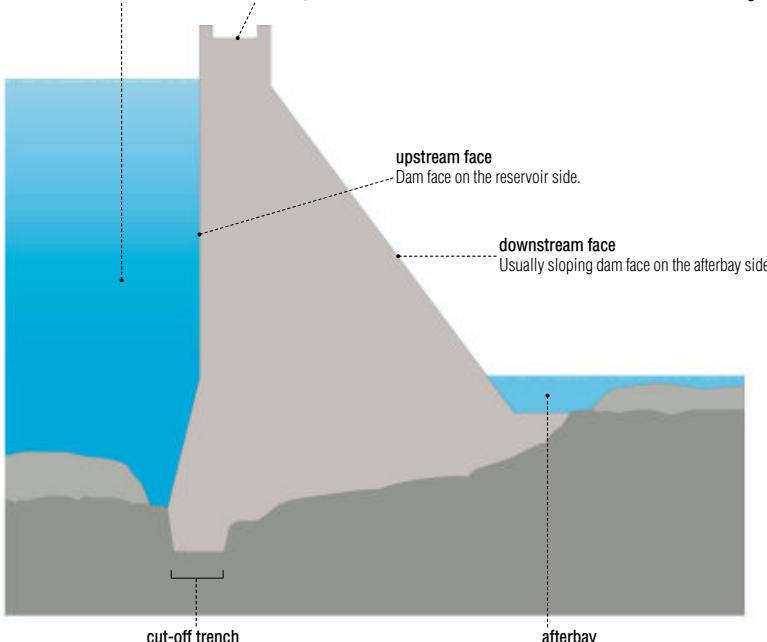
Its huge mass resists the thrust of the water to prevent it from overturning or sliding; this type of dam is usually used to hold back large volumes of water.

reservoir

Basin formed by the construction of a dam; it holds back a very large volume of water so that the flow rate can be controlled.

top of dam

Upper part of the dam that usually contains a roadway.

cross section of a gravity dam**cut-off trench**

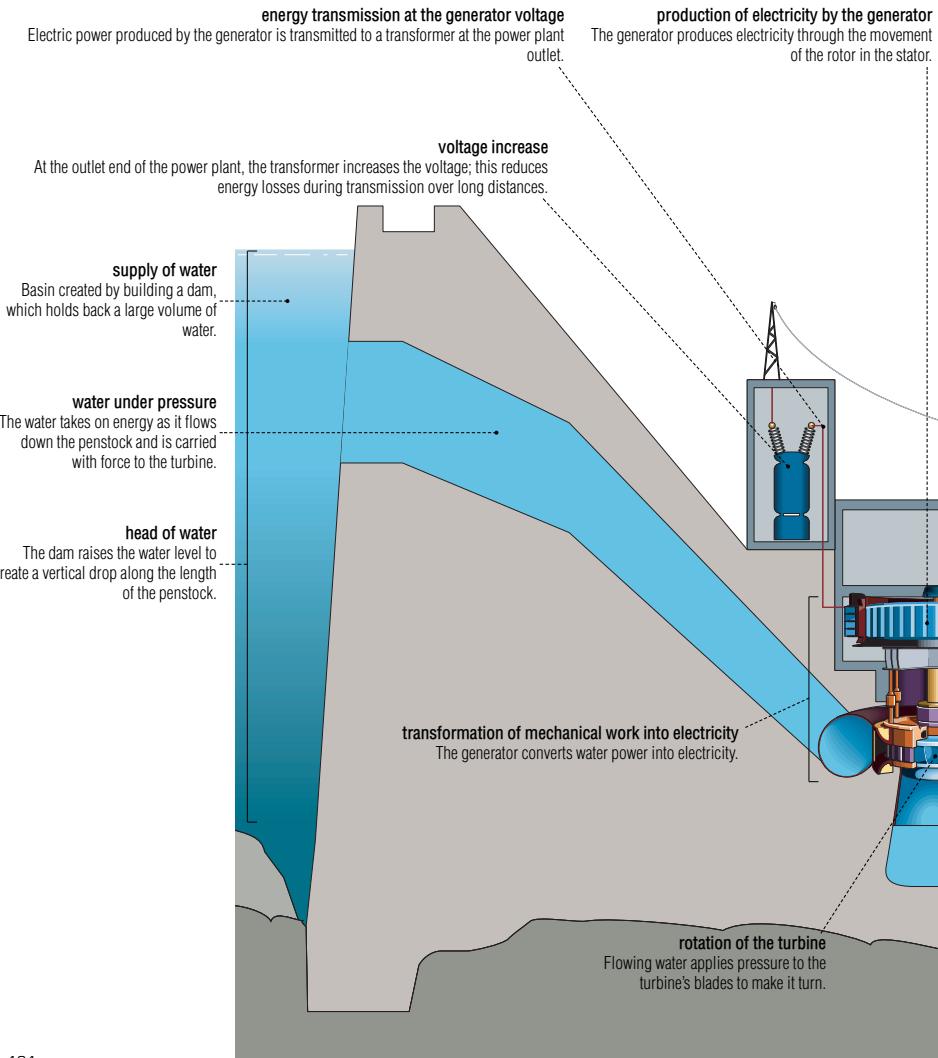
Watertight structure that extends the foundations of the dam into the ground; it limits leakage and infiltration under the dam.

afterbay

Area of the watercourse where water is discharged after passing through the turbines.

steps in production of electricity

In a hydroelectric power plant, water is turned into electricity, which is carried to consumers along a transportation and distribution network.



energy integration to the transmission network

The electricity produced is integrated into the network.

high-tension electricity transmission

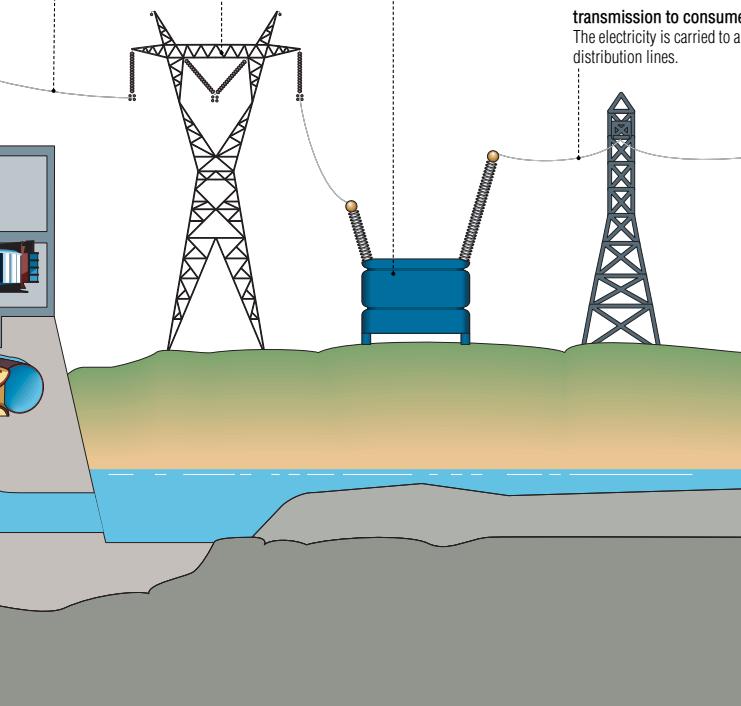
Using high-voltage lines to transmit electricity over long distances reduces the strength of the current and, as a result, energy losses.

voltage decrease

Before integrating the electricity into the distribution network, the voltage is progressively decreased to 240 V.

transmission to consumers

The electricity is carried to areas of consumption by low-voltage distribution lines.

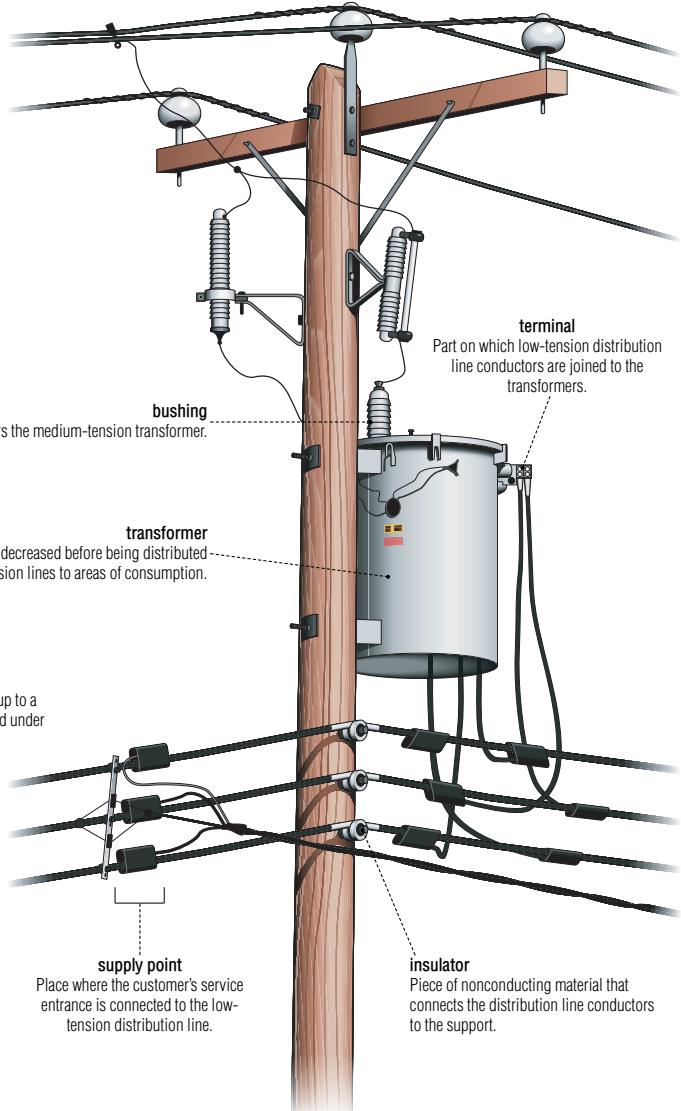


electricity transmission

Electricity is carried by overhead and underground lines; due to high cost, underground lines are used mainly in cities.

overhead connection

The equipment and overhead conductors that connect a subscriber's electric system to the public distribution network.



medium-tension distribution line

Overhead line that distributes electricity at a voltage between 750 and 50,000 V; its conductors are located at the top of electricity poles.

brace

Slanted part that connects the pole to the crossarm to hold it in place horizontally.

hot line connector

Linking piece with a bolt, which is tightened to bring together two conductors to establish an electric connection between them.

insulator

Piece of nonconducting material that connects the distribution line conductors to the support.

crossarm

Horizontal element located at the top of an electricity pole; insulators are attached to it.

lightning arrester

Device that protects the electric facilities from power surges caused by lightning.

fuse holder

Electric junction point where the fuse is attached and on which it articulates so the fuse can fall over.

fuse cutout

Unit that consists of a fuse and a fuse holder.

fuse

Protection device for the electric circuit; it falls from the fuse holder to cut the current in the event of a surge.

electricity transmission

pylon

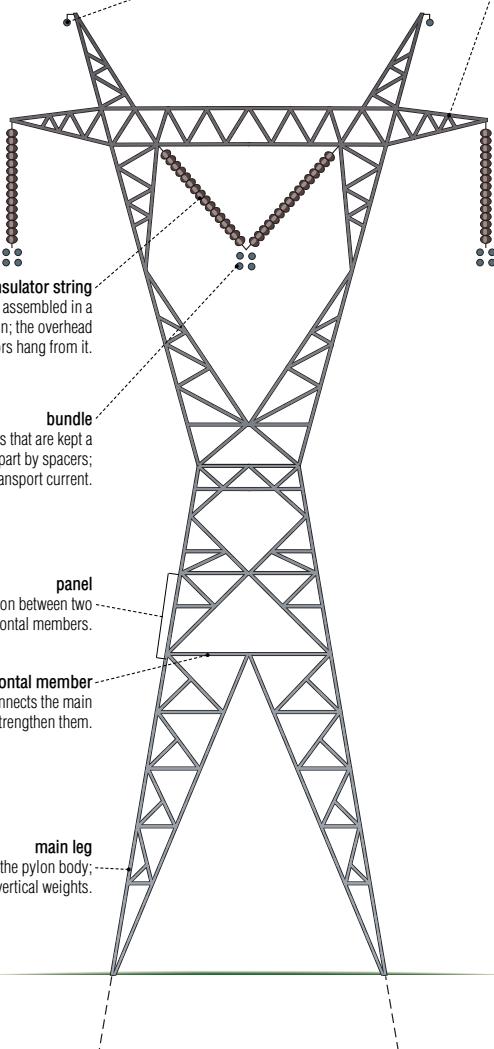
Metal beam that supports the electric conductors along the overhead transportation lines.

overhead ground wire

Conductor that is connected to the ground and attached above the bundles of the overhead lines to protect them from lightning.

crossarm

Horizontal element that protrudes on each side of the pylon; it supports the bundles by means of suspension insulator strings.

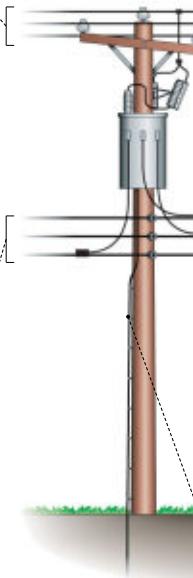


network connection

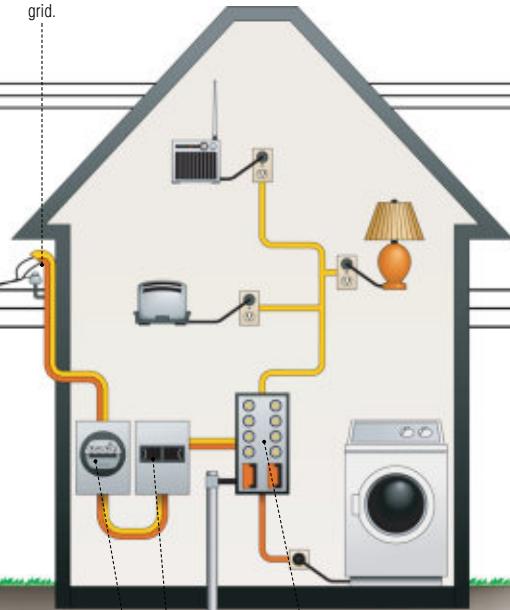
Set of equipment and conductors allowing a customer's electric installation to be connected to the public grid.

medium-tension distribution line

Overhead electricity-distribution lines with tension between 750 and 50,000 volts; its conductors are located at the top of the poles.

**connection point**

Place where the customer's electric hookup is connected to the electric grid.

**low-tension distribution line**

Overhead electricity-distribution line with a maximum tension of 750 volts; its conductors are located under the transformer.

ground wire

Metal conductor inserted into the ground ensuring that accidental electric leakages are conducted to the earth.

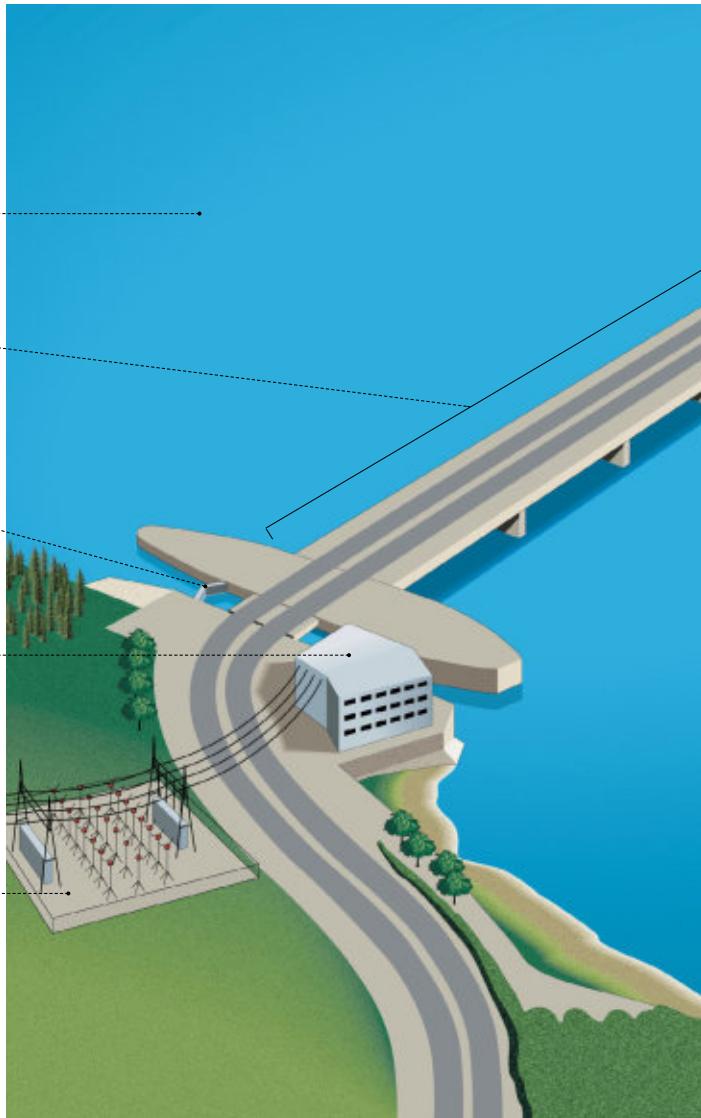
electricity meter
Device measuring the consumption of electricity by a dwelling.

distribution panel
Set of devices forming the junction of the public electricity grid and the electric circuits of a dwelling.

main switch
Mechanism allowing a dwelling's current to be cut off.

tidal power plant

Plant that harnesses tidal power (the motion of the rising and falling tides) to produce electric power.



sea

Vast body of saltwater at some distance inland; it is not as deep as an ocean.

power plant

Part of the dam housing bulb units that are powered by the rise and fall of the sea to produce electricity.

lock

Structure with doors and gates that is built between the sea and the basin; it allows boats to pass from one level to the other.

administrative building

substation

The devices (such as transformers and changeover switches) that increase the voltage of the electricity and carry it to the network.

**bank**

Strip of land bordering the sea.

gate

Movable vertical panel that controls the rate of flow of the water between the sea and the basin.

operating dam

Structure with gates that control the basin level in relation to the level of the sea.

inactive dike

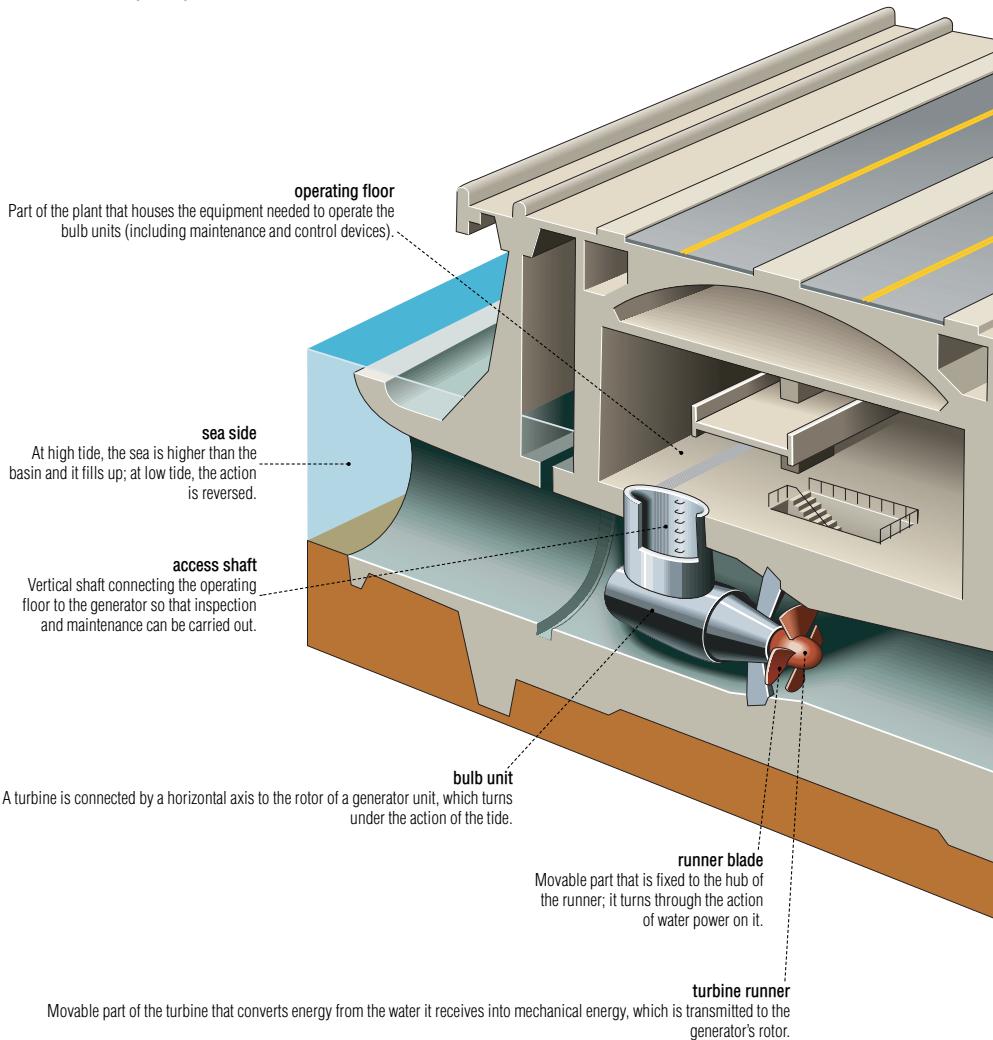
Part of the dam made up mainly of rocky material; it is built between the plant and the operating dam to separate the basin from the sea.

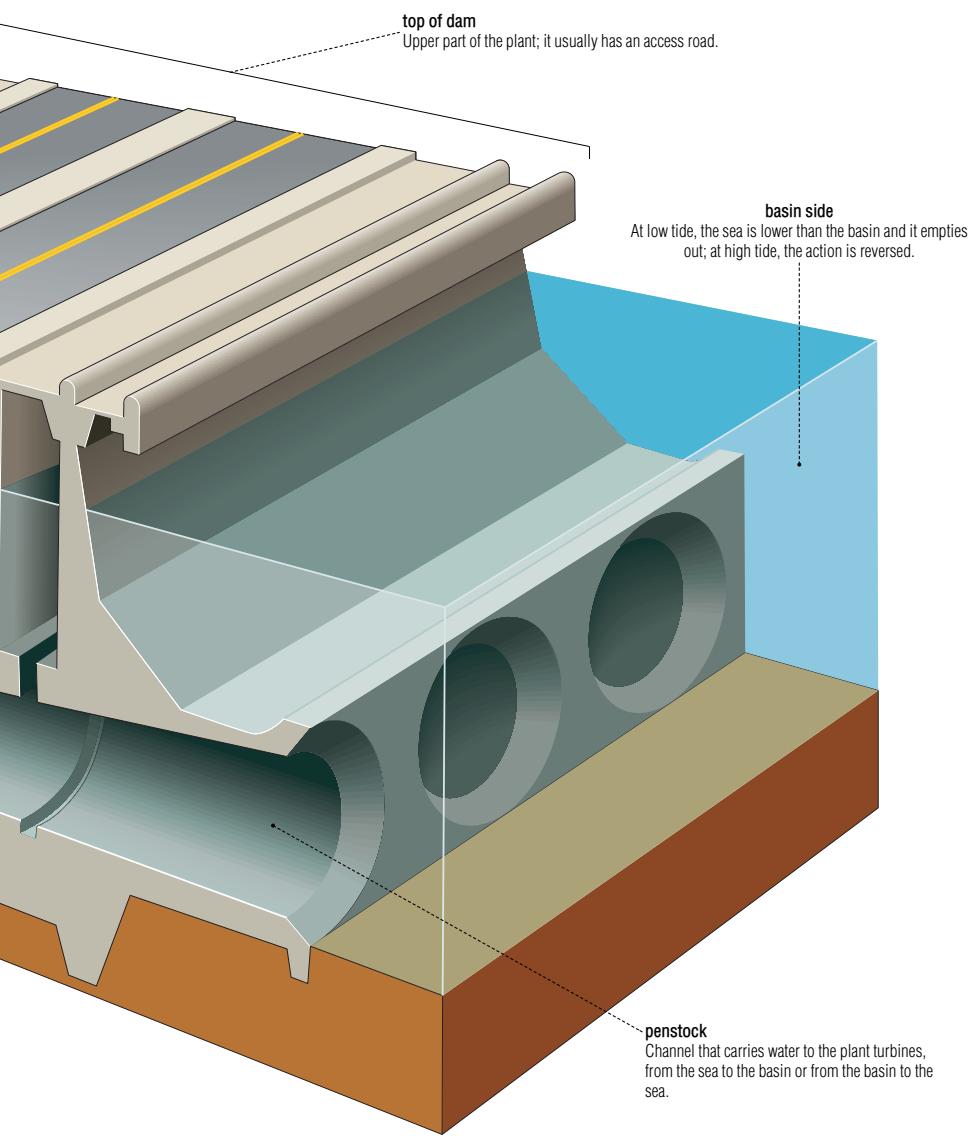
basin

Area in which water is stored at high tide; the basin empties out through the penstocks at low tide.

tidal power plant

cross section of a power plant



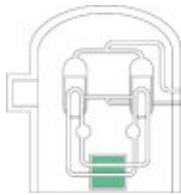


production of electricity from nuclear energy

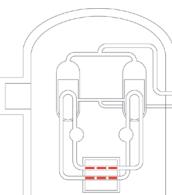
A nuclear fission chain reaction is started and controlled inside the reactor to produce electricity.

**coolant**

Liquid or gas (including heavy water and carbon dioxide) that circulates inside the reactor; it harnesses and transports the heat released during fission of the fuel.

**moderator**

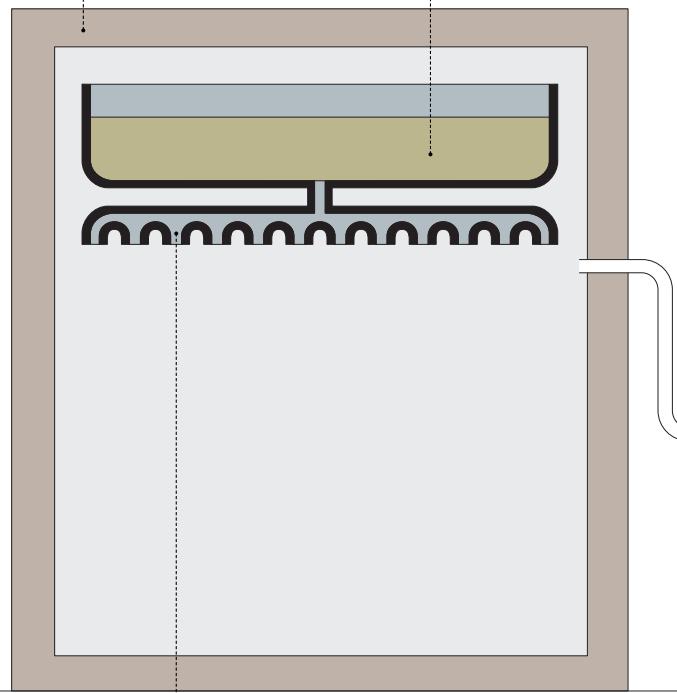
Substance (ordinary water, heavy water, graphite) that slows the fast-moving neutrons emitted during fission to increase the probability of new collisions.

**fuel**

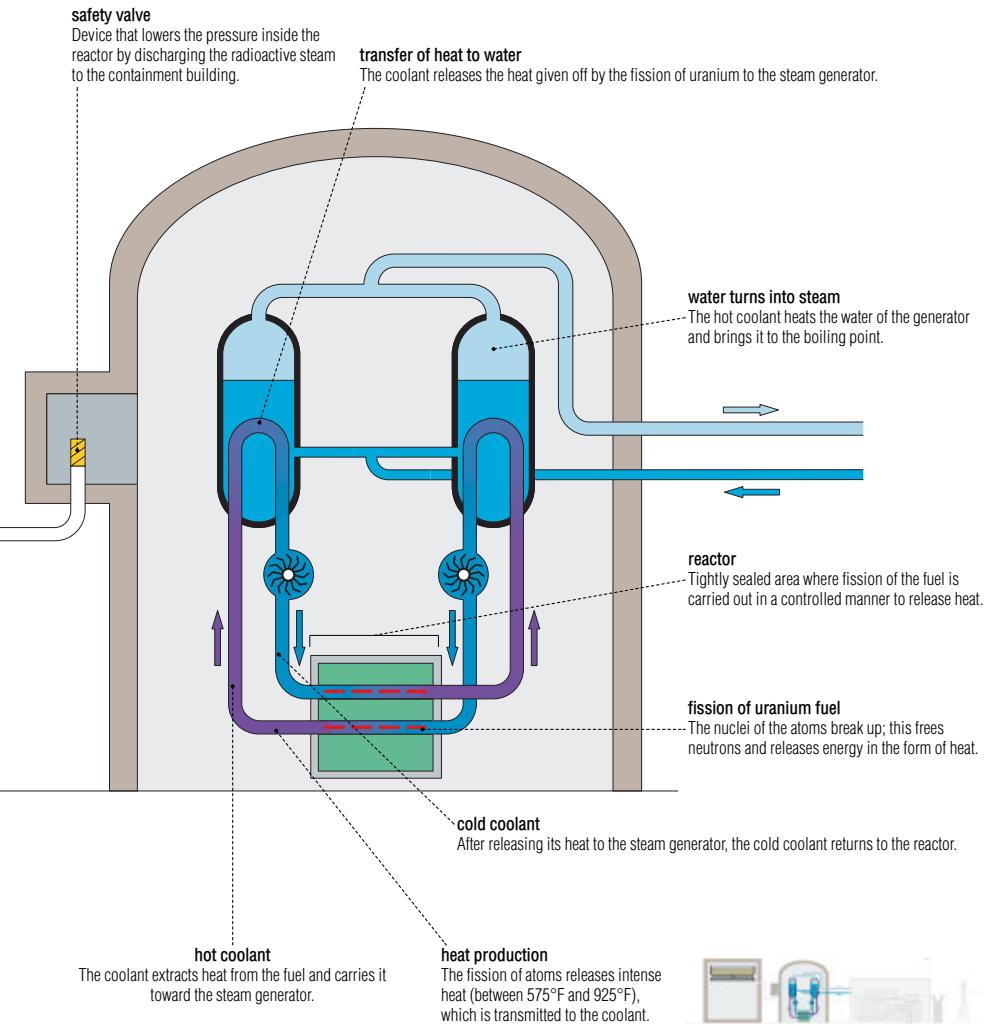
Matter placed in the core of the reactor that contains heavy atoms (uranium, plutonium); energy is extracted from it by fission.

containment building
Concrete building used to collect the radioactive steam from the reactor in the event of an accident.

dousing water tank
Vat that contains water to cool the radioactive steam in the reactor in the event of an accident; this prevents a rise in pressure.

**sprinklers**

Devices that release water to condense radioactive steam.

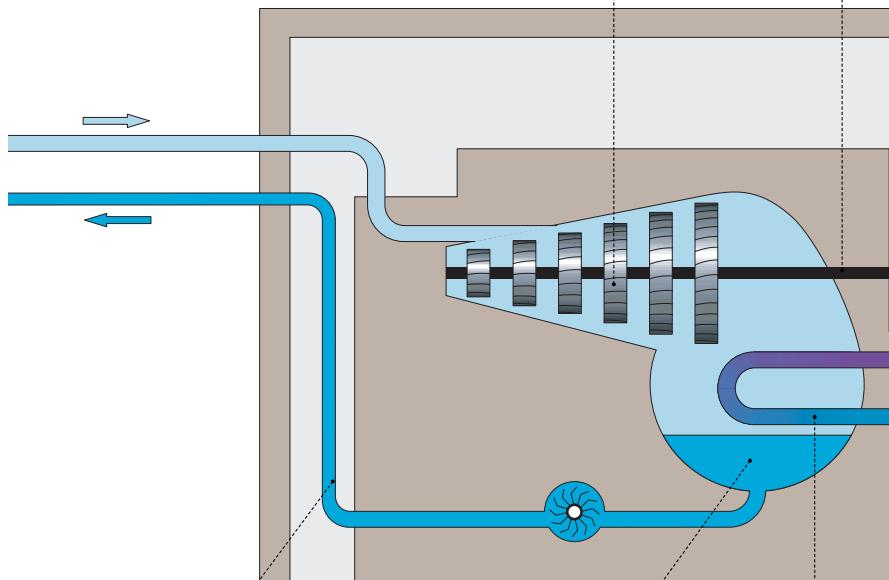


turbine shaft turns generator

The rotational movement of the turbine is transmitted to the generator's rotor.

steam pressure drives turbine

Steam from the steam generator turns the turbine runner, which is connected to the generator.

**water is pumped back into the steam generator**

After passing through the turbine, water produced by the condensation of the steam returns to the steam generator.

condensation of steam into water

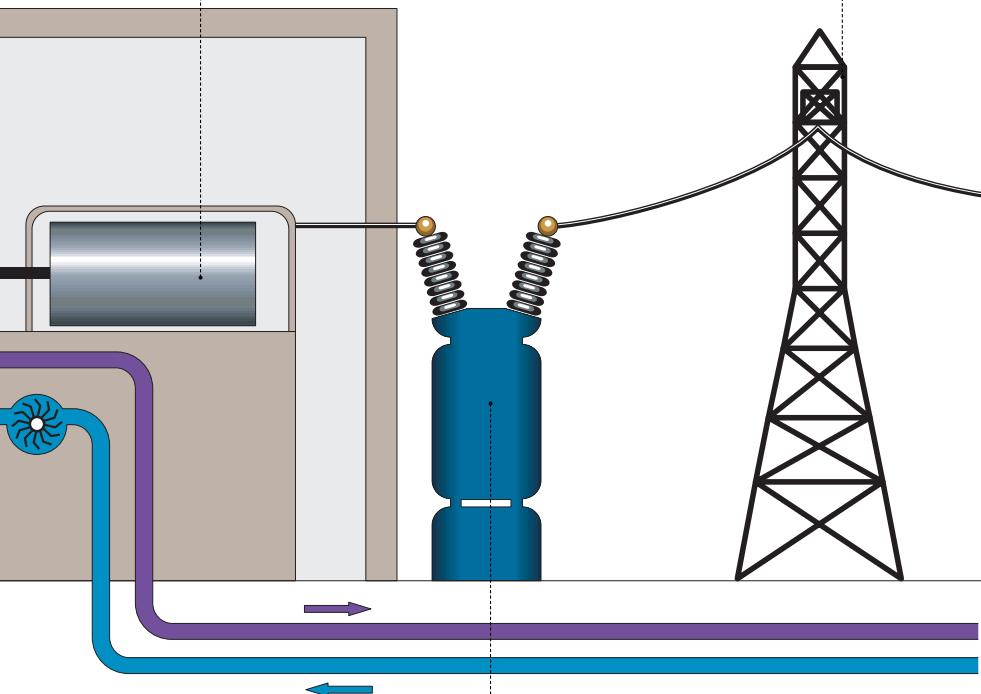
At the turbine outlet, the steam cools and condenses into water.

water cools the used steam

Cooling of the steam from the turbine is done with river or lake water.

production of electricity by the generator

The generator produces electricity through the movement of the rotor in the stator.

**electricity transmission**

Using high-voltage lines to transmit electricity over long distances reduces the strength of the current and, as a result, energy losses.

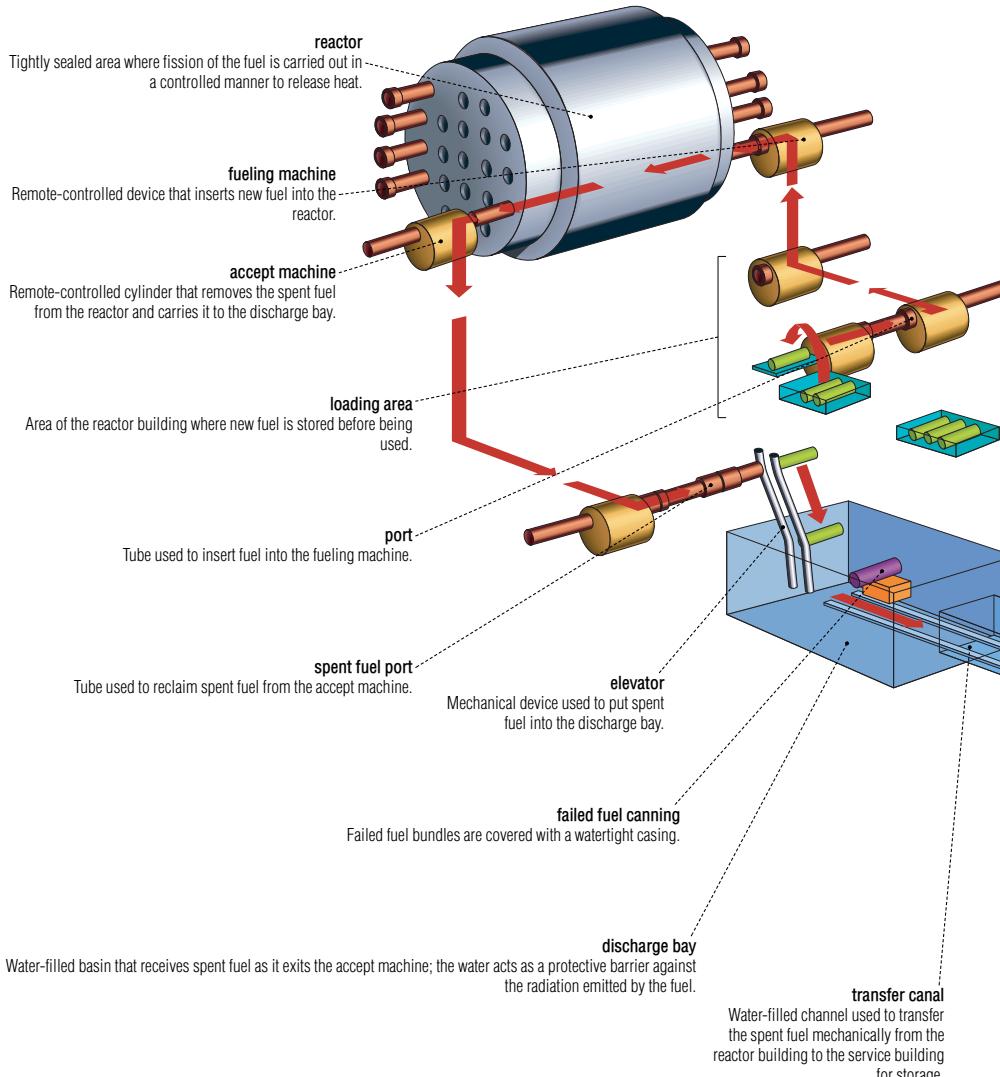
voltage increase

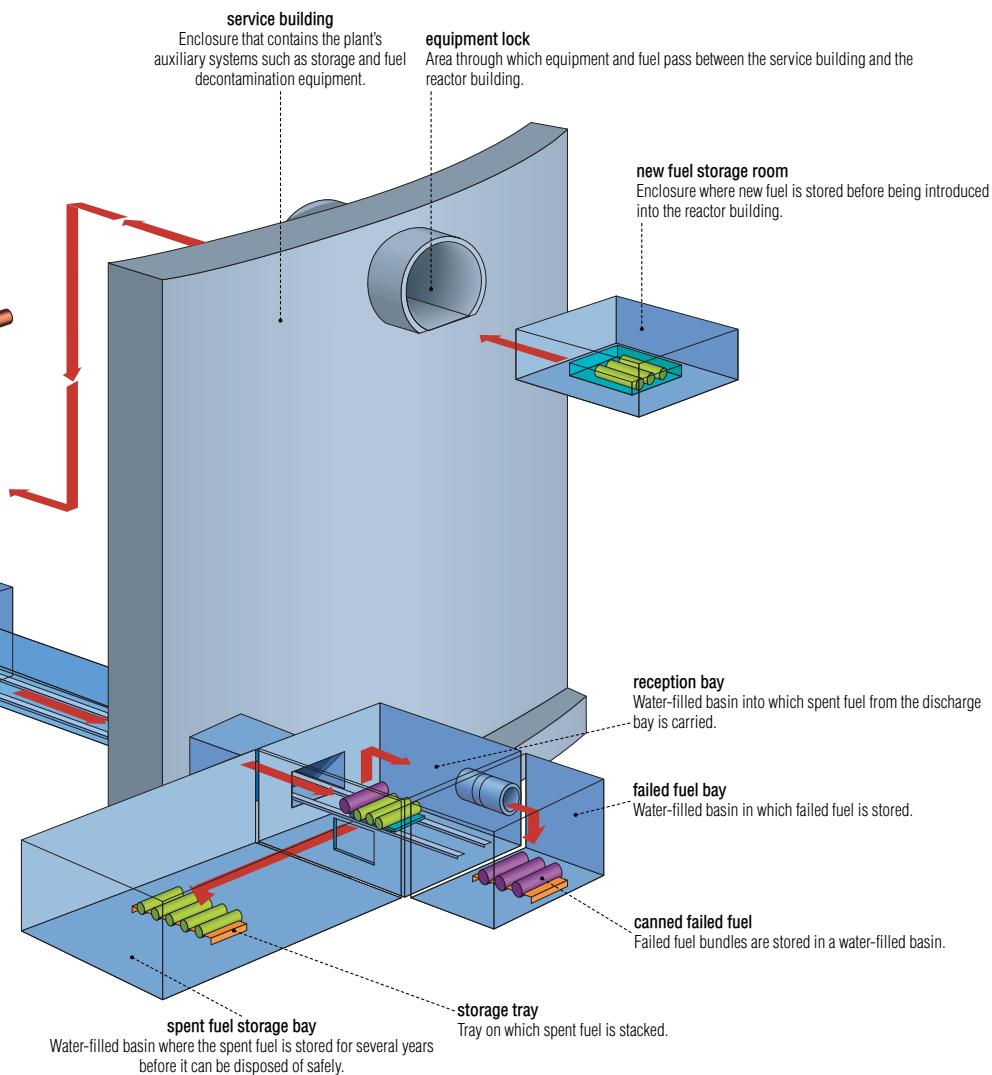
At the outlet end of the power plant, the transformer increases the voltage; this reduces energy losses during transmission over long distances.



fuel handling sequence

Uranium is made into pellets, which are pressed into fuel bundles to be used in the reactor and then stored in cooling bays.





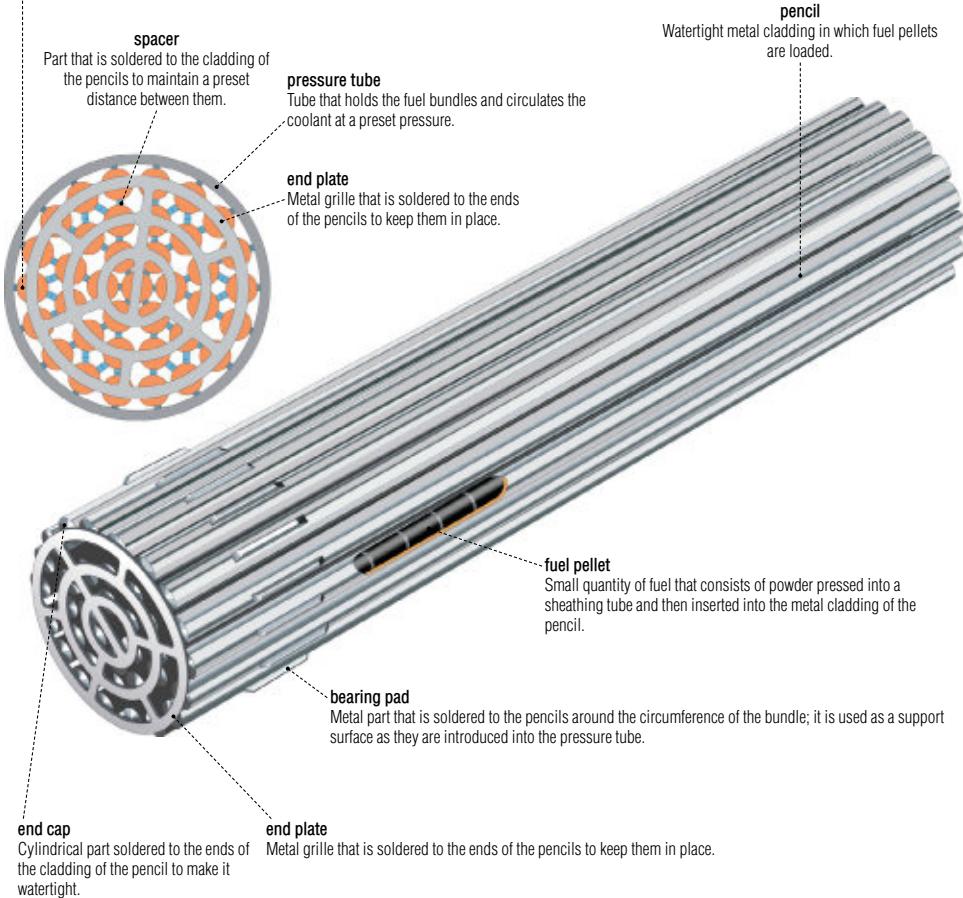
Water-filled basin where the spent fuel is stored for several years before it can be disposed of safely.

fuel bundle

Fuel pencils that are grouped in parallel for introduction into the reactor.

pencil

Watertight metal cladding in which fuel pellets are loaded.



Tightly sealed area where fission of the fuel is carried out in a controlled manner to release heat.

fuel pellet

Small quantity of fuel that consists of powder pressed into a sheathing tube and then inserted into the metal cladding of the pencil.

fuel bundle

Fuel pencils that are grouped in parallel for introduction into the reactor.

reactor building

Concrete structure surrounding the reactor vessel; it is a protective barrier against radioactivity.

containment building

Concrete structure surrounding the reactor vessel; it is a protective barrier against radioactivity.


pressure tube

Tube that holds the fuel bundles and circulates the coolant at a preset pressure.


reactor vessel

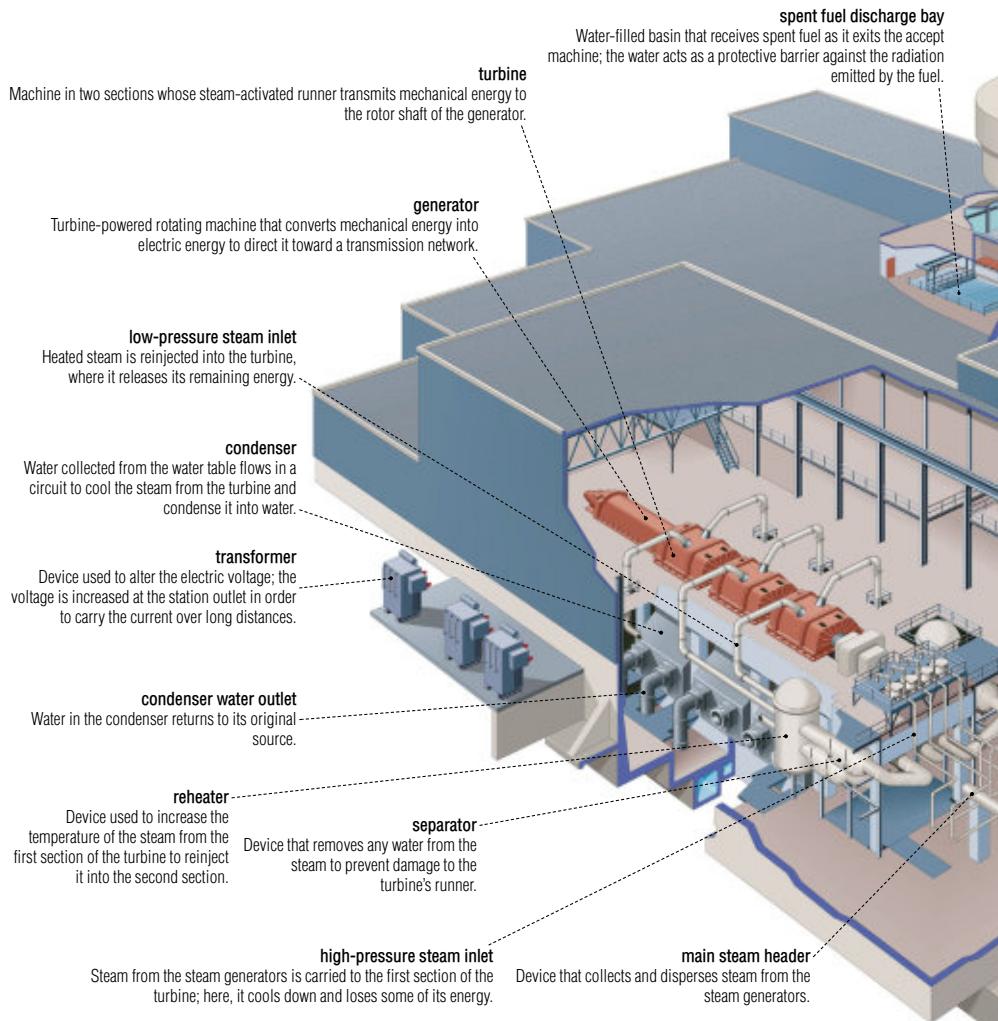
The core of the nuclear reactor consists of tubular spaces where fission is produced and the coolant and moderator circulate.

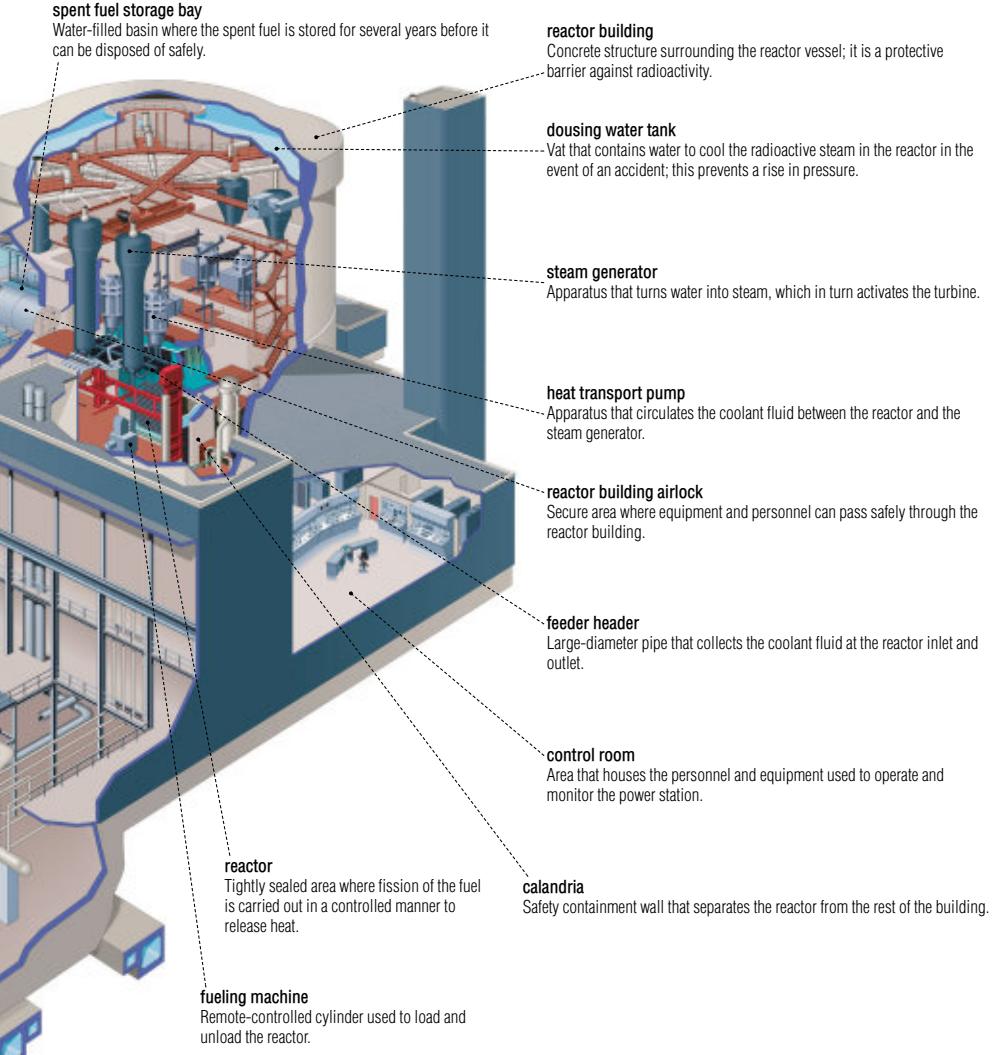

spent fuel storage bay

Water-filled basin where the spent fuel is stored for several years before it can be disposed of safely.

nuclear generating station

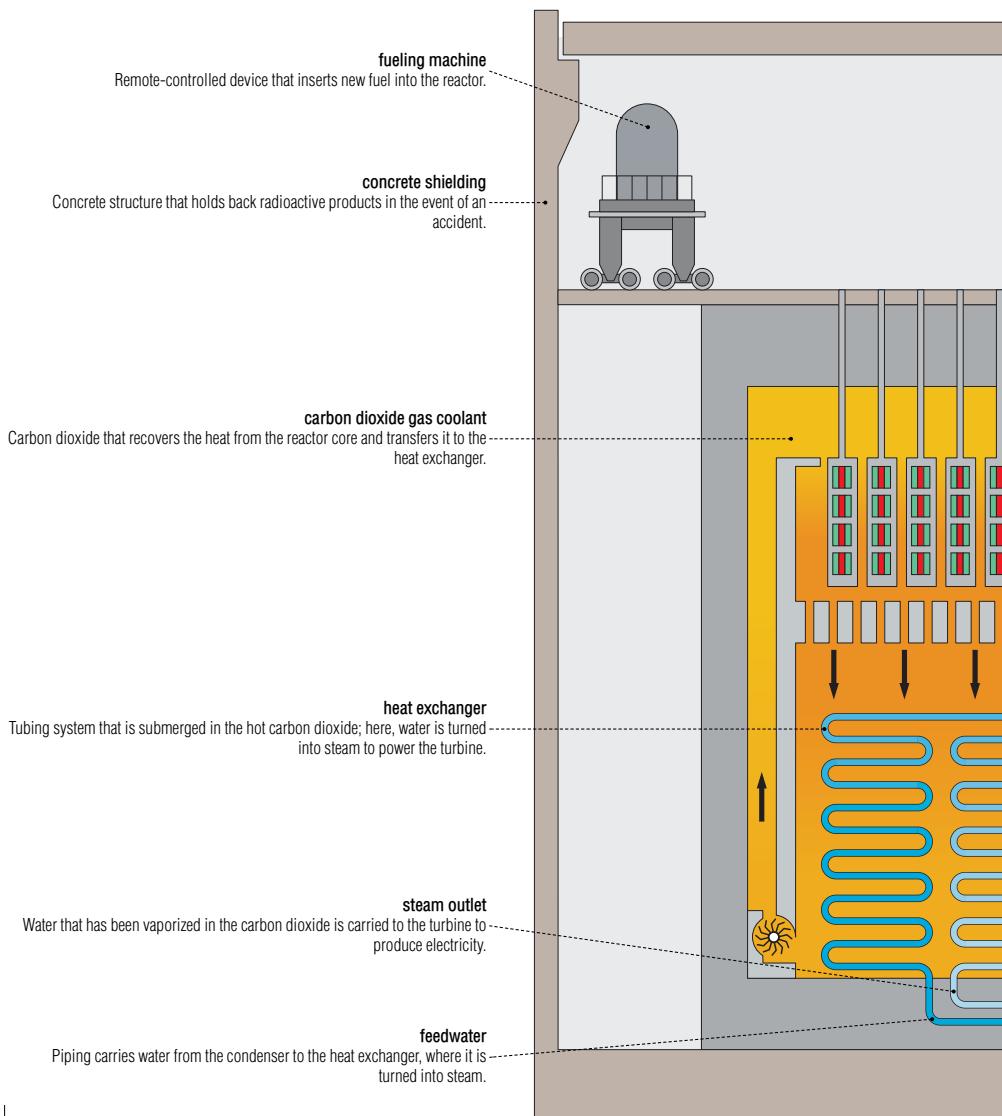
Plant that produces electricity from thermal energy generated by the fission of fuel atoms in a reactor.

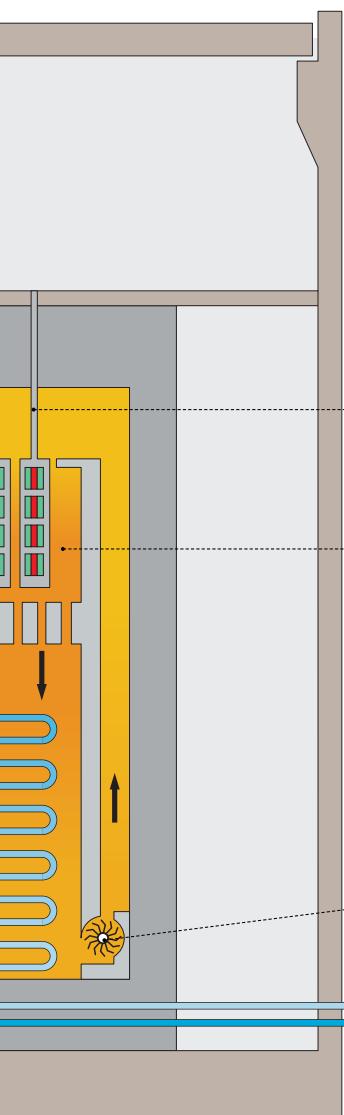




carbon dioxide reactor

Developed for the most part in Great Britain and France, it was replaced by the pressurized water reactor, which performs better and is less expensive.



**control rod**

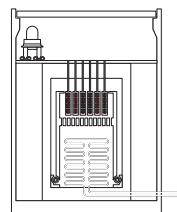
Tube that contains a neutron-absorbing material (boron, cadmium) that is introduced into the reactor core to control its power.

reactor core

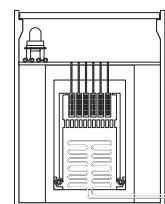
Center section of the nuclear reactor where fission reactions take place.

blower

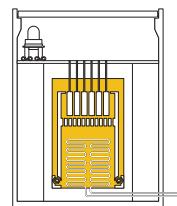
Device that circulates carbon dioxide in the reactor core.

**fuel: natural uranium**

Natural uranium: fuel extracted from mines; it consists of a mixture of three uranium isotopes (uranium-234, -235 and -238).

**moderator: graphite**

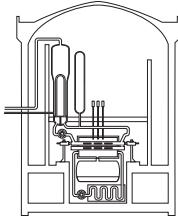
Moderator: medium that slows the speed of the neutrons to maintain a continuous chain reaction.

**coolant: carbon dioxide**

Carbon dioxide: gas that is heavier than air and is produced by burning graphite.

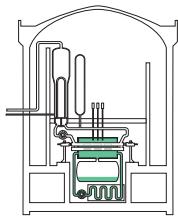
heavy-water reactor

The advantage of this type of reactor is that it does not require fuel enrichment; it is used mainly in Canada, Argentina and India.



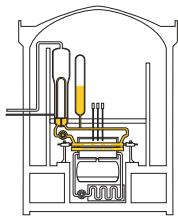
fuel: natural uranium

Natural uranium: fuel extracted from mines; it consists of a mixture of three uranium isotopes (uranium-234, -235 and -238).



moderator: heavy water

Heavy water: water consisting of heavy hydrogen (deuterium) and oxygen; it can slow down neutrons.



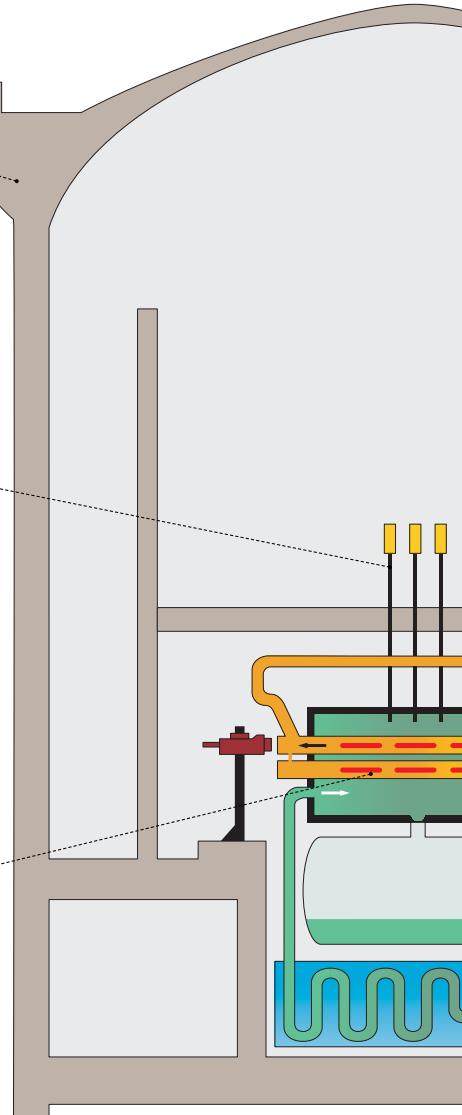
coolant: pressurized heavy water

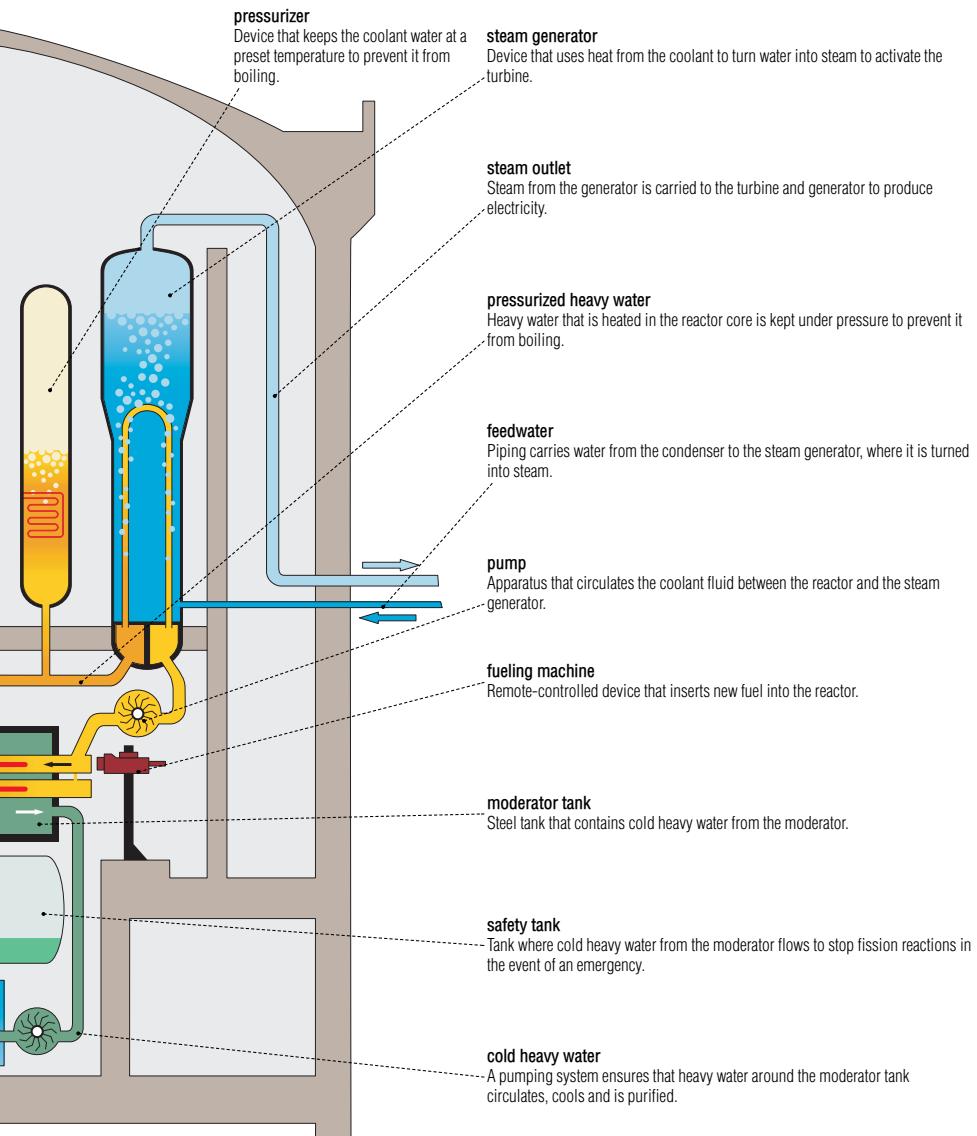
Heavy water is kept at a set pressure to prevent it from boiling.

concrete shielding
Concrete structure that holds back radioactive products in the event of an accident.

control rod
Tube that contains a neutron-absorbing material (boron, cadmium) that is introduced into the reactor core to control its power.

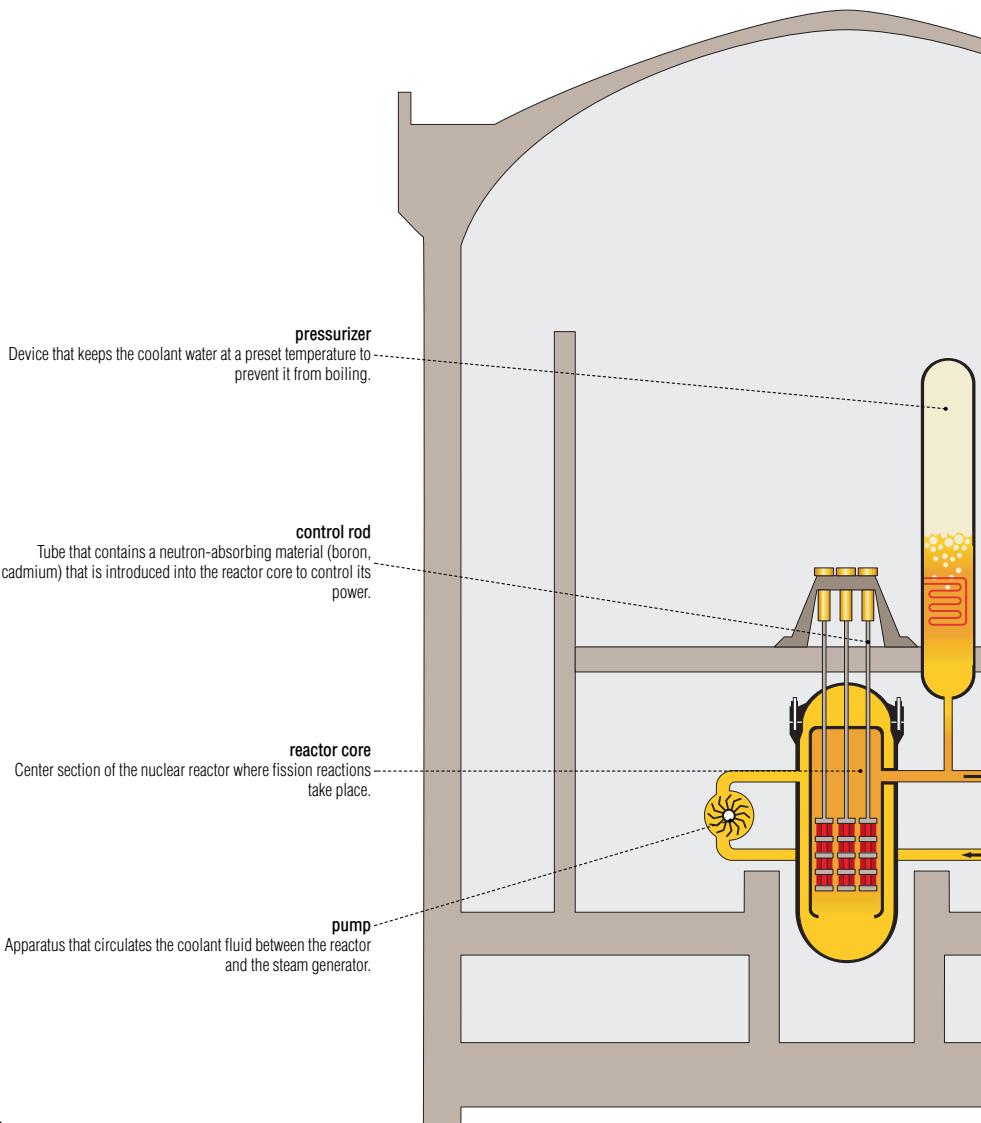
fuel
Matter that is placed in the reactor core; it contains heavy atoms (uranium, plutonium) from which power is extracted by fission.

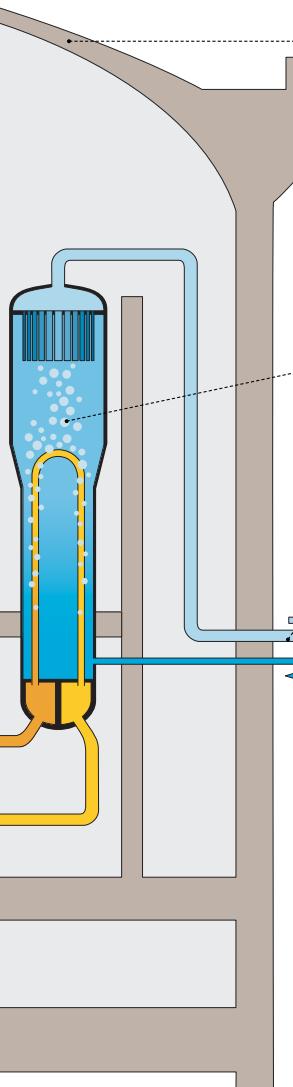




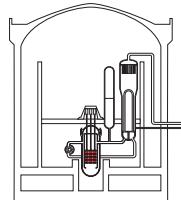
pressurized-water reactor

The most common type of reactor in the world; water from the coolant is kept under heavy pressure to prevent it from vaporizing.



**concrete shielding**

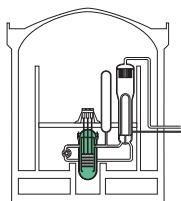
Concrete structure that holds back radioactive products in the event of an accident.

**fuel: enriched uranium**

Enriched uranium: uranium produced by treating natural uranium to increase the quantity of fissionable isotopes (uranium-235) contained in it.

steam generator

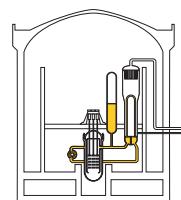
Device that uses heat from the coolant to turn water into steam to activate the turbine.

**moderator: natural water**

Natural water: water found in its natural state.

steam outlet

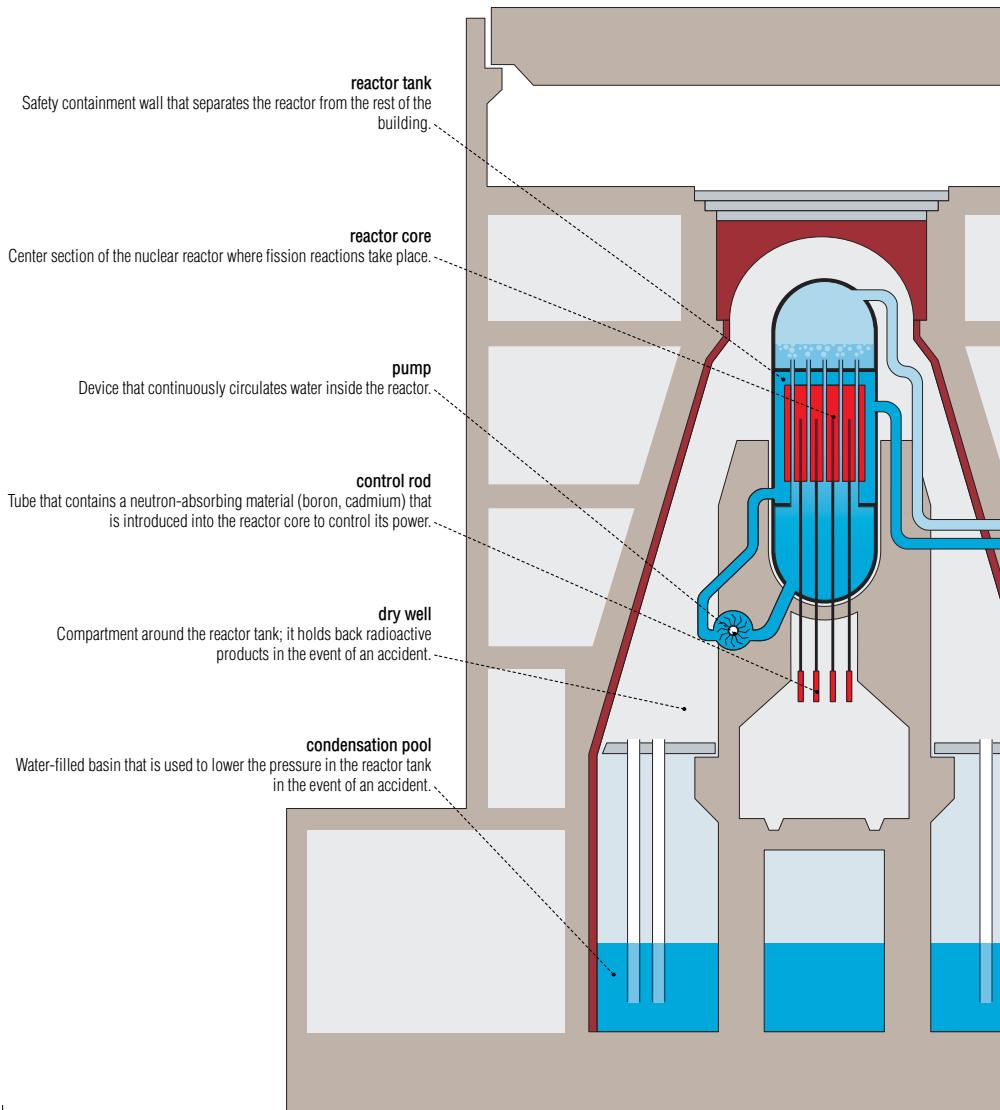
Steam from the generator is carried to the turbine and generator to produce electricity.

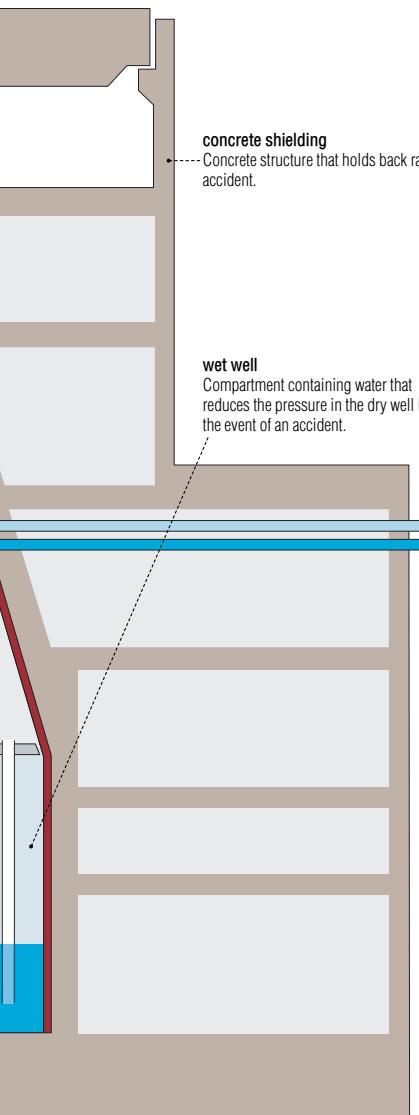
**coolant: pressurized water**

Pressurized water: natural water kept under a preset pressure to prevent it from boiling.

boiling-water reactor

In this second most common reactor, boiling occurs directly in the reactor core; it is used mainly in the United States, Sweden and Japan.



**concrete shielding**

Concrete structure that holds back radioactive products in the event of an accident.

wet well

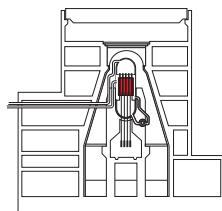
Compartment containing water that reduces the pressure in the dry well in the event of an accident.

steam outlet

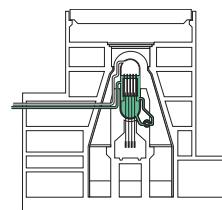
Steam produced in the reactor tank is carried to the turbine to produce electricity.

feedwater

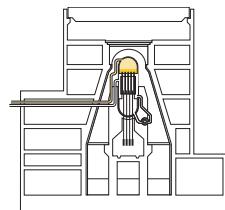
Piping that carries water from the condenser into the reactor tank, where it is converted into steam.

**fuel: enriched uranium**

Enriched uranium: uranium produced by treating natural uranium to increase the quantity of fissionable isotopes (uranium-235) contained in it.

**moderator: natural water**

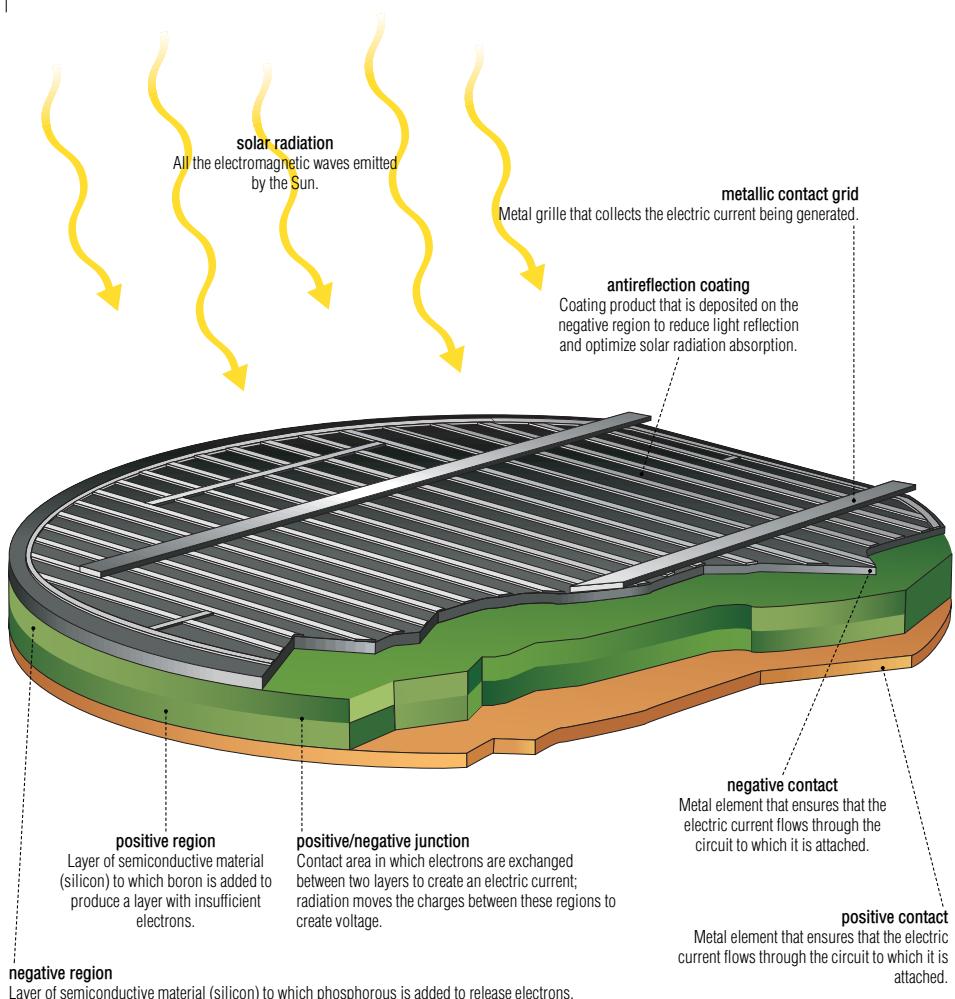
Natural water: water found in its natural state.

**coolant: boiling water**

Boiling water: natural water that boils and vaporizes on contact with the heat released by the fuel.

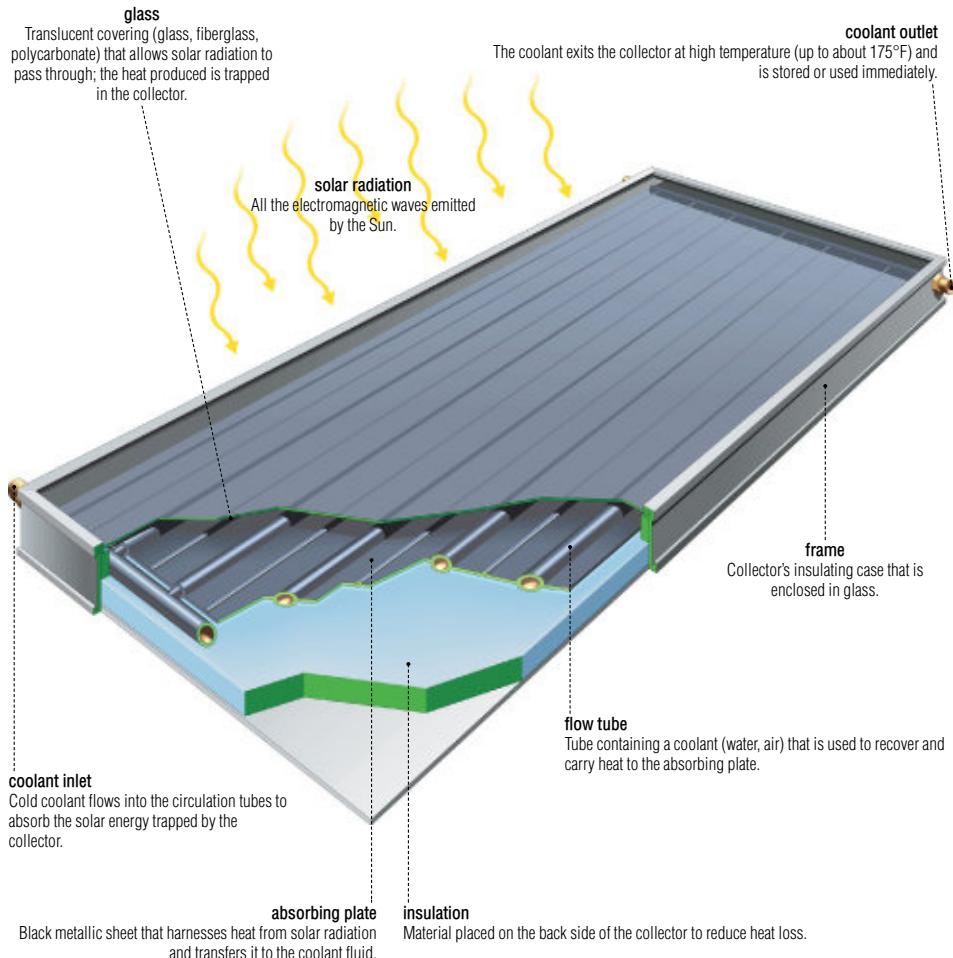
solar cell

Device used to convert solar energy directly into electric energy (photovoltaic effect).



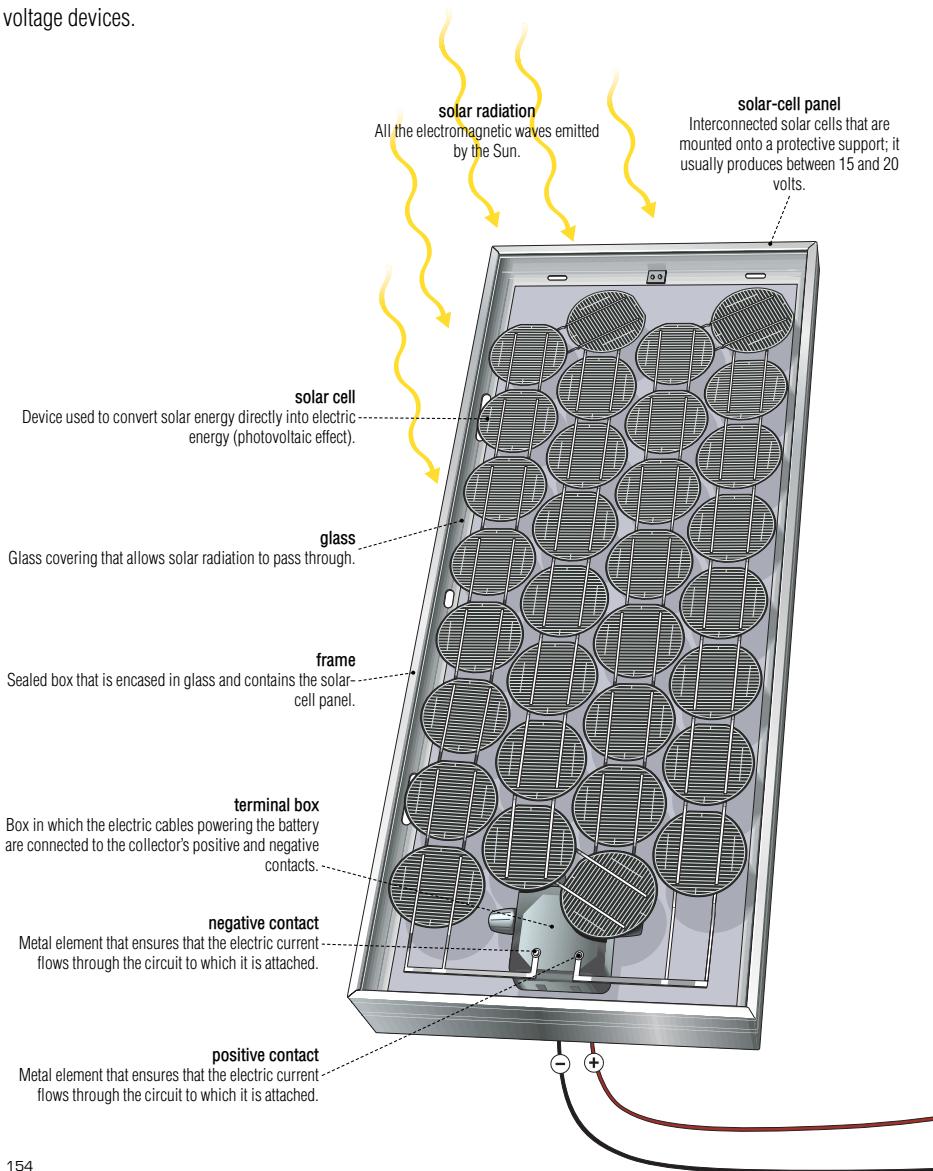
flat-plate solar collector

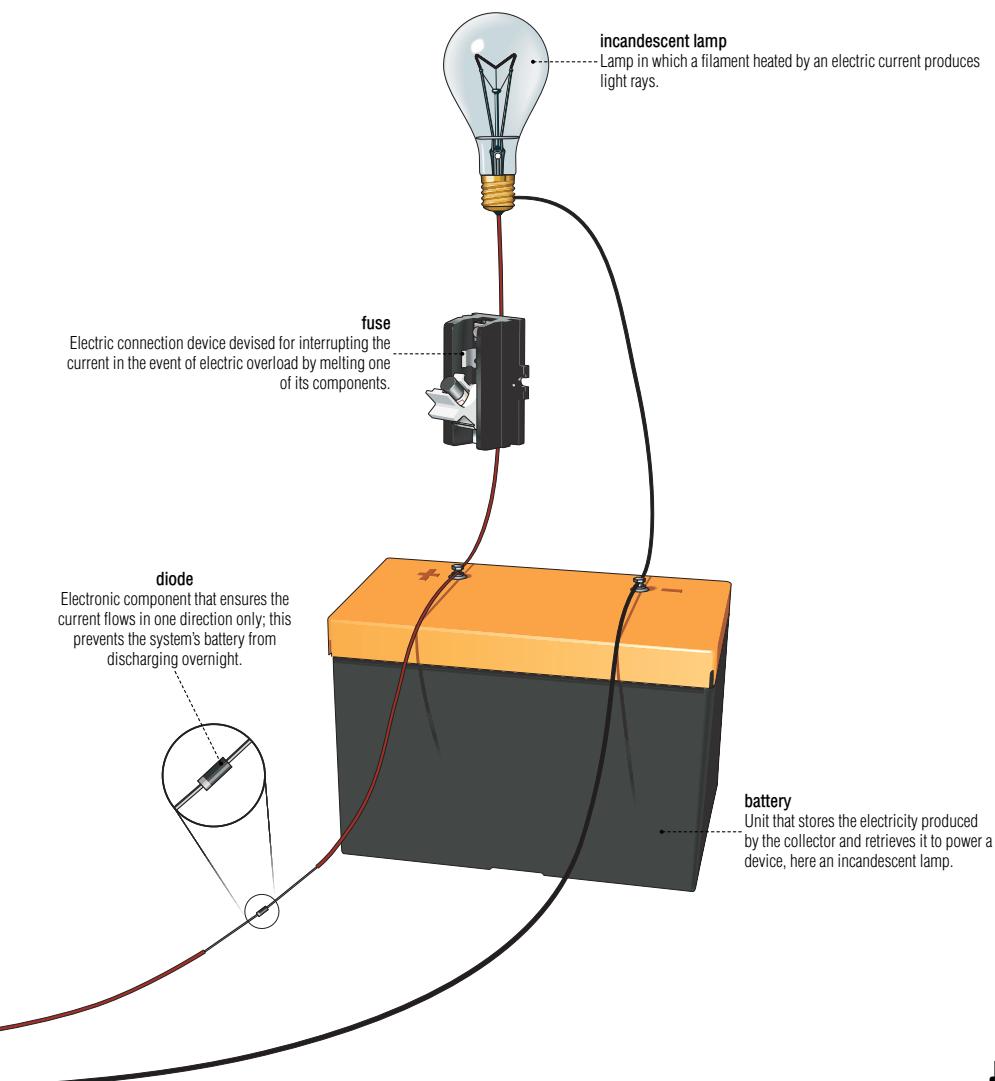
Device that collects solar radiation and heats a coolant, which in turn will be used in residential settings to heat water or the home.



solar-cell system

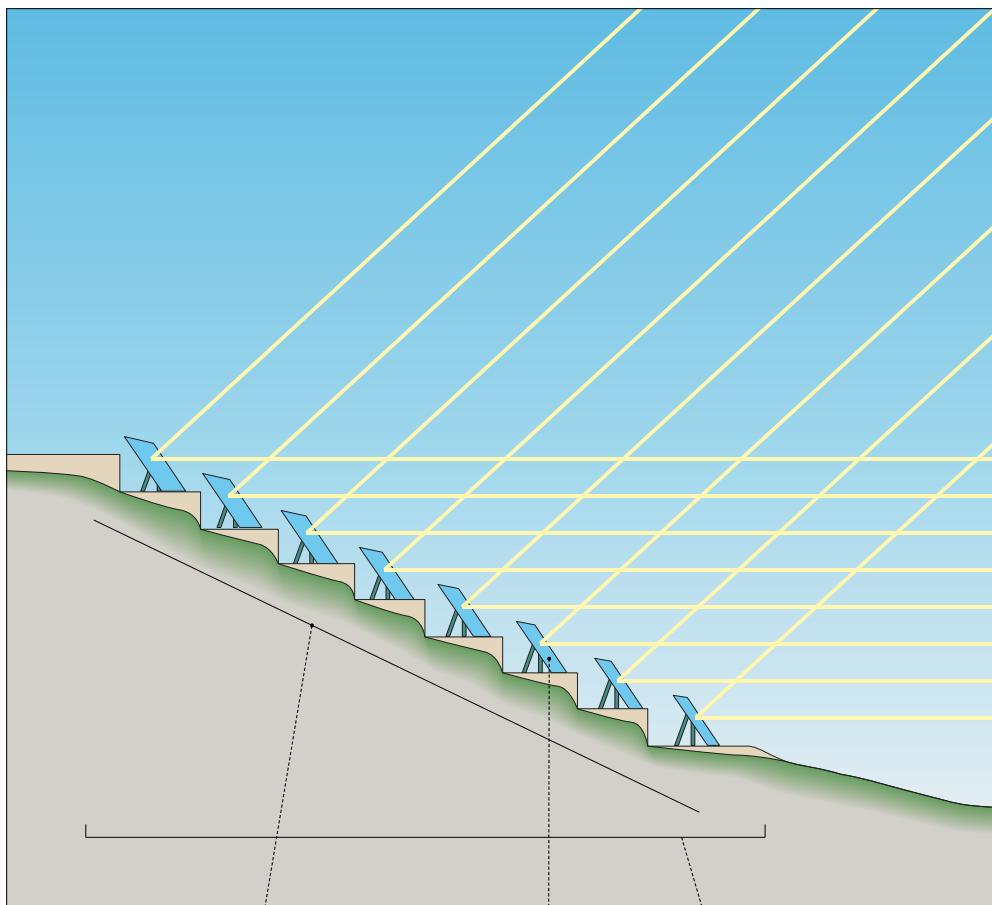
Unit that is usually made up of 36 solar cells, each of which produces a voltage of 0.5 V; it is used to power low-voltage devices.





solar furnace

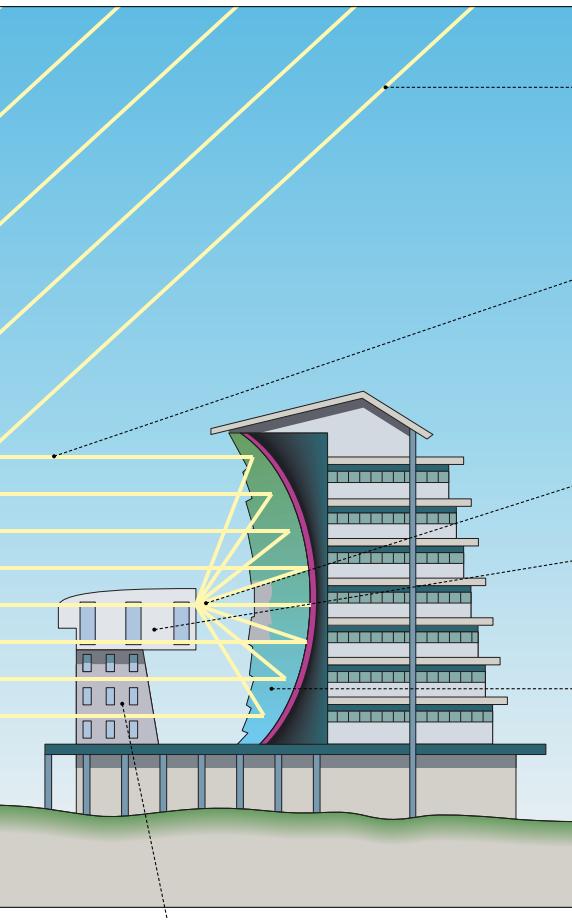
Plant that concentrates solar radiation to reach very high temperatures (over 5,400°F) as part of a research effort to develop experimental materials (including astronautic materials and ceramics).



heliostats
Heliostats are placed on slopes to prevent energy loss due to shade or the interception of reflected rays by neighboring mirrors.

reflecting surface
Polished metallized glass that receives solar radiation and direct it to the parabolic mirror.

bank of heliostats
Heliostats: remote-controlled adjustable mirrors that follow the Sun's trajectory and concentrate solar radiation toward the boiler at the top of the tower.

**solar radiation**

All the electromagnetic waves emitted by the Sun.

solar ray reflected

Solar rays that reach the heliostats are sent to the parabolic mirror.

target area

Point where solar rays reflected by the parabolic mirror converge.

furnace

Reaching temperatures of over 5,400°F, it is mainly used to process and develop materials.

parabolic mirror

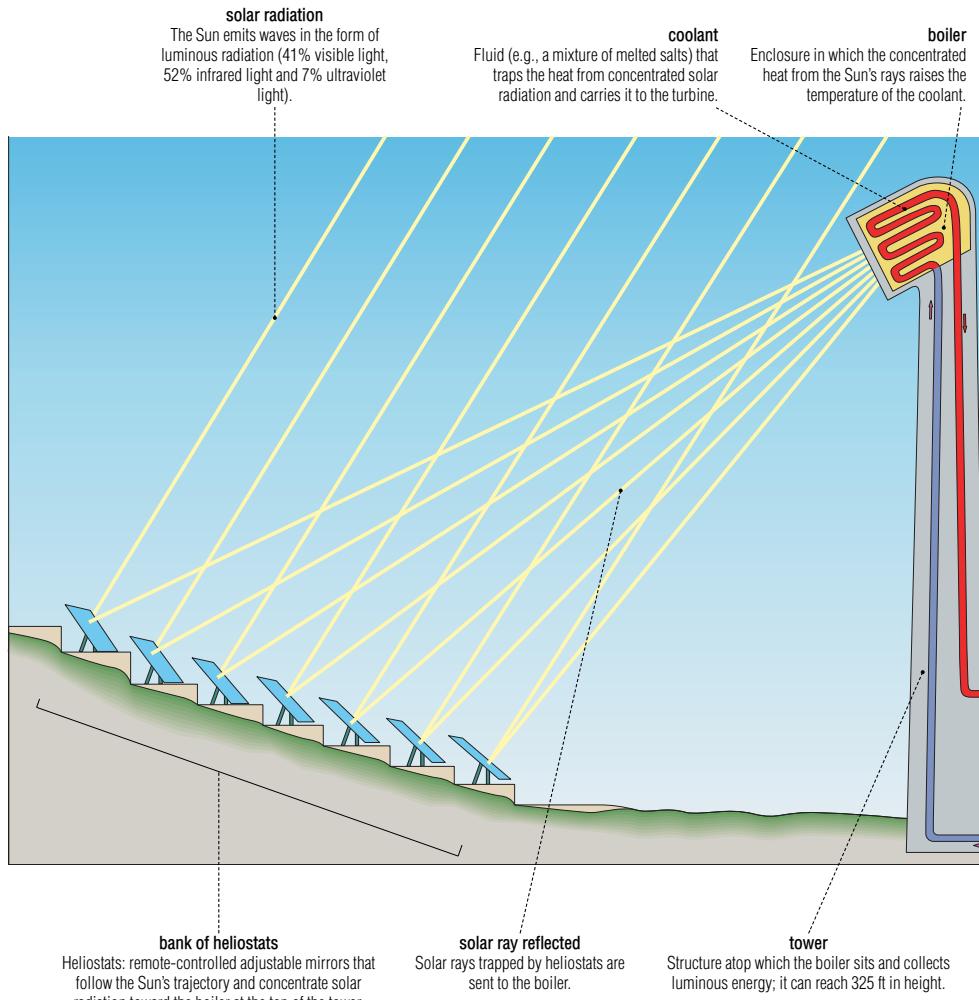
Curved mirror that concentrates the Sun's rays toward one point in the furnace (the target area).

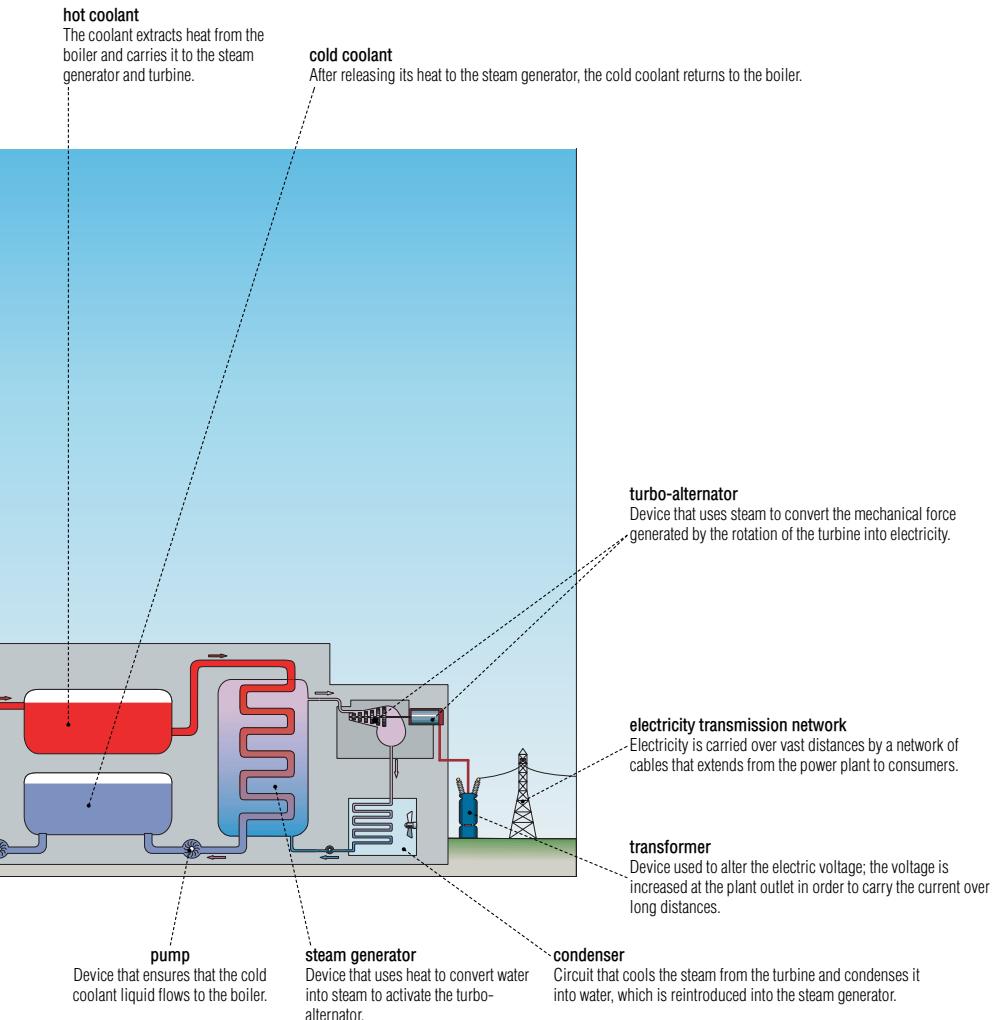
**tower**

Structure atop which the furnace is placed to collect luminous energy; it usually reaches a height of 65 ft.

production of electricity from solar energy

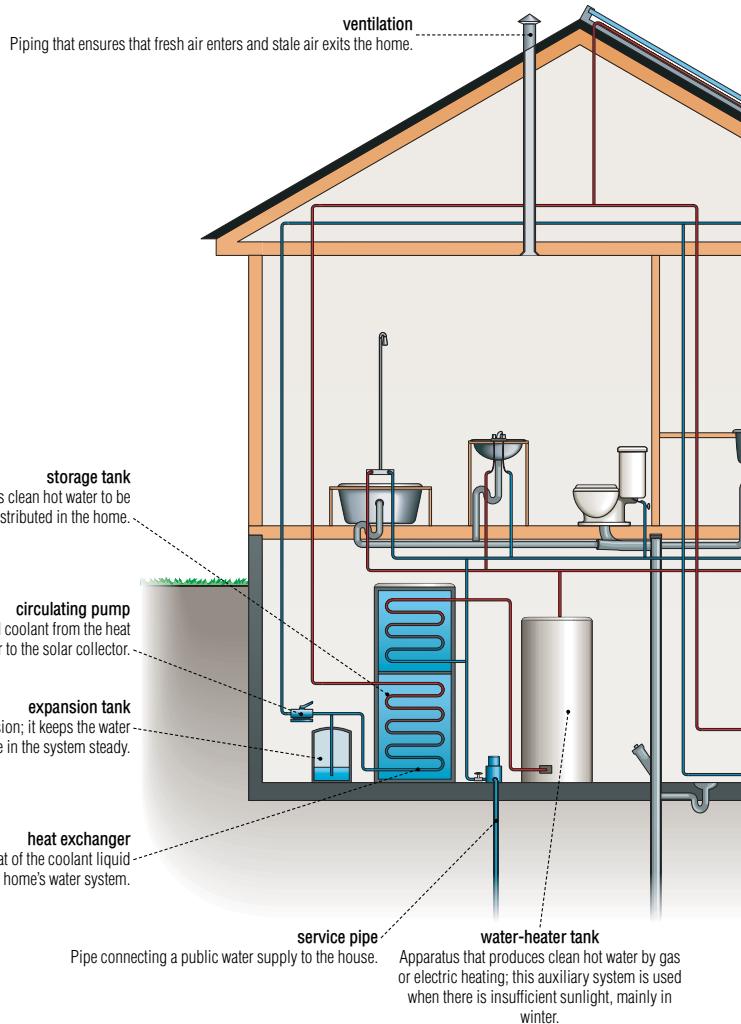
Heating the coolant directly with solar rays turns water into steam, which then turns the turbo-alternator to produce electricity.

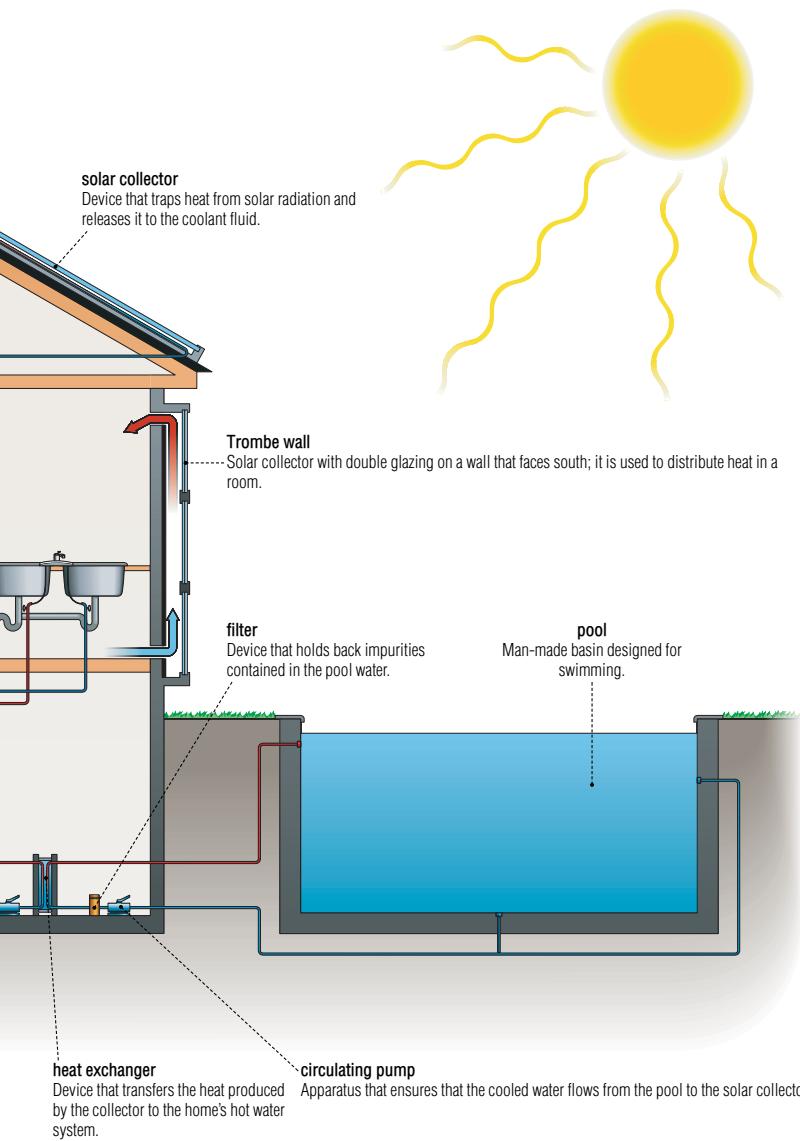




solar house

Solar energy can be used to heat and supply hot water to a home.





solar house

Trombe wall

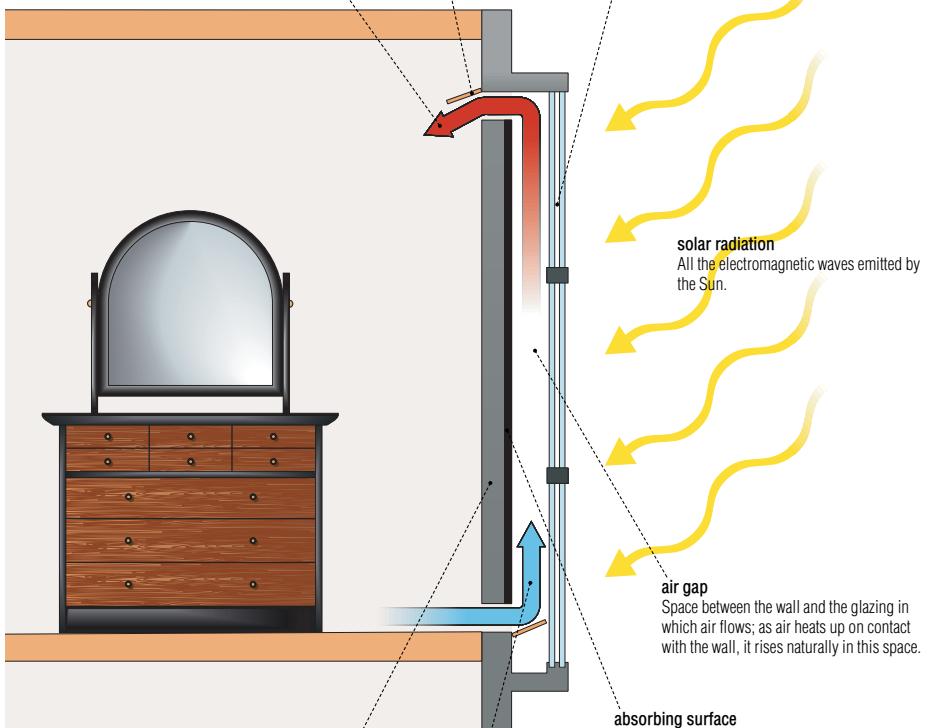
Solar collector with double glazing on a wall that faces south; it is used to distribute heat in a room.

shutter

Flap gate used to control the entry of heat into the home.

warm air

Air heated by solar radiation is introduced into the room by convection.



concrete wall
Masonry structure that is about 15 in thick; it has a black surface to absorb heat from the Sun to heat the air.

cold air
Fresh air enters the home at the base of the air gap and is heated on contact with the wall.

double glazing

Each of two glass plates placed in front of the concrete wall; they allow solar radiation to penetrate and retain the heat.

solar radiation

All the electromagnetic waves emitted by the Sun.

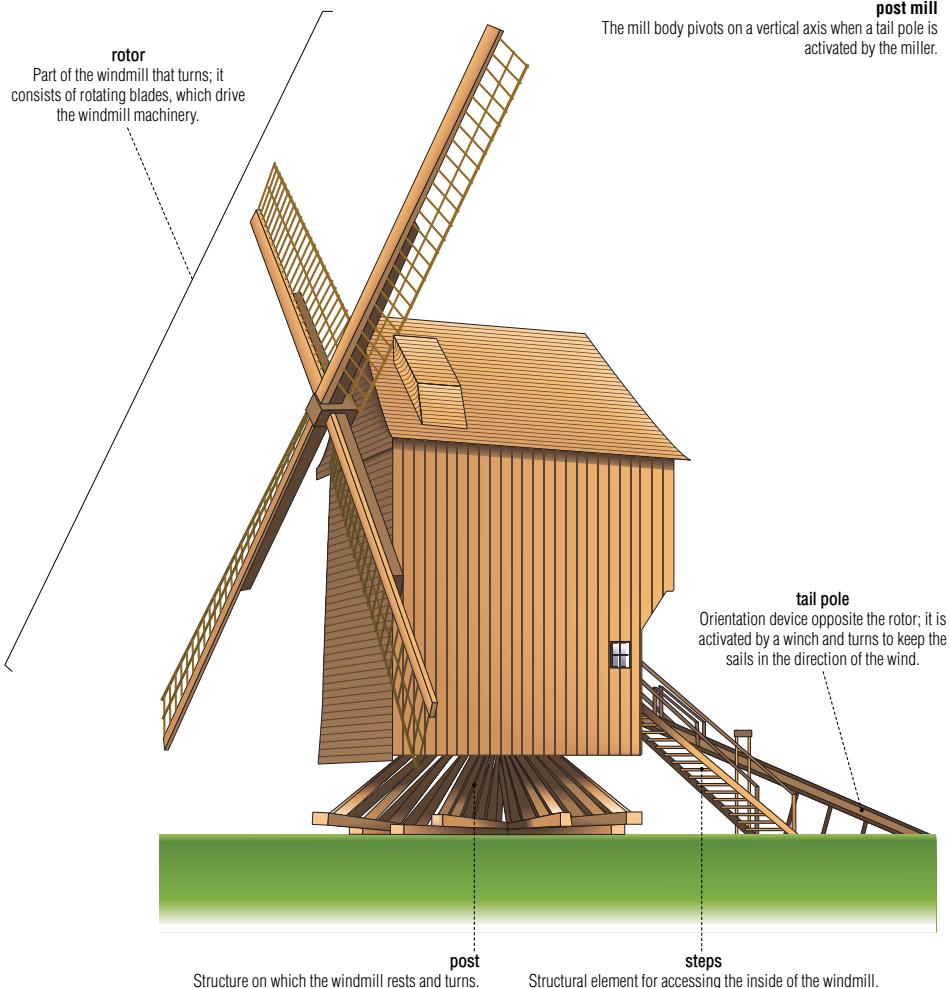
air gap

Space between the wall and the glazing in which air flows; as air heats up on contact with the wall, it rises naturally in this space.

absorbing surface

Black wall that catches solar radiation and converts it into heat.

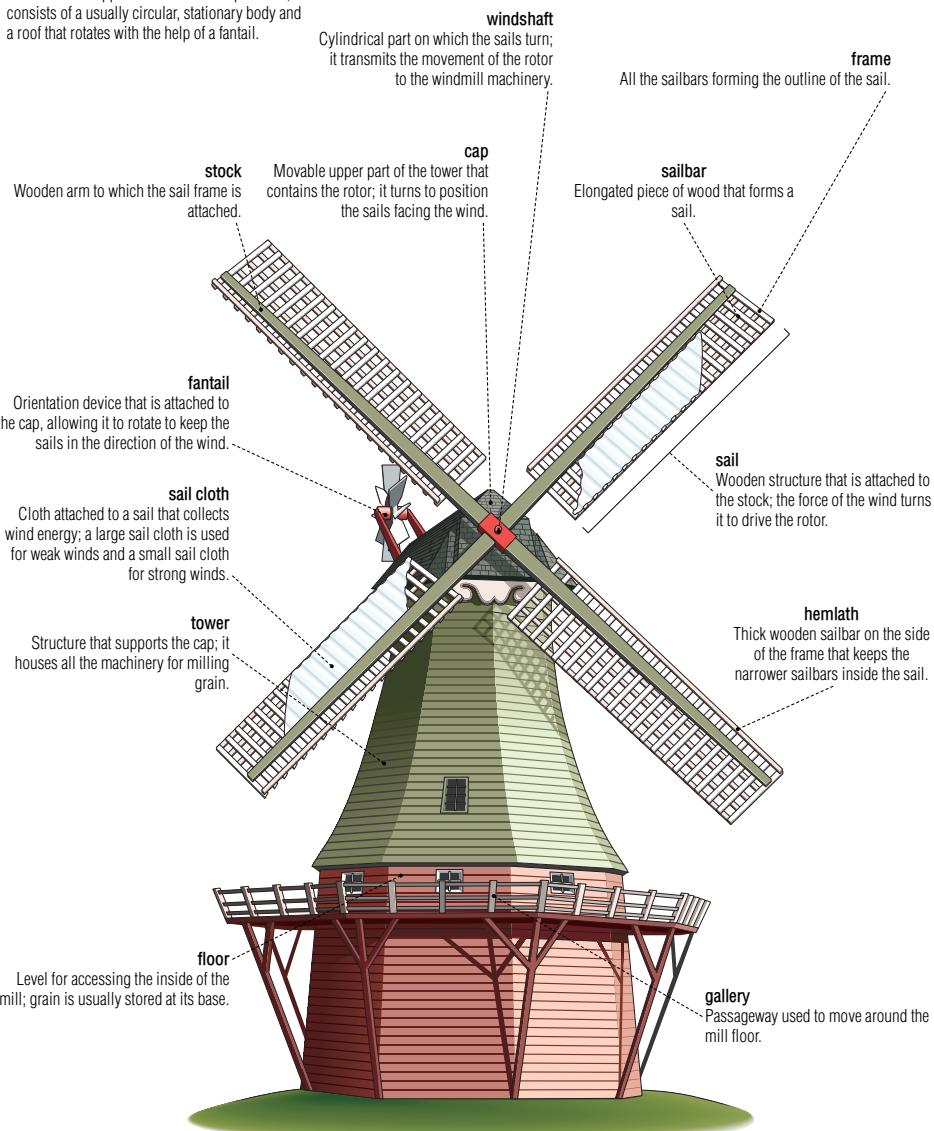
Machine that converts wind energy into mechanical energy; it was used in the past to mill grain and pump water.



windmill

tower mill

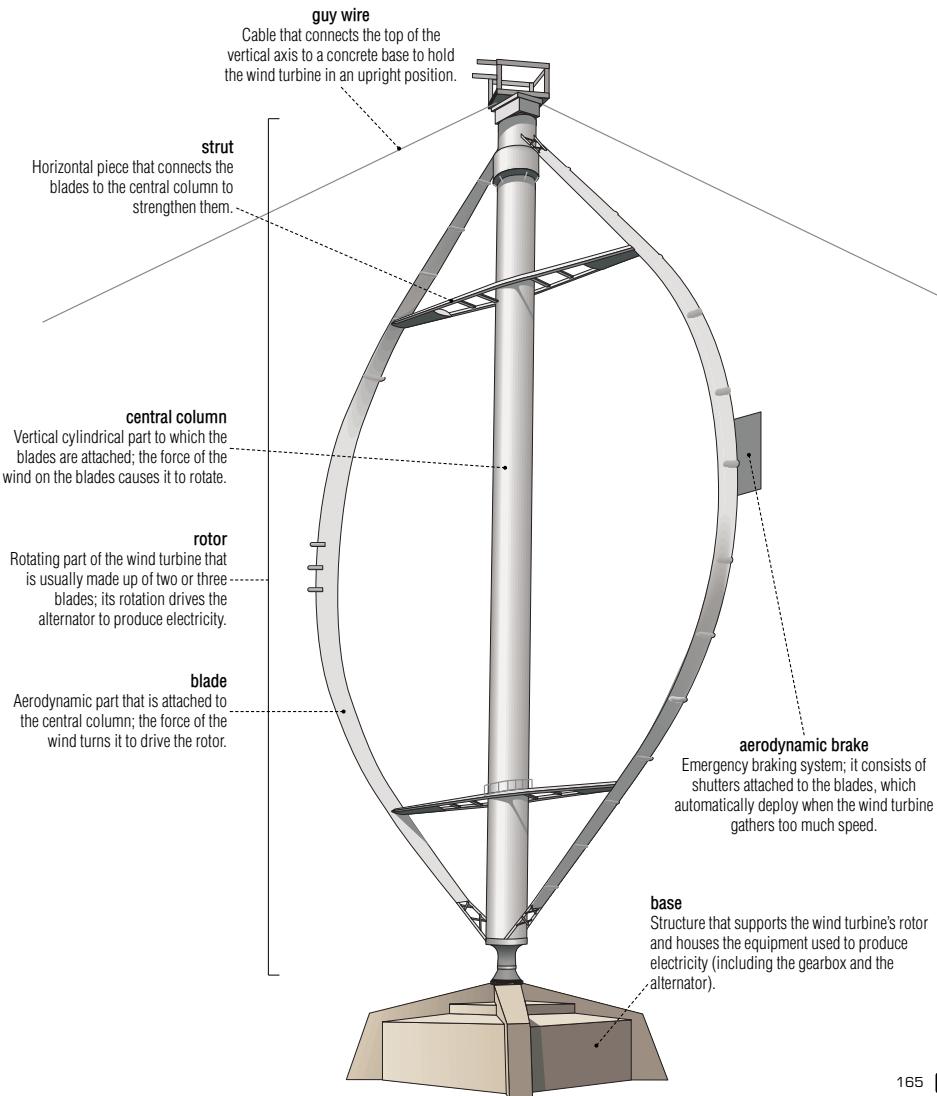
The tower mill appeared later than the post mill; it consists of a usually circular, stationary body and a roof that rotates with the help of a fantail.



Wind turbine: machine that harnesses energy from the wind and converts it into mechanical energy to activate the alternator.

vertical-axis wind turbine

Wind turbine whose axis is perpendicular to the wind.

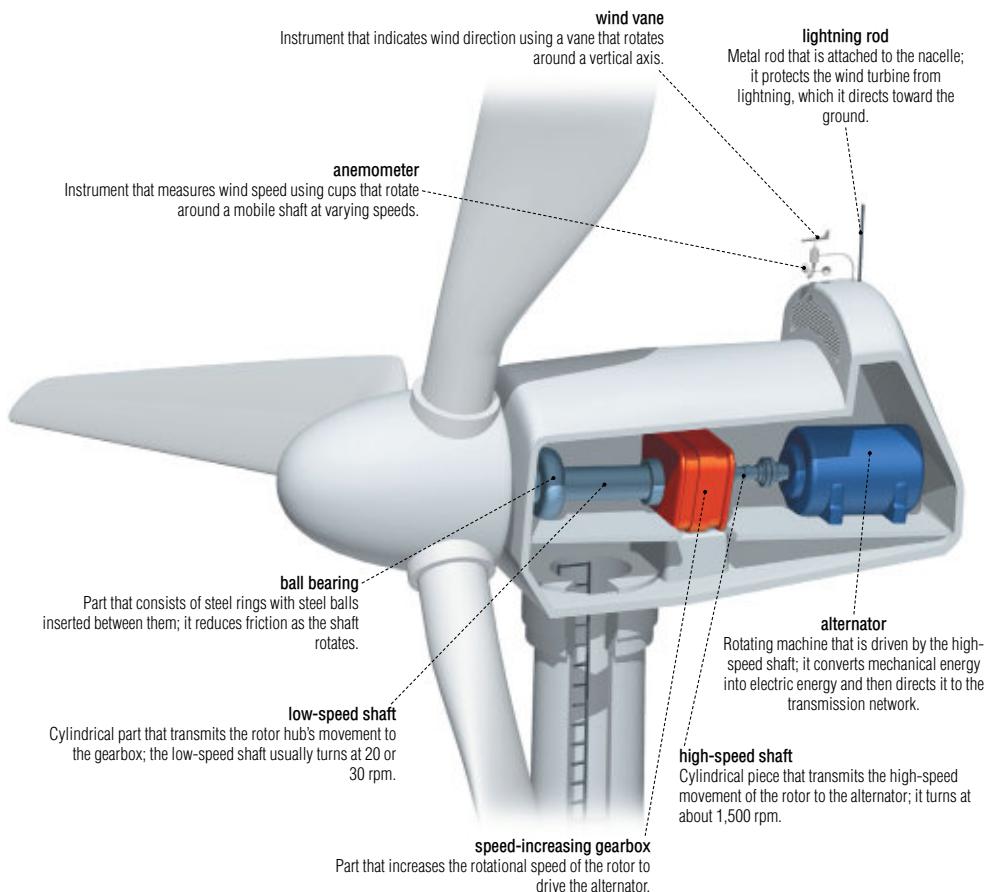


wind turbines and electricity production

horizontal-axis wind turbine

The most common type of wind turbine; its axis positions itself in the direction of the wind.





wind turbines and electricity production

production of electricity from wind energy

Wind farms contain a group of wind turbines, which are driven by the wind; they produce electricity and carry it along the transmission and distribution networks to which they are connected.

horizontal-axis wind turbine

The most common type of wind turbine whose axis is parallel to the direction of the wind.

energy integration to the transmission network

The electricity produced is integrated into the network.

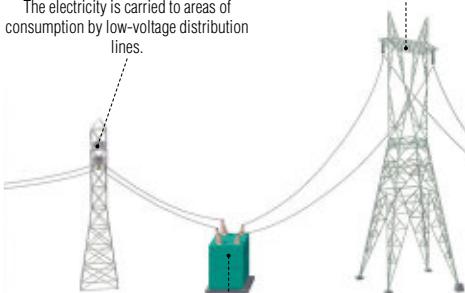


high-tension electricity transmission

Using high-voltage lines to transmit electricity over long distances reduces the strength of the current and, as a result, energy losses.

transmission to consumers

The electricity is carried to areas of consumption by low-voltage distribution lines.



voltage decrease

Before integrating the electricity into the home network, the voltage is progressively decreased to 240 V.

second voltage increase

first voltage increase

Increase in voltage: transformers carry high-voltage electricity produced by the alternator to reduce loss during transport.

ENGLISH INDEX

1/10 second hand 55

A

aboveground pipeline 106

absorbing plate 153

absorbing surface 162

accept machine 138

access gallery 115

access shaft 132

actinides 20

actinium 20

acute angle 75, 76

additive color synthesis 37

adjustment for horizontal-circle image 68

adjustment for vertical-circle image 68

administrative building 130

aerodynamic brake 165

afterbay 113, 115, 122, 123

air gap 162

air hose 90

air leg 90

alcohol bulb 52

alcohol column 52

alidade 68

alidade level 69

alkali metals 14

alkaline earth metals 14

alkaline manganese-zinc cell 31

alternator 29, 167

altitude clamp 50, 51

altitude fine adjustment 50, 51

aluminum 15

americium 20

amorphous solid 9

amount of substance, measurement 71

ampere 70

amplitude 35

analog watch 57

analytical balance 63

anchor wires 100

anemometer 167

angles, examples 76

angles, measure 68

anode 30, 31

anticline 99

antimony 15

antireflection coating 152

anvil 67

aperture changer 46

aperture diaphragm 46

arc 77

arch dam 122

arch dam, cross section 122

argon 16

arm 43, 45

armature 28

armature core 29

armature winding 29

arsenic 15

asphalt 103

astatine 16

astigmatism 39

atom 8

atomic number 12

atoms 8

attraction 34

automatic tank gauge 108

azimuth clamp 50, 51

azimuth fine adjustment 50, 51

B

ball bearing 167

bank 131

bank of heliostats 156, 158

bar 24

barium 14

base 22, 43, 44, 60, 61, 165

base plate 69

basin 131

basin side 133

bathroom scale 64

battery 27, 155

beaker 21

beam 60

beam balance 60

beam diameter reduction 46

bearing pad 140

bequerel 71

belt loader 84, 89

bench 86

bench height 87

berkelium 20

berm 121

beryllium 14

bevel gear 25

bevel square 69

biconcave lens 40

biconvex lens 40

bimetallic helix 54

bimetallic thermometer 54

binocular microscope 44

biology 74

birth 74

bismuth 15

bit 90, 98

black 37

blade 118, 165, 166

blasting charge 97

blood factor negative 74

blood factor positive 74

blower 145

blue 37

body 42, 58

body tube 45

bohrium 19

boiler 158

boiling-water reactor 150

boron 15

bottle 21

bottom cap 30, 31

bottom deck 111

bottom ring 117

bottom road 95

brace 127

branch 27

breather valve 108

bridge 42

bromine 16

brush 28

brushes 29

bucket 118

bucket ring 118

bucket wheel excavator 89

buffer tank 106

bulb 27

bulb unit 132

bulldozer 89

bulwark 108

bundle 128

bunker oil 103

bushbar 114

bushing 113, 115, 126

buttress 119

butress dam 119

butress dam, cross section 119

c

cadmium 19

calandria 143

calcium 14

californium 20

candela 71

canned failed fuel 139

cantilever 122

cap 164

capacitor 32

capillary tube 53

carbon 16

carbon dioxide gas coolant

144

carbon dioxide reactor 144

carbon rod 30

carbon-zinc cell 30

case 54, 55, 65

casing first string 102

cathode 30, 31

cells 27

Celsius 71

Celsius scale 52

Celsius temperature, measurement 71

center 77

center wheel 56, 59

central column 165

central focusing wheel 42

central pumping station 106

ceramic capacitor 32

cerium 17

cesium 14

chain 58

chain reaction 10

chemical bond 8

chemical elements 12

chemistry 8

chemistry symbols 24

chlorine 16

Christmas tree 101, 102, 106

chromium 18

chuck 91

chute 94

circle, parts 77

circuit breaker 114

circulating pump 160, 161

circumference 77

clamp/holder 22

clamping block 66

clamping screws 66

claw-pole rotor 29

clay core 120

click 56, 59

clinical thermometer 53

clip 65

clock mechanism 59

coal mine 86

coal storage yard 84

coal-fired thermal power plant 85

coarse adjustment knob 45

cobalt 18

coil 28

cold air 162

cold coolant 135, 159

cold heavy water 147

collector rings 29

color synthesis 37

column of mercury 53

commutator 28

concave lens 39, 40

concrete drain 109

concrete shielding 144, 146,

149, 151

concrete wall 162

condensation 9

condensation of steam into

water 136

condensation pool 150

condenser 43, 44, 46, 82, 85, 142, 159

condenser adjustment knob 44

condenser height adjustment 45

condenser water outlet 142

conduction 11

cone 81

confined aquifer 82

connection pin 33

connection point 129

constriction 53

containment building 134, 141

control lever 91

control panel 47

control rod 145, 146, 148, 150

control room 113, 143

control visual display 47

conviction 11

convection current 11

converging lenses 40

convex lens 39, 40, 48

conveyor 84, 88, 93

coolant 134, 158

coolant inlet 153

coolant outlet 153

coolant: boiling water 151

coolant: carbon dioxide 145

coolant: pressurized heavy water 146

coolant: pressurized water 149

cooling cylinder 41

cooling tower 83, 85

copper 18

cornea 38

coulomb 70

coupling bolt 118

crane 101

crater 86

crest 35

crest of spillway 112

cross cut 94

cross section of a buttress dam 119

cross section of a gravity dam 123

cross section of a buttress

dam 119

cross section of a gravity dam 123

cross section of a buttress

dam 119

cross section of an electron microscope 46 cross section of an embankment dam 120 crossarm 127, 128 crown 57 crude-oil pipeline 106 crusher 84 crystallization 9 cube 80 curium 20 cut-off trench 120, 123 cyan 37 cylinder 81

D

d quark 8 dam 113 darmstadtium 19 data record system 47 death 74 decagon 79 deck 95 declination setting scale 50, 51 degree 75 degree Celsius 71 depolarizing mix 30 derrick 98, 100, 106 dial 54, 57, 58, 60 diameter 77 diesel oil 103 digital display 64 digital thermometer 54 digital watch 57 diode 155 direction of electron flow 27, 31 discharge bay 138 displacement 35 display 62 distance traveled 65 distance, measure 65 distribution panel 129 diverging lenses 40 diversion tunnel 112 divided by 72 dodecagon 79 door access 63 double glazing 162 double pulley system 26 dosing water tank 134, 143 dovetail 49 downstream face 123 downstream shoulder 121 downstream toe 121 draft tube 115, 116 draft tube liner 117 drain valve 109, 110 drainage blanket 121 drainage layer 121 draw tube 44 drift 94 drill collar 98 drill pipe 98

drill rod 90 drill ship 104 drilling drawworks 98 drilling rig 98 drive pulley 29 drum 59 dry cells 30 dry well 150 dual-in-line package 33 dubnium 19 dump 88, 92 dynamo 28 dysprosium 17

E

effort 24, 26 einsteinium 20 electric charge, measurement 70 electric current, measurement 70 electric potential difference, measurement 70 electric resistance, measurement 70 electrical circuit, parallel 27 electricity 27 electricity meter 129 electricity transmission 126, 137 electricity transmission network 159 electrolytic capacitors 32 electrolytic separator 30 electromagnetic spectrum 36 electron 8 electron beam 46 electron beam positioning 46 electron collector 31 electron flow, direction 27, 31 electron gun 46, 47 electron microscope elements 47 electron microscope, cross section 46 electronic scale 62 electronics 32 elements, table 12 elevation adjustment 49 elevator 94, 138 embankment dam 120 embankment dam, cross section 120 emergency support vessel 104 empty set 72 end cap 140 end plate 140 energy 8 energy integration to the transmission network 125, 168 energy release 10 energy transmission at the generator voltage 124 energy, measurement 70

engine 99 equals 72 equipment lock 139 erbium 17 erecting lenses 49 Erlenmeyer flask 21 escape wheel 56, 59 europeum 17 evaporation 9 examples of angles 76 examples of dams 119 exciter 117 exhaust port 91 expansion chamber 53 expansion tank 160 eyepiece 42, 43, 44, 49, 50, 51

F

face 86, 88, 94 factorial 73 Fahrenheit scale 52 measurement 70 failed fuel bay 139 failed fuel canning 138 fan wheel 28, 29 faintail 164 feeder header 143 feedwater 144, 147, 149, 151 female 74 fermium 20 field electromagnet 28 field lens 49 field lens adjustment 44 field line 34 field winding 29 fifty 74 filling inlet 111 filter 161 finderscope 50, 51 fine adjustment knob 45 fine adjustment wheel 66 finely threaded screw 67 first voltage increase 168 fission of uranium fuel 135 fission products 10 fissionable nucleus 10 five 74 five hundred 74 fixed jaw 66 fixed platform 105 fixed-roof tank 108 flame 11 flare 101 flash tube 41 flat-plate solar collector 153 flexible hose 91 flexible hose connection 91 floating roof 110 floating-roof tank 110 floor 164 flow bean 102 flow tube 153 fluorine 16 focus 38, 39 focusing knob 50, 51

focusing lenses 46 focusing ring 42 force, measurement 70 fork 59 foundation 119 foundation blockage 119 foundation of dam 121 fourth wheel 56 fraction 73 frame 28, 29, 67, 153, 154, 164 Francis runner 118 francium 14 freezing 9 frequency, measurement 70 front beam 61 fuel 134, 146 fuel bundle 140, 141 fuel handling sequence 138 fuel pellet 140, 141 fuel: enriched uranium 149, 151 fuel: natural uranium 145, 146 fueling machine 138, 143, 144, 147 fulcrum 24 fully reflecting mirror 41 function keys 62 furnace 157 fuse 127, 155 fuse cutout 127 fuse holder 127

G

gadolinium 17 gallery 164 gallium 15 gamma rays 36 gantry crane 113, 114, 115 gas 9, 98 gas burner 22 gas lift module 101 gases, noble 16 gasoline 103 gate 114, 115, 131 gate operating ring 117 gear tooth 25 gearing systems 25 generator 82, 117, 142 generator unit 115, 116 generators 28 geometrical shapes 76 geometry 75 geothermal and fossil energy 82 geothermal energy 82 geothermal field 82 germanium 15 glass 153, 154 glass case 63 glass slide 43, 44 gnomon 57 gold 19 graduated cylinder 23

graduated scale 61, 64 grandfather clock 58 gravity dam 123 gravity dam, cross section 123 greases 103 green 37 ground 111 ground surface 87 ground wire 129 guy wire 165

H

hafnium 19 hairspring 56 hammer drill 90 handle 48, 91 hassium 19 haulage road 86 head of water 124 headbay 113 headframe 93, 94 heat exchanger 144, 160, 161 heat production 135 heat transfer 11 heat transport pump 143 heating oil 103 heavy-water reactor 146 heliostats 158 helipad 100 helium 16 helix 80 hemisphere 80 hemlath 164 hendecagon 79 heptagon 79 hertz 70 hexagon 79 high-pressure steam inlet 142 high-speed shaft 167 high-tension electricity transmission 83, 85, 125, 168 hill 156 hinge 42 hoist room 93 holder 22 holmium 17 hook 64 horizontal clamp 68 horizontal member 128 horizontal-axis wind turbine 166, 168 hot coolant 135, 159 hot line connector 127 hour hand 58, 59 hub 118, 166 hub cover 118 hull column 101 hydraulic turbine 116 hydroelectric complex 112 hydroelectric power plant, cross section 114 hydroelectricity 112 hydrogen 13 hyperopia 39

- I**
- illumination mirror 69
 - impervious rock 99
 - inactive dike 131
 - incandescent lamp 155
 - incident neutron 10
 - indium 15
 - infinity 73
 - infrared radiation 36
 - injection well 83
 - insulation 153
 - insulator 126, 127
 - integral 73
 - integrated circuit 33
 - intermediate booster station 107
 - international system of units 70
 - intersection of two sets 72
 - iodine 16
 - iridium 19
 - iron 18
 - is an element of 73
 - is approximately equal to 72
 - is equivalent to 72
 - is greater than 73
 - is greater than or equal to 73
 - is identical with 72
 - is included in/is a subset of 72
 - is less than 73
 - is less than or equal to 73
 - is not an element of 73
 - is not equal to 72
 - is not identical with 72
 - is not parallel to 75
 - is parallel to 75
-
- M**
- jack-up platform 105
 - jacket 30
 - jackleg drill 90
 - jet fuel 103
 - jewel 56
 - joule 70
-
- K**
- Kaplan runner 118
 - kelly 99
 - kelvin 71
 - kerosene 103
 - kilogram 71
 - krypton 16
-
- L**
- laboratory equipment 21
 - ladder 111
 - lagging 109
 - lamp 44
 - landing 95
 - lanthanides 17
 - lanthanum 17
 - laser beam 41
 - lawrencium 20
 - lead 15
 - length, measure 65
 - length, measurement 71
-
- lens** 38
- lens system 42
 - lenses 40
 - level 95
 - leveling head 68
 - leveling head level 68
 - leveling head locking knob 68
 - leveling screw 63, 69
 - lever 24
 - lid 33
 - lifeboat 101
 - light ray 38
 - lightning arrester 115, 127
 - lightning rod 167
 - limb top 44
 - liquid 9, 11
 - liquid crystal display 57
 - liquid nitrogen tank 47
 - lithium 14
 - load 24, 26
 - loading area 138
 - loading bunker 92
 - lock 130
 - lock nut 67
 - log chute 112
 - low-pressure steam inlet 142
 - low-speed shaft 167
 - low-tension distribution line 126, 129
 - lower confining bed 83
 - lubricating oils 103
 - lubricator 91
 - luminous intensity, measurement 71
 - lutetium 17
-
- M**
- machine hall 113, 115
 - magenta 37
 - magma chamber 83
 - magnesium 14
 - magnet 34
 - magnetic damping system 61
 - magnetic field 34
 - magnetism 27, 34
 - magnifying glass 43, 48
 - main fan 92
 - main inlet 108
 - main leg 128
 - main scale 66
 - main scope tube 48
 - main steam header 142
 - main switch 129
 - main tube 50, 51
 - main wheel 59
 - maintenance shop 92
 - male 74
 - manganese 18
 - manganese mix 31
 - manhole 109, 110
 - manifold 101
 - manometer 109
 - marway 94
 - maritime transport 93
 - mass, measurement 71
 - master gate valve 102
-
- mathematics** 72
- matter 8
 - mean position 35
 - measure of angles 68
 - measure of distance 65
 - measure of length 65
 - measure of temperature 52
 - measure of thickness 66
 - measure of time 55
 - measure of weight 60
 - measurement of amount of substance 71
 - measurement of Celsius temperature 71
 - measurement of electric charge 70
 - measurement of electric current 70
 - measurement of electric potential difference 70
 - measurement of electric resistance 70
 - measurement of energy 70
 - measurement of force 70
 - measurement of frequency 70
 - measurement of length 71
 - measurement of luminous intensity 71
 - measurement of mass 71
 - measurement of power 70
 - measurement of pressure 71
 - measurement of radioactivity 71
 - measurement of thermodynamic temperature 71
 - measuring devices 52
 - mechanical shovel 89
 - mechanical stage 45
 - mechanical stage control 45
 - mechanical watch 56
 - mechanics 25
 - medium-tension distribution line 127, 129
 - meitnerium 19
 - melting 9
 - mendelevium 20
 - mercury 19
 - mercury bulb 53
 - metallic contact grid 152
 - metalloids 15
 - metals, alkali 14
 - metals, alkaline earth 14
 - metals, transition 18
 - meter 71
 - micrometer caliper 67
 - micrometer screw 68
 - microscope 43
 - microscope, binocular 44
 - microscopes 43
 - microwaves 36
 - miners' changing-room 93
 - minus/negative 72
 - minute 75
 - minute hand 55, 58, 59
 - mirror 43
-
- moderator** 134
- moderator tank 147
 - moderator: graphite 145
 - moderator: heavy water 146
 - moderator: natural water 149,
 - 151
 - mole 71
 - molecule 8
 - molybdenum 18
 - Moon dial 58
 - mud injection hose 98
 - mud pit 99
 - mud pump 99
 - multipled by 72
 - myopia 39
-
- N**
- nacelle 166, 167
 - nacelle, cross-section 167
 - negative charge 24
 - negative contact 152, 154
 - negative meniscus 40
 - negative region 152
 - negative terminal 27, 30
 - neodymium 17
 - neon 16
 - neptunium 20
 - network connection 129
 - neutral line 34
 - neutron 8
 - new fuel storage room 139
 - newton 70
 - nickel 18
 - niobium 18
 - nitrogen 16
 - nobelium 20
 - noble gases 16
 - node 27
 - non-metals 16
 - nonagon 79
 - normal vision 38
 - north pole 34
 - notch 61
 - nuclear energy 134
 - nuclear energy, production of electricity 134
 - nuclear fission 10
 - nuclear fuel handling sequence 138
 - nuclear generating station 142
 - nuclear reactor 141
 - nucleus 8
 - nucleus splitting 10
 - numeric keyboard 62
-
- O**
- object 38
 - objective 43, 44
 - objective lens 42, 48
 - obtuse angle 75, 76
 - offshore drilling 104
 - offshore prospecting 97
 - offshore well 106
 - ohm 70
 - oil 96, 98
 - oil processing area 100
 - oil/gas separator 101
-
- P**
- packaged integrated circuit 32, 33
 - palladium 18
 - pallet 59
 - pan 60, 61, 63
 - pan hook 61
 - panel 95, 128
 - parabolic mirror 157
 - paraffins 103
 - parallel electrical circuit 27
 - parallelepiped 81
 - parallelogram 78
 - partially reflecting mirror 41
 - parts of a circle 77
 - pascal 71
 - pediment 58
 - pedometer 65
 - Pelton runner 118
 - pencil 140
 - pendulum 58
 - pendulum bob 59
 - pendulum rod 59
 - penstock 113, 114, 133
 - pentagon 79
 - percent 73
 - peripheral joint 122
 - perpendicular 75
 - Petri dish 23
 - petrochemicals 103
 - petroleum trap 96, 97
 - phosphorus 16
 - photographic chamber 47
 - photon 41
 - physics 35
 - pi 75
 - pier 104
 - pillar 94
 - pinion 59
 - pipeline 102, 107
 - pitching 120
 - pithead 92
 - pivot 24
 - plane surfaces 77
 - plane-concave lens 40
 - plane-convex lens 40

- plastic film capacitor 32
 platform 62
 platinum 19
 plinth 58
 plus or minus 73
 plus/positive 72
 plutonium 20
 pneumatic hammer 91
 pointer 54, 60, 64
 polonium 15
 polygons 78
 pontoon 100
 pool 161
 Porro prism 42
 port 138
 positive charge 24
 positive contact 152, 154
 positive meniscus 40
 positive region 152
 positive terminal 27, 30
 positive/negative junction 152
 post 163
 post mill 163
 potassium 14
 power plant 113, 130
 power plant, cross section 132
 power source 27
 power, measurement 70
 praseodymium 17
 pressure gauge 102
 pressure tube 140, 141
 pressure, measurement 71
 pressurized heavy water 147
 pressurized-water reactor 148
 pressurizer 147, 148
 printed circuit 32
 printed circuit board 32
 printout 62
 prism binoculars 42
 product code 62
 production of electricity by the generator 124, 137
 production of electricity from geothermal energy 82
 production of electricity from nuclear energy 134
 production of electricity from solar energy 158
 production of electricity from thermal energy 84
 production of electricity from wind energy 168
 production of electricity, steps 124
 production platform 100, 107
 production well 82
 production/export riser system 100
 promethium 17
 propagation 35
 protactinium 20
 proton 8
 protractor 69
 pulley 26
 pulsed ruby laser 41
 pulverizer 85
 pulvino 122
 pump 147, 148, 150, 159
 pumping station 107
 pylon 128
 pylon body 128
 pylon foot 128
 pylon top 128
 pyramid 81
-
- Q**
- quadrant 77
 quadrilateral 78
 quark 8
-
- R**
- rack and pinion gear 25
 radiation 11
 radio mast 101
 radio waves 36
 radioactive nuclei 10
 radioactivity, measurement 71
 radius 14
 radius 77
 radon 16
 rail track 93
 ramp 87
 rare earth 17
 ratchet knob 67
 ratchet wheel 56, 59
 reaction direction 24
 reactor 135, 138, 143
 reactor building 141, 143
 reactor building airlock 143
 reactor core 145, 148, 150
 reactor tank 150
 reactor vessel 141
 reactor, carbon dioxide 144
 reactor, heavy-water 146
 rear beam 61
 reception bay 139
 rectangle 78
 red 37
 reentrant angle 76
 refinery 106
 refinery products 103
 reflecting cylinder 41
 reflecting surface 156
 reflecting telescope 50
 refracting telescope 51
 regular decagon 79
 regular dodecagon 79
 regular hendecagon 79
 regular heptagon 79
 regular hexagon 79
 regular nonagon 79
 regular octagon 79
 regular octahedron 81
- regular pentagon 79
 reheater 142
 repulsion 34
 reservoir 113, 114, 119, 120, 122, 123
 reset button 55, 65
 resistors 32
 retainer 91
 reticle 49
 retina 38
 reversible reaction 24
 revolving nosepiece 43, 44
 rhodium 19
 rhodium 18
 rhombus 78
 right angle 75, 76
 right ascension setting scale 50, 51
 ring 55, 64, 118
 Roberval's balance 60
 rod 22
 roentgenium 19
 Roman numerals 74
 roof 88
 room 94
 rope 26
 rotary system 99
 rotary table 99
 rotation of the turbine 124
 rotor 116, 163, 165
 round-bottom flask 21
 rubidium 14
 ruby cylinder 41
 ruler 65, 66
 runner 116
 runner blade 117, 118, 132
 runners 118
 ruthenium 18
 rutherfordium 19
-
- S**
- safety tank 147
 safety valve 135
 sail 164
 sail cloth 164
 sailbar 164
 samarium 17
 sand 120
 scale 53, 65
 scandium 18
 science 8
 scientific symbols 70
 screen 114
 scroll case 115
 sea 130
 sea side 132
 seaborgium 19
 sealing material 31
 sealing plug 30, 31
 sealing ring 111
 second 75
 second hand 55
- second voltage increase 168
 secondary inlet 108
 sector 77
 seismographic recording 96, 97
 selenium 15
 semi-metals 15
 semicircle 77
 semisubmersible platform 105
 separator 31, 82, 142
 serological pipette 23
 service building 139
 service pipe 160
 shadow 57
 shaft 25, 28, 29, 54, 117
 shell 110
 shock wave 96, 97
 shunt 27
 shutter 162
 silencer 91
 silicon 15
 silver 19
 skip 95
 sliding jaw 66
 sliding weight 61
 sodium 14
 soil 122
 solar cell 152, 154
 solar collector 153, 161
 solar energy 152
 solar energy, production of 158
 solar furnace 156
 solar house 160
 solar radiation 152, 153, 154, 157, 158, 162
 solar ray reflected 157, 158
 solar-cell panel 154
 solar-cell system 154
 solid 9, 11
 solids 80
 south pole 34
 spacer 140
 specimen chamber 47
 specimen positioning control 47
 spectrometer 47
 speed-increasing gearbox 167
 spent fuel discharge bay 142
 spent fuel port 138
 spent fuel storage bay 139, 141, 143
 sphere 80
 spillway 112
 spillway chute 112
 spillway gate 112
 spindle 59, 67
 spiral case 116
 spiral staircase 109
 splash plate 109
 spray nozzle 108
- spring balance 64
 sprinklers 134
 spur gear 25
 square 78
 square root of 73
 stack 85
 stage 43, 45, 46
 stage clip 43, 44
 stairs 111
 stand 22
 start button 55
 states of matter 9
 stator 117
 stay ring 117
 stay vane blade 117
 steam 82
 steam generator 85, 143, 147, 149, 159
 steam outlet 144, 147, 149, 151
 steam pressure drives turbine 136
 steel casing 31
 steelard 61
 stem 53
 step setting 65
 steps 163
 steps in production of electricity 124
 stock 164
 stop button 55
 stopwatch 55
 storage tank 160
 storage tray 139
 stove oil 103
 straight stopcock burette 23
 strap 57
 strip mine 88
 strontium 14
 strut 165
 sublimation 9
 submarine pipeline 107
 substation 130
 substructure 98
 subtractive color synthesis 37
 sulfur 16
 sum 73
 sump 95
 sundial 57
 superconducting 9
 supply of water 124
 supply point 126
 surface prospecting 96
 suspension insulator string 128
 suspension spring 59
 switch 27
 swivel 98
 symbol 12
 system of units 70

T

table of elements 12
tail pole 163
tailrace 115
tank farm 107
tank gauge float 108
tanks 108
tantalum 19
target area 157
technetium 18
telescope 69
telescopic sight 48
tellurium 15
temperature, measure 52
ten 74
terbium 17
terminal 106, 126
terminal box 154
test tube 23
thallium 15
theodolite 68
thermal energy 84
thermodynamic temperature, measurement 71
thermometer 52, 111
thickness, measure 66
thimble 67
third wheel 56, 59
thorium 20
throttle valve 91
thrust bearing 116
thulium 17
tidal power plant 130
tin 15

titanium 18
tool 91
toothed wheel 25
top cap 30
top deck 111
top of dam 112, 120, 123, 133
top road 95
toric lens 39
torus 80
total 62
tower 157, 158, 164, 166
tower mill 164
transfer canal 138
transfer of heat to water 135
transformation of mechanical work into electricity 124
transformer 114, 126, 142, 159
transition metals 18
transmission to consumers 85, 125, 168
trapezoid 78
traveling crane 115
treatment plant 93
trench 88
triangle 78
tripod 51
Trombe wall 161, 162
trough 35
tubing 102
tubing head 102
tubing valve 102
tubular member 100

tungsten 19
turbine 82, 142
turbine headcover 116
turbine runner 132
turbine shaft turns generator 136
turbo-alternator 159
turbo-alternator unit 85
turret cap 49

U

u quark 8
ultraviolet radiation 36
underground mine 94
union of two sets 72
unit price 62
ununbium 19
upper confining bed 82
upstream blanket 120
upstream face 123
upstream shoulder 120
upstream toe 120
uranium 20

V

vacuum chamber 46
vacuum manifold 46
vacuum system console 47
vanadium 18
vapor 11
ventilation 160
vernier 61, 66
vernier caliper 66
vernier scale 66
vertical shaft 94

vertical-axis wind turbine 165
vibrating mudscreen 98
visible light 36
vision 38
vision defects 39
visual transmission 46
volt 70
voltage decrease 85, 125, 168
voltage increase 83, 85, 124, 137

W

warm air 162
wash bottle 21
washer 30
water 83
water cools the used steam 136
water hose 90
water intake 114
water is pumped back into the steam generator 136
water separator 90
water turns into steam 135
water under pressure 124
water-heater tank 160
water-steam mix 82
watt 70
wave 35
wave wall 120
wavelength 35
weighing platform 64
weight 58, 59, 60, 62
weight, measure 60

weight-driven clock
mechanism 59
wet well 151
white 37
wicket gate 116
wind energy 163
wind turbine, horizontal-axis 168
wind turbines 165
wind vane 167
winder 56

winding adjustment 49
winding mechanism 59
winding shaft 95
winding tower 93, 95
windmill 163
windshaft 164
winze 94
wire 33
worm gear 25

X

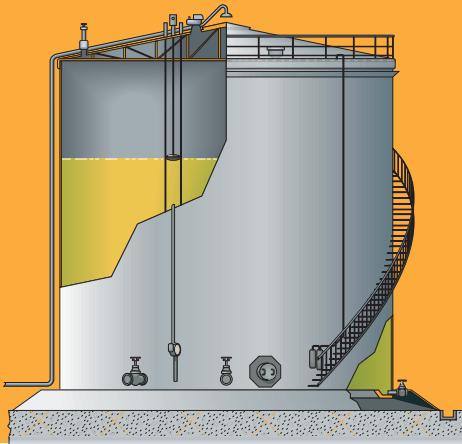
X-rays 36
xenon 16

Y

yellow 37
ytterbium 17
yttrium 18

Z

zinc 18
zinc can 30
zinc-electrolyte mix 31
zirconium 18



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